

Communications

One-Pot Synthesis of (*E*)-2-Arylideneindan-1-ones from the Baylis-Hillman Adducts via the Successive Inter- and Intramolecular Friedel-Crafts Reactions

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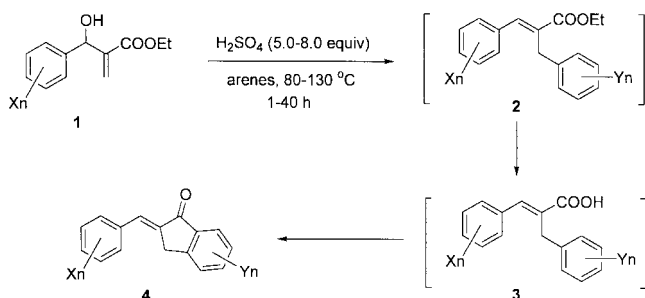
The Baylis-Hillman reaction is well known as a coupling reaction of aldehydes and activated alkenes catalyzed by tertiary amines or tertiary phosphines.¹ The reaction with ethyl acrylate serves α -methylene- β -hydroxy esters, which have been transformed to various useful compounds.²

Friedel-Crafts reaction of the Baylis-Hillman adducts is well documented in the literature.³ Recently, we described on the Friedel-Crafts reaction of the Baylis-Hillman adducts of *N*-tosylimines with arenes under the catalytic influence of sulfuric acid providing a facile stereoselective synthesis of ethyl 3-aryl-2-benzylprop-2-enoates.^{3a} We felt that the synthesis of 2-arylideneindan-1-one derivatives would be possible from ethyl 3-aryl-2-benzylprop-2-enoates (**2** in Scheme 1) via the intramolecular Friedel-Crafts acylation by using excess amounts of sulfuric acid. 2-Arylideneindan-1-one derivatives are useful for the synthesis of various biologically important compounds.⁴ These compounds can be synthesized by a Claisen-Schmidt condensation between appropriate aryl aldehydes and 1-indanones.^{5,6} However, synthesis of 1-indanone derivatives must be preceded in

order to synthesize 2-arylideneindan-1-ones.

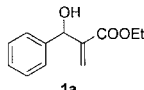
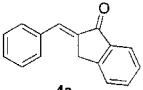
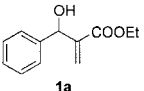
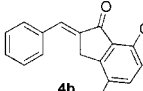
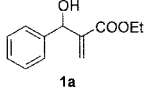
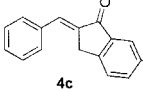
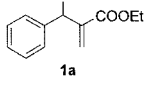
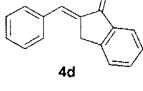
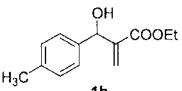
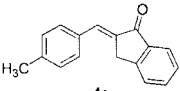
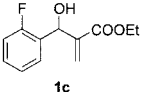
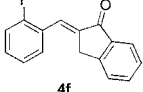
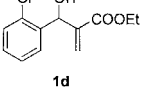
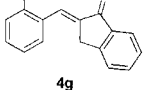
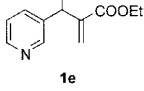
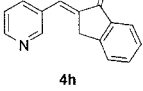
Thus, we examined the reaction of the Baylis-Hillman adduct and benzene in the presence of sulfuric acid. A mixture of **1a** (412 mg, 2 mmol) and H₂SO₄ (980 mg, 10 mmol, 5 equiv) in dry benzene (5 mL) was stirred during reflux for 3 h. After the usual work-up and column chromatography (hexane/ether, 9 : 1), we could obtain the desired (*E*)-2-benzylideneindan-1-one (**4a**) in 65% yield as a pale yellow solid.⁷

The reaction mechanism is depicted in Scheme 1. The reaction of the Baylis-Hillman adduct **1** and arene gave ethyl (2*E*)-3-aryl-2-arylmethylprop-2-enoate **2** regioselectively via the intermolecular Friedel-Crafts reaction as reported.³ Successive intramolecular Friedel-Crafts acylation of **2** gave the desired product (*E*)-2-arylideneindan-1-ones **4**. To a certain extent, hydrolysis of **2** to the corresponding carboxylic acid **3** would occur in the reaction conditions. As shown in Table 1, various Baylis-Hillman adducts **1a-e** in the reaction with some arenes (benzene, toluene, xylene, chlorobenzene) gave similar results. Unexpectedly, we obtained one regioisomer in the cases of mono-substituted arenes such as toluene (**4c** in entry 3) and chlorobenzene (**4d** in entry 4). The other possible regioisomer can not be isolated.⁸ The structures of **4c** and **4d** were identified by NOE experiment (Figure 1). Irradiation of the methylene protons ($\delta = 4.02$ ppm) at the 3-position of **4c** showed an increase of aromatic protons at the benzylidene portion ($\delta = 7.66-7.70$ ppm, 3%) and at the indanone ring (H₄, $\delta = 7.40-7.50$ ppm, 0.5%). Irradiation of the methyl protons ($\delta = 2.44$ ppm) at the 6-position of **4c** showed an increase of H₇ ($\delta = 7.72$ ppm, 0.6%) and H₅ ($\delta = 7.40-7.50$ ppm, 0.5%) at the indanone ring. Similar NOE results were observed in the case of **4d** as shown in Figure 1.



Scheme 1

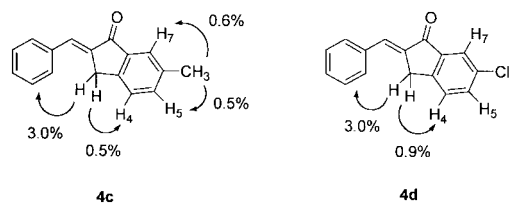
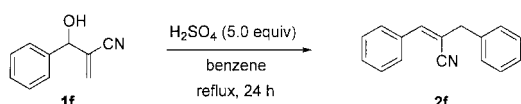
Table 1. Synthesis of (*E*)-2-arylideneindan-1-ones **4**

Entry	B-H adduct (1)	Conditions	Product (% yield)
1		benzene H ₂ SO ₄ (5 equiv) reflux 3 h	 (65)
2		<i>p</i> -xylene H ₂ SO ₄ (5 equiv) 110-120 °C 1 h	 (79)
3		toluene H ₂ SO ₄ (8 equiv) reflux 36 h	 (63)
4		chlorobenzene H ₂ SO ₄ (6 equiv) 120-130 °C 5 h	 (37)
5		benzene H ₂ SO ₄ (6 equiv) reflux 5 h	 (81)
6		benzene H ₂ SO ₄ (8 equiv) reflux 12 h	 (77)
7		benzene H ₂ SO ₄ (6 equiv) reflux 40 h	 (60)
8		benzene H ₂ SO ₄ (6 equiv) reflux 20 h	 (42)

From the NOE experiment, we could confirm the structure as well as the stereochemistry (*E*) of the double bond.

As expected from the previous reports,³ for nitrile substituted Baylis-Hillman adduct such as **1f**, the first intermolecular Friedel-Crafts reaction with benzene gave the corresponding *Z*-form olefin **2f**,^{3a} which did not undergo the next intramolecular Friedel-Crafts acylation in the reaction conditions.

The use of trifluoromethanesulfonic acid (5 equiv) instead of sulfuric acid in the reaction of **1a** and benzene (reflux, 2 h) gave increased yield of **4a** (87%). However, when we use trifluoromethanesulfonic acid or sulfuric acid lesser than 5.0 equivalents low yields of **4** were obtained. Instead, **2** and **3**

**Figure 1****Scheme 2**

were isolated as the major products.

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References

- (a) Drewes, S. E.; Roos, G. H. P. *Tetrahedron* **1988**, *44*, 4653. (b) Basavaiah, D.; Rao, P. D.; Hyma, R. S. *Tetrahedron* **1996**, *52*, 8001. (c) Ciganek, E. In *Organic Reactions*; Paquette, L. A., Ed.; Wiley: New York, 1997; Vol. 51, pp 201-350.
- (a) Kim, J. N.; Lee, H. J.; Lee, K. Y.; Kim, H. S. *Tetrahedron Lett.* **2001**, *42*, 3737. (b) Kim, J. N.; Im, Y. J.; Gong, J. H.; Lee, K. Y. *Tetrahedron Lett.* **2001**, *42*, 4195. (c) Kim, J. N.; Lee, K. Y.; Kim, H. S.; Kim, T. Y. *Org. Lett.* **2000**, *2*, 343. (d) Kim, H. S.; Kim, T. Y.; Lee, K. Y.; Chung, Y. M.; Lee, H. J.; Kim, J. N. *Tetrahedron Lett.* **2000**, *41*, 2613. (e) Lee, H. J.; Chung, Y. M.; Lee, K. Y.; Kim, J. N. *Bull. Korean Chem. Soc.* **2000**, *21*, 843. (f) Lee, H. J.; Kim, H. S.; Kim, J. N. *Tetrahedron Lett.* **1999**, *40*, 4363. (g) Basavaiah, D.; Kumaragurubaran, N.; Sharada, D. S. *Tetrahedron Lett.* **2001**, *42*, 85. (h) Yang, K.-S.; Chen, K. *Org. Lett.* **2000**, *2*, 729. (i) Chamakh, A.; Amri, H. *Tetrahedron Lett.* **1998**, *39*, 375. (j) Basavaiah, D.; Kumaragurubaran, N. *Tetrahedron Lett.* **2001**, *42*, 477.
- (a) Lee, H. J.; Seong, M. R.; Kim, J. N. *Tetrahedron Lett.* **1998**, *39*, 6223. (b) Basavaiah, D.; Pandiaraju, S.; Padmaja, K. *Synlett* **1996**, 393. (c) Basavaiah, D.; Krishnamacharyulu, M.; Hyma, R. S.; Pandiaraju, S. *Tetrahedron Lett.* **1997**, *38*, 2141. (d) Saib, D.; Foucaud, A. *J. Chem. Res. (S)* **1987**, 372. (e) Basavaiah, D.; Bakthadoss, M.; Reddy, G. J. *Synthesis* **2001**, 919.
- (a) Levai, A.; Patonay, T. *J. Heterocycl. Chem.* **1999**, *36*, 747. (b) Dimmock, J. R.; Kandepu, N. M.; Nazarali, A. J.; Kowalchuk, T. P.; Motaganahalli, N.; Quail, J. W.; Myktytiuk, P. A.; Audette, G. F.; Prasad, L.; Perjesi, P.; Allen, T. M.; Santos, C. L.; Szydowski, J.; De Clercq, E.; Balzarini, J. *J. Med. Chem.* **1999**, *42*, 1358. (c) El-Rayyes, N.; Al-Qatami, S.; Edun, M. *J. Chem. Eng. Data* **1987**, *32*, 481. (d) Turk, C. F. *Ger. Offen.* 2,520,171, 1975; *Chem. Abstr.* **1976**, *84*, 74261s. (e) Coombs, R. V.; Houlihan, W. J. *US* 3,843,665, 1974; *Chem. Abstr.* **1975**, *82*, 57687b.
- (a) Berthelette, C.; McCooye, C.; Leblanc, Y.; Trimble, L. A. *J. Org. Chem.* **1997**, *62*, 4339. (b) Hassner, A.; Cromwell, N. H. *J. Am. Chem. Soc.* **1958**, *80*, 893. (c) Leblanc, Y.; Dufresne, C.; Dhawan, R.; Ollerenshaw, J.; Littke, A.; Trimble, L. A.; Tsou, N. N. *Can. J. Chem.* **2000**, *78*, 784. (d) Houlihan, W. J.; Shapiro, M. J.; Chin, J. A. *J. Org. Chem.* **1997**, *62*, 1529.
- Recently, Basavaiah and Reddy have reported the synthesis of 2-arylideneindan-1-ones from the Baylis-Hillman adducts obtained from *tert*-butyl acrylate. Basavaiah, D.; Reddy, R. M. *Tetrahedron Lett.* **2001**, *42*, 3025.
- The structure of **4** was confirmed by comparison with the reported data (mp or ¹H NMR spectrum).⁴⁻⁶ As an example, spectral data of **4a** is as follows: mp 109-110 °C (lit,⁶ 109-110 °C); IR (KBr) 1693, 1624 cm⁻¹; ¹H NMR (CDCl₃) δ 4.06 (s, 2H), 7.37-7.70 (m, 9H), 7.92 (d, *J* = 7.8 Hz, 1H); ¹³C NMR (CDCl₃) δ 32.43, 124.44, 126.14, 127.66, 128.90, 129.61, 130.68, 133.89, 134.57, 134.76, 135.44, 138.05, 149.62, 194.26.
- It is interesting to note that the possible regioisomers, 4-methyl-2-benzylideneindan-1-one or 4-chloro-2-benzylideneindan-1-one, was not isolated nor detected on tlc.