

# Facile Synthesis of 5-Arylpent-4-enoates from the Baylis-Hillman Acetates

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Recently, Basavaiah *et al.* have published some papers dealing with the Johnson-Claisen rearrangement of the Baylis-Hillman adducts.<sup>1</sup> 5-Arylpent-4-enoates or 4-cyanoalk-4-enoates can be obtained from the above reaction in moderate yields. Shen *et al.* have also reported the synthesis of the latter compounds by using the sequential Michael reaction and Horner-Wadworth-Emmons (HWE) reaction of phosphonates.<sup>2</sup>

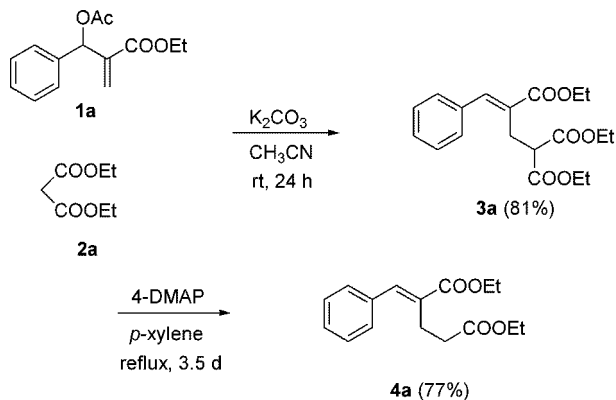
During our studies on the Baylis-Hillman chemistry<sup>3</sup> we found another efficient method for the synthesis of the abovementioned compounds. As shown in Scheme 1 the reaction of the Baylis-Hillman acetate **1a** and diethyl malonate (**2a**) in CH<sub>3</sub>CN in the presence of K<sub>2</sub>CO<sub>3</sub> gave the allylic rearrangement product **3a** in good yield. The structure of **3a** was exclusively *E*-form as in our previous papers.<sup>3</sup> Trace amount (*ca.* 5%) of the corresponding *Z*-form was observed in <sup>1</sup>H NMR spectrum. The separation of *E* and *Z*-form was difficult at this stage. Thus, we used the mixture directly in the next reaction without further purification. Following decarboxylation was conducted in *p*-xylene in the presence of 4-dimethylaminopyridine (4-DMAP).<sup>4a,4b</sup> We could isolate the desired compound **4a** in 77% yield. At this stage, pure **4a-E** could be separated easily from the minor component, **4a-Z**. In the reaction, DABCO (1,4-diazabicyclo[2.2.2]octane) and DBN (1,5-diazabicyclo[4.3.0]non-5-ene) could also be used as reported in similar systems.<sup>4c,4d</sup> However, the use of DMAP in refluxing xylene gave the best results.<sup>5</sup>

The representative results for the synthesis of the allylic rearrangement products **3a-g** are summarized in Table 1. Besides of diethyl malonate (**2a**, entries 1-3) and dimethyl

**Table 1.** Synthesis of allylic rearrangement products **3**

Entry	B-H acetate <b>1</b>	<b>2</b>	Conditions	Product	Yield (%)
1			K <sub>2</sub> CO <sub>3</sub> CH <sub>3</sub> CN rt, 24 h		81 <sup>a</sup>
2			K <sub>2</sub> CO <sub>3</sub> CH <sub>3</sub> CN rt, 24 h		87 <sup>a</sup>
3			K <sub>2</sub> CO <sub>3</sub> CH <sub>3</sub> CN rt, 20h		78 <sup>a</sup>
4			K <sub>2</sub> CO <sub>3</sub> CH <sub>3</sub> CN rt, 20 h		73 <sup>a</sup>
5			K <sub>2</sub> CO <sub>3</sub> CH <sub>3</sub> CN rt, 16h		40
6			K <sub>2</sub> CO <sub>3</sub> CH <sub>3</sub> CN rt, 32 h		55
7			K <sub>2</sub> CO <sub>3</sub> CH <sub>3</sub> CN rt, 24 h		72

<sup>a</sup>Trace amounts (*ca.* 5%) of the corresponding *Z*-isomer were observed in their <sup>1</sup>H NMR spectra.

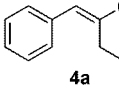
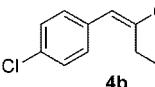
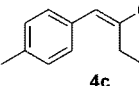
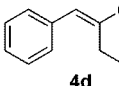
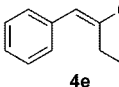
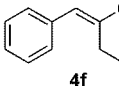
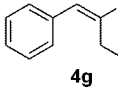


**Scheme 1**

malonate (**2b**, entry 4), some other activated methylene compounds such as ethyl cyanoacetate (**2c**, entry 5) and ethyl acetoacetate (**2d**, entry 6) gave similar results. The Baylis-Hillman acetate **1d**, derived from benzaldehyde and methyl vinyl ketone, gave **3g** similarly (entry 7). The results of selective decarboxylation of **3a-g** with 4-DMAP (5 equiv.) are summarized in Table 2.

A typical procedure for the synthesis of **3a** and **4a** is as follows: To a stirred solution of **1a** (496 mg, 2.0 mmol) and diethyl malonate (360 mg, 2.2 mmol) in CH<sub>3</sub>CN (5 mL) was added K<sub>2</sub>CO<sub>3</sub> (305 mg, 2.2 mmol) and the mixture was stirred at room temperature for 24 h. After usual workup and column chromatographic purification (hexane/ether, 8 : 1)

**Table 2.** Synthesis of 5-arylpent-4-enoate derivatives **4**

Entry	<b>3</b>	Conditions	Product <sup>a</sup>	Yield (%) <sup>b</sup>
1	<b>3a</b>	4-DMAP (5.0 equiv.) <i>p</i> -xylene reflux, 3.5 days	 <b>4a</b>	77
2	<b>3b</b>	4-DMAP (5.0 equiv.) <i>p</i> -xylene reflux, 3 days	 <b>4b</b>	67
3	<b>3c</b>	4-DMAP (5.0 equiv.) <i>p</i> -xylene reflux, 6 days	 <b>4c</b>	40
4	<b>3d</b>	4-DMAP (5.0 equiv.) <i>p</i> -xylene reflux, 2 days	 <b>4d</b>	50
5	<b>3e</b>	4-DMAP (5.0 equiv.) <i>p</i> -xylene reflux, 2 days	 <b>4e</b>	57
6	<b>3f</b>	4-DMAP (5.0 equiv.) <i>p</i> -xylene 120 °C, 4 days	 <b>4f</b>	64
7	<b>3g</b>	4-DMAP (5.0 equiv.) <i>p</i> -xylene reflux, 6 days	 <b>4g</b>	45

<sup>a</sup>Pure *E*-form. <sup>b</sup>Products were obtained as clear oil except for **4f** (mp 44–46 °C).

**3a** was obtained as a clear oil, 565 mg (81%).<sup>6</sup> A stirred solution of **3a** (348 mg, 1.0 mmol) and 4-DMAP (610 mg, 5 mmol) in dry xylene (3 mL) was heated to reflux under nitrogen atmosphere for 3.5 days. After removal of the solvent and column chromatographic purification (hexane/ether, 8 : 1) **4a** was obtained as an oil, 213 mg (77%).<sup>6</sup>

In conclusion we disclosed a facile synthetic method of synthetically useful 5-arylpent-4-enoate derivatives.<sup>1,2</sup>

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## References and Notes

- (a) Basavaiah, D.; Pandiaraju, S. *Tetrahedron Lett.* **1995**, 36, 757. (b) Basavaiah, D.; Pandiaraju, S.; Krishnamacharyulu, M. *Synlett* **1996**, 747.
- (a) Shen, Y.; Zhang, Z. *Synth. Commun.* **2000**, 30, 445. (b) Shen, Y.; Zhang, Z. *J. Chem. Res. (S)* **1999**, 556.
- (a) Kim, J. N.; Lee, H. J.; Lee, K. Y.; Gong, J. H. *Synlett* **2002**, 173. (b) Gong, J. H.; Im, Y. J.; Lee, K. Y.; Kim, J. N. *Tetrahedron Lett.* **2002**, 43, 1247. (c) Chung, Y. M.; Gong, J. H.; Kim, T. H.; Kim, J. N. *Tetrahedron Lett.* **2001**, 42, 9023. (d) Kim, J. N.; Kim, H. S.; Gong, J. H.; Chung, Y. M. *Tetrahedron Lett.* **2001**, 42, 8341. (e) Kim, J. N.; Im, Y. J.; Gong, J. H.; Lee, K. Y. *Tetrahedron Lett.* **2001**, 42, 4195. (f) Kim, J. N.; Lee, H. J.; Lee, K. Y.; Kim, H. S. *Tetrahedron Lett.* **2001**, 42, 3737. (g) Kim, H. S.; Kim, T. Y.; Lee, K. Y.; Chung, Y. M.; Lee, H. J.; Kim, J. N. *Tetrahedron Lett.* **2000**, 41, 2613. (h) Kim, J. N.; Lee, K. Y.; Kim, H. S.; Kim, T. Y. *Org. Lett.* **2000**, 2, 343. (i) Lee, H. J.; Seong, M. R.; Kim, J. N. *Tetrahedron Lett.* **1998**, 39, 6223. (j) Lee, H. J.; Kim, H. S.; Kim, J. N. *Tetrahedron Lett.* **1999**, 40, 4363. (k) Chung, Y. M.; Lee, H. J.; Hwang, S. S.; Kim, J. N. *Bull. Korean Chem. Soc.* **2001**, 22, 799. (l) Im, Y. J.; Kim, J. M.; Mun, J. H.; Kim, J. N. *Bull. Korean Chem. Soc.* **2001**, 22, 349. (m) Kim, J. N.; Lee, K. Y. *Current Organic Chemistry* **2002**, 6, 627.
- (a) Taber, D. F.; Amedio, J. C., Jr.; Gulino, F. *J. Org. Chem.* **1989**, 54, 3474. (b) Taber, D. F.; Amedio, J. C., Jr.; Patel, Y. K. *J. Org. Chem.* **1985**, 50, 3618. (c) Miles, D. H.; Huang, B.-S. *J. Org. Chem.* **1976**, 41, 208. (d) Huang, B.-S.; Parish, E. J.; Miles, D. H. *J. Org. Chem.* **1974**, 39, 2647.
- Decarbethoxylation of **3a** with DABCO (10 equiv) in refluxing xylene gave **4a** in 45% yield after 2 days. The use of DBN (5 equiv.) in similar reaction conditions gave intractable mixtures.
- Selected data for **3a** and **4a**. **3a**: oil; IR (KBr) 1746, 1733, 1707 cm<sup>-1</sup>; <sup>1</sup>H NMR (CDCl<sub>3</sub>) δ 1.15 (t, *J* = 7.2 Hz, 6H), 1.34 (t, *J* = 7.2 Hz, 3H), 3.20 (d, *J* = 7.8 Hz, 2H), 3.79 (t, *J* = 7.8 Hz, 1H), 3.98–4.15 (m, 4H), 4.27 (q, *J* = 7.2 Hz, 2H), 7.25–7.38 (m, 5H), 7.77 (s, 1H); <sup>13</sup>C NMR (CDCl<sub>3</sub>) δ 13.79, 14.13, 26.15, 50.42, 60.87, 61.18, 127.83, 128.14, 128.46, 129.01, 134.92, 141.44, 167.32, 168.72. **4a**: oil; IR (KBr) 1734, 1707 cm<sup>-1</sup>; <sup>1</sup>H NMR (CDCl<sub>3</sub>) δ 1.15 (t, *J* = 7.2 Hz, 3H), 1.27 (t, *J* = 7.2 Hz, 3H), 2.45–2.51 (m, 2H), 2.77–2.83 (m, 2H), 4.03 (q, *J* = 7.2 Hz, 2H), 4.20 (q, *J* = 7.2 Hz, 2H), 7.27–7.32 (m, 5H), 7.65 (s, 1H); <sup>13</sup>C NMR (CDCl<sub>3</sub>) δ 14.07, 14.19, 22.99, 33.44, 60.32, 60.81, 128.47, 128.49, 129.05, 131.38, 135.21, 140.01, 167.69, 172.61.