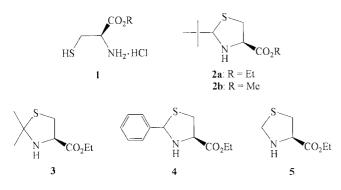
Enantioselective Addition of Diethylzinc to Aldehydes with Easily Prepared Chiral Thiazolidine Catalysts

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Keywords: Enantioselective addition, Diethylzinc, Aldehyde, L-Cysteine ester, Chiral thiazolidine.

Asymmetric metal catalysis is now recognized as the most promising area in the synthesis of optically active organic compounds. One attractive method that leads to the formation of optically active secondary alcohols is catalytic enantioselective addition of dialkylzinc reagent to aldehydes.¹ Numerous elegant and efficient catalysts have been developed for this purpose, in which most of them are based on amino alcohols,² diols,³ diamines,⁴ and their derivatives.⁵ In spite of the variety of chiral catalysts, the develoment of catalysts obtainable by simple synthetic routes remains still as an active research subject. We have easily prepared 4carboethoxy-1,3-thiazolidines 2-5 from L-cysteine ester through one step. Previously, thiazolidine 4 was found to catalyze the dialkylzinc-aldehyde addition.⁶ Following on from previous work, we wish to report the results on the enantioselective dialkylzinc-aldehyde addition in the presence of several N,S-chelate thiazolidine ligands.



Chiral thiazolidines 2-5 were easily prepared by condensation of L-cysteine ester 1 with the corresponding carbonyl compounds. Thiazolidines 2 and 4 were obtained as inseparable ca. 1:1.9 diastereomeric mixtures.⁷ (2R,4R)-cis-Thiazolidines were assigned to be major diastereomers on a combination of the ¹H NMR data and the previous studies.⁸ The catalytic behavior of the ligands 2-5 for the diethylzincaldehyde addition was then examined. The reaction conditions and results are summarized in Table 1. All the aldehydes were converted to the corresponding (S)-alcohols with moderate to excellent enantiomeric excess in high yields. Enantioselectivity was somewhat dependent on the amount of the ligand. Increasing the amount of ligand from 4 to 8% led to a small increase in the enantioselectivity (entries 1-3) and 6 mol% of ligand was enough to afford good enantioselectivity and reactivity. It is noteworthy that the ee

]	RCHO + Et ₂ Zn		2 - 5	$\rightarrow R \xrightarrow{(S)}_{Et} H$	
Entry	\mathbb{R}^1	Ligand (mol%)	Time (h)	Yield ^b (%)	e.e. ^c (%)
1^d	Ph	2a (4)	4	95	87
2	Ph	2a (6)	4	99	90
3	Ph	2a (8)	4	99	90
4^d	p-ClC ₆ H ₄	2a (6)	4	100	95
5	<i>p</i> -MeOC ₆ H ₄	2a (6)	4	99	94
6	o-MeOC ₆ H ₄	2a (6)	4	100	90
7	$cyclo-C_6H_{11}$	2a (6)	4	100	92
8	2-naphtyl	2a (6)	4	100	90
9	Ph	2b (6)	4	98	86
10	p-ClC ₆ H ₄	2b (6)	4	100	92
11	Ph	3 (4)	6	94	87
12	Ph	3 (6)	6	99	90
13	p-ClC ₆ H ₄	3 (6)	6	99	90
14	<i>p</i> -MeOC ₆ H ₄	3 (6)	6	100	82
15	Ph	4 (6)	12	84	80
16	p-ClC ₆ H ₄	4 (6)	6	98	87
17	<i>p</i> -MeOC ₆ H ₄	4 (6)	12	90	76
18	Ph	5 (6)	12	90	67
19	p-ClC ₆ H ₄	5 (6)	12	95	72

^{*a*}Reactions were carried out in hexane at 0 °C \rightarrow RT. Molar ratio, Et₂Zn : aldehyde : ligand = 2 : 1 : 0.04-0.08. Absolute configuration was assigned by the sign of the optical rotation and elution order from a chiral column. ^{*b*}Measured as %-conversion into the product by GC. ^{*c*}determined by HPLC analysis (chiralcel OD column) or GC analysis (β -DEX chiral column).

of the product depends on the size of substituents on thiazolidine ring. Thiazolidine **2a** having larger carboalkoxy moiety showed higher enantioselectivity than **2b** (entries 2, 4 versus 9, 10). Although R group affects the enantioselectivity, the bulkiness of C2-substituent on the thiazolidine ring is more important. Higher enantioselectivity was obtained with increasing size of the C2-substituent. Introduction of *tert*-butyl group on C2 led to the most improved enantioselectivity in the range of 90-95% ee. It is interesting that high enantioselectivity could be obtained with diastereomeric mixture **2**. This result suggests that the chirality of substituent on C2 is less important role for the stereochemical outcome. Each of diasteromers that differ in the

 Table 1. Enantioselective Addition of Diethylzinc to Aldehydes^a

configuration at C2 would generate similar asymmetric induction environment. Accordingly, diastereomerically pure ligand is not necessarily required in order to obtain high enantioselectivity.^{2d,9}

In conclusion, we have prepared chiral thiazolidines 2-5 derived from L-cysteine ethyl ester. In particular 2a could catalyze efficiently the enantioselective addition of dialkyl-zinc to aldehydes.

Experimental Section

Reactions were carried out under an inert nitrogen atmosphere using dried glassware. All the commercially available reagents were used without further purification. NMR spectra were recorded on a Bruker AC 250 NMR spectrometer. Optical rotation were measured with a Perkin-Elmer 241 polarimeter. IR spectra were obtained on a Bruker IFS-48.

Preparation of 2-substituted-1,3-thiazolidines-4-carboxylate 2-5. Thiazolidines $2a^{6.8}$ and 4^9 have been previously reported. According to the previous method, 2-5 were prepared by the reaction of L-cysteine ester hydrochloride with the corresponding carbonyl compounds.

2-tert-Butyl-4-carboethoxy-1,3-thiazolidine (2a). Yield 93%; colorless liquid; $[\alpha]_D^{20}$ -153.2 (c=1.0, CHCl₃); major diastereomer: ¹H-NMR (CDCl₃, 250 MHz) δ 4.40 (s, 1H, SC*H*N), 4.18 (q, *J* = 7.3 Hz, 2H, OCH₂CH₃), 3.72 (dd, *J* = 8.3, 6.8 Hz, 1H, SCH₂C*H*), 3.23 (dd, *J* = 8.4, 6.8 Hz, 1H, SC*H*₂CH), 2.62 (t, *J* = *J*' = 10.0 Hz, 1H, SC*H*₂CH), 2.17 (br, 1H, NH), 1.24 (t, *J* = 7.2 Hz, 3H, OCH₂C*H*₃), 1.02 (s, 9H, C(CH₃)₃); minor diasteromer: 4.40 (s, 1H, SC*H*N), 4.15 (q, *J* = 5.3 Hz, 2H, OCH₂CH₃), 4.04 (t, *J* = *J*' = 6.0 Hz, 1H, SCH₂C*H*), 3.07 (dd, *J* = 8.6, 6.4 Hz, 1H, SC*H*₂C*H*), 2.94 (dd, *J* = 8.2, 6.0 Hz, 1H, SC*H*₂C*H*), 2.17 (br, 1H, NH), 1.23 (t, *J* = 7.2 Hz, 3H, OCH₂C*H*₃), 0.92 (s, 9H, C(CH₃)₃); IR (neat) $\nu_{C=0}$ 1736 cm⁻¹; MS (EI) *m*/*z* 217 (M⁺).

2,2-Dimethyl-4-carboethoxy-1,3-thiazolidine (3). yield 81%; colorless liquid, $[\alpha]_D^{20}$ -178.6 (c=1.4, CHCl₃), ¹H-NMR (CDCl₃, 250 MHz) δ 4.18 (q, J = 7.2 Hz, 2H, OCH₂CH₃), 4.02 (dd, J = 9.1, 7.0 Hz, 1H, SCH₂CH), 3.38 (dd, J = 10.5, 7.0 Hz, 1H, SCH₂CH), 2.95 (br, 1H, NH), 2.96 (dd, J = 10.3, 9.2 Hz, 1H, SCH₂CH), 1.65 (s, 3H, CCH₃), 1.47 (s, 3H, CCH₃), 1.24 (t, J = 7.1 Hz, 3H, OCH₂CH₃); IR (neat) $v_{C=0}$ 1742 cm⁻¹; MS (EI) *m/z* 189 (M⁺).

4-Carboethoxy-1,3-thiazolidines (5). yield 67%; colorless liquid, $[\alpha]_D^{20}$ -145.4 (c=1.1, CHCl₃), ¹H-NMR (CDCl₃, 250 MHz) δ 4.35 (d, J = 9.6 Hz, 1H, SCH₂N), 4.20 (q, J = 7.5 Hz, 2H, OCH₂CH₃), 4.08 (d, J = 9.6, Hz, 1H, SCH₂N), 3.79 (t, J = J' = 7.5 Hz, 1H, SCH₂CH), 3.22 (dd, J = 10.0, 4.5Hz, 1H, SCH₂CH), 2.82 (dd, J = 13.8, 8.0 Hz, 1H, SCH₂CH), 2.32 (br, 1H, NH), 1.26 (t, J = 7.1 Hz, 3H, OCH₂CH₃); IR (neat) $v_{C=0}$ 1736 cm⁻¹; MS (EI) m/z 161 (M⁺). General Procedure for the enantioselective addition of diethylzinc to aromatic aldehydes. To a solution of thiazolidine (0.06 mmol) in hexane (2.0 mL) was added Et₂Zn (2.0 mL, 1.0 M in hexane) at 0 °C. The mixture was allowed to warm to RT. Aldehyde (1.0 mmol) was added dropwise and the reaction mixture was stirred at RT for a given time, observing the progress of the reaction by GC. The reaction was quenched at 0 °C by addition of 1.0 M HCl and the resulting mixture was extracted with CH₂Cl₂. After removal of solvent, the residue was purified by flash chromatography on silica gel (hexane/AcOEt=8/2). The enantiomeric excess was determined by HPLC (chiralcel OD column) or GC (β -DEX chiral column).

Acknowledgment. This work was supported by the Center for Advanced Bioseparation Technology, Inha University.

References

- For reviews: (a) Noyori, R.; Kitamura, M. Angew. Chem. Int. Ed. Engl. 1991, 30, 49. (b) Soai, K.; Niwa, S. Chem. Rev. 1992, 92, 833.
- (a) Sola, L.; Reddy, K. S.; Vidal-Ferran, A.; Moyano, A.; Perica, M. A.; Riera, A.; Alvarez-Larena, A.; Piniella, J. J. Org. Chem. **1998**, 63, 7078. (b) Cho, B. T.; Chun, Y. S. Tetrahedron: Asymmetry **1998**, 9, 1489. (c) Tanner, D.; Korno, T.; Guijarro, D.; Andersson, P. G. Tetrahedron **1998**, 54, 14213. (d) Bolm, C.; Muniz, K.; Hildebrand, J. P. Org. Lett. **1999**, 1, 491. (e) Xu, Q.; Pan, X.; Chan, A.; Yang, T. Tetrahedron Lett. **2001**, 42, 6171.
- (a) Rosini, C.; Franzini, L.; Pini, D.; Salvadori, P. *Tetrahedron:* Asymmetry **1990**, 1, 587. (b) Schmit, B.; Seebach, D. Angew. Chem. Int. Ed. Engl. **1991**, 30, 1321. (c) Weber, B.; Seebach, D. Tetrahedron **1994**, 50, 7473. (d) Huang, W.-S.; Hu, Q.-S.; Pu, L. J. Org. Chem. **1999**, 64, 7940.
- (a) Soai, K.; Niwa, S.; Yamada, Y.; Inoue, H. *Tetrahedron Lett.* 1987, 28, 4841. (b) Rosini, C.; Franzini, L.; Iuliano, A.; Pini, D.; Salvadori, P. *Tetrahedron Asymmetry* 1991, 2, 363. (c) Conti, S.; Falorni, M.; Giacomelli, G.; Soccolini, F. *Tetrahedron* 1992, 48, 8993.
- (a) Yoshioka, M.; Kawakita, T.; Ohno, M. *Tetrahedron Lett.* **1989**, 30, 1657. (b) Katsuji, I.; Kimula, Y.; Okamura, H.; Katsuki, T. *Synlett* **1992**, 573. (c) Kang, J.; Kim, J. W.; Lee, J. W.; Kim, D. S.; Kim, J. L. *Bull. Korean Chem. Soc.* **1996**, *17*, 1135. (d) Vettel, S.; Lutz, C.; Diefenbach, A.; Harderlein, G.; Hammerschmidt, S.; Kuhling, K.; Mofid, M.-R.; Zimmermann, T.; Knochel, P. *Tetrahedron: Asymmetry* **1997**, *8*, 779. (e) Ramon, D. J.; Yus, M. *Tetrahedron: Asymmetry* **1997**, *8*, 2479. (d) Jin, M.-J.; Kim, S.-H.; Jung, J.-A.; Lee, H.-Y. *Bull. Korean Chem. Soc.* **2000**, *21*, 33.
- 6. Kim, S.-H.; Ahn, S.-J.; Chung, S.-T.; Jin, M.-J. J. Ind. & Eng. Chem. 1997, 3, 37.
- The diastereomeric ratio was be determined by the intensity ratio of two singlets for C2 proton in the ¹H NMR spectrum which is attributed to two diastereomers.
- 8. Calmes, M.; Escale, F.; Paolini, F. *Tetrahedron: Asymmetry* **1997**, 8, 3691.
- 9. Brunner, H.; Becker, R.; Riepl, G. Organometallics 1984, 3, 1354.