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Stereoselective Synthesis of (–)-Centrolobine[†]

Eun Lee,* Hak Joong Kim, and Won Suk Jang

School of Chemistry and Molecular Engineering, Seoul National University, Seoul 151-747, Korea Received March 22, 2004

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(–)-Centrolobine (1) was isolated from the heartwood of *Centrolobium robustum* and from the stem of *Brosimum potabile* in the amazon forest.^{1,2} It features a *cis*-2,6-disubstituted tetrahydropyran core structure. It is now well established that *cis*-2,5-disubstituted tetrahydropyrans are obtained stereo-selectively *via* radical cyclization of β -alkoxyacrylates,³ and we intended to use radical cyclization reaction of β -alkoxyvinyl ketones in the synthesis of this compound.

In the retrosynthetic analysis, radical cyclization reaction of the β -alkoxyvinyl ketone **B** is expected to give the tetrahydropyran **A**, which may be transformed into 1 *via* carbonyl reduction. The intermediate **B** may be prepared *via* combination of the secondary alcohol **C** and the arylpropynone **D** (Scheme 1). Reaction of the acid chloride 2 with arylcopper reagent⁴ provided the ketone **3**, which was converted into the secondary alcohol **4** *via* asymmetric reduction by (–)-*B*-chlorodiisopinocampheylborane.⁵ LAH reduction of **4** and selective TBS protection led to the intermediate **5**. The complementary fragment arylpropynone **7** was synthesized from 4-hydroxybenzaldehyde (**6**) in a 3-step sequence (Scheme 2).

The β -alkoxyvinyl ketone **8** was prepared *via* reaction of **5** and **7** in the presence of *N*-methylmorpholine,⁶ which was then converted into the bromide **9** *via* TBS deprotection and bromide substitution. In the presence of tris(trimethylsilyl)-silane and AIBN in benzene under reflux, the bromide **9** was converted into the tetrahydropyran **10** in good yield. The final conversion from **10** to **1** required use of sodium cyanoborohydride and boron trifluoride (Scheme 3).^{7,8}

In this synthesis, the *cis*-2,6-disubstituted tetrahydropyran core of (–)-centrolobine was successfully introduced *via* radical cyclization of a β -alkoxyvinyl ketone intermediate.



[†]Dedicated to Prof. Yong Hae Kim in commemoration of his distinguished academic career.

*Corresponding Author. e-mail: eunlee@snu.ac.kr



Scheme 2. 1) 1.2 eq. 4-MeOPhMgBr, 1.2 eq. CuBr, 2.4 eq. LiBr, THF, r.t. 30 min; 2) 1.1 eq. (-)-Ipc₂BCl, THF, -25 °C, 36 h; 3) 1.5 eq. LAH, ether, 0 °C ~ r.t. 1 h; 4) 1.05 eq. TBSCl, 2.5 eq. imidazole, DCM, 0 °C ~ r.t. 1 h; 5) 2.0 eq. NaH, 1.5 eq. TBSCl, THF, 0 °C ~ r.t. 1 h; 6) 1.5 eq. CHCMgBr, THF, 0 °C ~ r.t. 1 h; 7) 2.0 eq. BaMnO₄, DCM, r.t. 24 h.



Scheme 3. 1) 1.5 eq. 7, 0.5 eq. NMM, DCM, r.t. 24 h; 2) 2.5 eq. TBAF, THF, r.t. 1 h; 3) 1.5 eq. CBr₄, 1.2 eq. PPh₃, 3.0 eq. TEA, DCM, r.t. 1 h; 4) 1.2 eq. (Me₃Si)₃SiH, 0.2 eq. AlBN, PhH, reflux, 3 h; 5) 5.0 eq. BF₃·OEt₂, 4.0 eq. NaBH₃CN, THF, r.t. 72 h.

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