

Isolation and Structure Determination of Three New Ceramides from the Starfish *Distolasterias nipon*

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Received May 6, 2005

Key Words : Ceramide, *Distolasterias nipon*, 2D NMR, FABMS/MS

Starfishes have been reported to contain a number of glycosphingolipids and sterol compounds with unconventional side chain.¹⁻³ Some glycosphingolipids have exhibited cytotoxicity, histidine decarboxylate inhibition, DNA topoisomerase I inhibition, moderate wound-healing activity and induced apoptotic DNA damage and cell death in mammalian cell lines.⁴⁻⁷ In our search for biologically active and structurally novel lipids from marine organisms, we recently isolated mixtures of several sphingolipids from the starfish *Distolasterias nipon* collected off the coast of East Sea, Korea. Further HPLC of the fraction resulted in the purification of three new ceramides which are composed of the aglycone part of glycosphingolipids. In this paper we report the isolation and structure determination of new ceramides (compound **1**, **2**, and **3**) isolated from the starfish *D. nipon*.

Compound **1** was isolated as a white amorphous solid

which determined for C₃₅H₆₅NO₄ by high resolution FAB-MS. The IR spectrum displayed absorption bands at 3314 (hydroxyl), 1650, and 1540 cm⁻¹ (amide). The ¹H NMR spectrum showed highly overlapped signals at δ 1.29-1.31, indicative of long aliphatic carbon chain, and the ¹³C NMR spectrum suggested the presence of one carbonyl carbon, four carbons attached to heteroatoms, and six olefinic carbons on the basis of their characteristic chemical shifts. All these information of **1** allowed us to deduce a ceramide nature containing three double bonds. This is also supported by the ¹H NMR spectrum measured in pyridine-d₅ which showed a downfield proton signal at δ 8.38 (d, *J* = 8.8 Hz) corresponding to the secondary amide proton (Table 1).

The main framework of compound **1** was established by extensive 2D NMR techniques. The methine proton at δ 3.84, connected to a nitrogen-bearing carbon, was coupled to the nonequivalent oxymethylene protons (H-1) at δ 3.66 and

Table 1. Spectral data for compound **1** in MeOH-d₄ and pyridine-d₅

position	¹ H (MeOH-d ₄)	¹³ C (MeOH-d ₄)	COSY (MeOH-d ₄)	¹ H (pyridine-d ₅)
1a	3.66 (dd, 11.2, 3.9)	61.9, t	1b, 2	4.24 (dd, 10.3, 2.9)
1b	3.78 (dd, 11.2, 5.4)		1a, 2	4.49 (dd, 10.3, 3.9)
2	3.84 (ddd, 7.3, 5.4, 3.9)	56.0, d	3, 1a, 1b	4.71 (m)
3	4.09 (dd, 7.3, 7.3)	73.3, d	2, 4	4.86 (m)
4	5.50 (dd, 15.6, 7.3)	131.5, d	3, 5,	6.09 (dd, 15.6, 6.3)
5	5.72 (dt, 15.6, 6.4)	134.1, d	4, 6	6.00 (dd, 15.6, 5.9)
6	2.10 (dt, 6.4, 7.3)	33.5, t	5, 7	2.17 (m)
7	2.20 (dt, 7.3, 7.3)	28.7, t	6, 8	2.22 (m)
8	5.35 (t, 7.3)	130.3, d	7	5.52 (t, 6.8)
9		135.2, s		
10	6.02 (d, 15.6)	136.1, d	11	6.21 (d, 15.1)
11	5.54 (dt, 15.6, 7.3)	128.5, d	10, 12	5.65 (dt, 15.1, 6.8)
12	2.08 (dt, 7.3, 6.8)	34.0, t		2.10 (m)
13-17	1.29-1.31 (m)	23.7-33.1, t		1.24-1.34 (m)
18	0.89 (t, 6.9)	14.4, q		0.86 (t, 7.3)
19	1.71 (s)	12.8, q		1.76 (s)
NH				8.38 (d, 8.8)
1'		177.2, s		
2'	3.99 (dd, 7.8, 3.9)	73.0, d	3'a, 3'b	4.62 (t, 2.9)
3'a	1.55 (m)	35.9, t	2', 3'b	2.17 (m)
3'b	1.71 (m)		2', 3'a	2.06 (m)
4'-15'	1.29-1.39 (m)	23.7-33.1, t		1.24-1.34 (m)
16'	0.89 (t, 6.9)	14.4, q		0.85 (t, 7.3)

Data were recorded at 500 MHz for ¹H and 125 MHz for ¹³C. Assignments were made by COSY, TOCSY, HSQC, and HMBC experiments.

3.78, and another oxymethine proton (H-3) at δ 4.09 which in turn was coupled to the olefinic proton (H-4) at δ 5.50 in the COSY experiment. The sequential COSY correlations from H-4 to H-8 and the HMBC correlations of the methyl proton (H-19) at δ 1.71 with three neighboring olefinic carbons at δ 130.3 (C-8, CH), 135.2 (C-9, C) and 136.1 (C-10, CH) revealed that one double bond was connected to a diene with a branched methyl at the C-9 position *via* an ethylene group. The existence of the conjugated diene at C-8 was also confirmed by the coupling between H-10 and H-11 and the UV spectrum (λ_{\max} = 235 nm in MeOH). The large coupling constants (J = 15.6, 15.6 Hz) for the olefinic protons in positions C-4, -5 and C-10, -11 and the NOE observations between H-8 and H-10, H-19 and H-7 indicated that three double bonds had the (*E*) configurations. Further analyses of COSY and HMBC experiments showed the existence of α -hydroxy fatty acyl chain within the compound **1**.

In order to determine the length of two alkyl chains of the ceramide, **1** was methanolized to afford a fatty acid methyl ester and a sphingosine. The methyl ester part was characterized to be methyl (*2R*)-hydroxyhexadecanoate (**4**) by analysis of GC-MS (m/z = 286 [M^+]) and the value of optical rotation, $[\alpha]_D^{25}$ -3.8° (c 0.06, CHCl_3), which is very close to that of similar molecules.^{5,8} On the other hand, the sphingosine unit was readily recognized as $\text{C}_{19}\text{H}_{35}\text{NO}_2$

according to a molecular ion $[M+H]^+$ observed at m/z = 310 by FAB-MS. In addition, the sphingosine unit was acetylated with Ac_2O /pyridine for the determination of stereochemistry of the asymmetric centers at the C-2 and -3 positions. As a result, the key proton chemical shifts and coupling constants were very close to those of triacetyl *D-erythro*-sphingosine and also the specific rotation was in agreement with that of the (*2S,3R,4E*)-sphingosine unit reported in literature.^{2,9} Accordingly, **1** was determined as (*2R*)-hydroxy-*N*-palmitoyl-*D-erythro*-(*2S,3R*)-9-methyloctadecaphinga-(*4E,8E,10E*)-trienine (Figure 1).

Compound **2** was given as a colorless amorphous powder and its molecular formula was established as $\text{C}_{38}\text{H}_{75}\text{NO}_5$ by HRFAB-MS. Analyses of the NMR, IR, and HRFAB-MS spectra data of **2** clearly showed that it was a ceramide containing one double bond. The methanolysis on **2** was performed to yield a fatty acid methyl ester and a long chain phytosphingosine moiety with an oxymethine proton (H-4) at δ 3.51 instead of an olefinic proton. The methyl ester was identified as methyl (*2R*)-hydroxyhexadecanoate (**4**) and the phytosphingosine component (**2a**) was revealed to be a docosene derivative on the basis of the molecular ion peak at m/z = 372 [$M + H$]⁺ of FAB-MS and the ^1H NMR spectral data. Compound **2a** was thought to possess one olefinic group somewhere in the long carbon chain, since the olefinic protons were only coupled to the protons overlapped at δ 1.29-1.31 region in the COSY spectrum. The location and geometry of the double bond were determined as follows. The detailed FABMS/MS showed a remarkable fragment ion peaks at m/z = 218 and 273 due to cleavage of bonds at the allylic position of double bond, indicating that the double bond in **2a** is positioned at C-13 (Figure 2). Furthermore, the geometry of this double bond can be determined by the ^{13}C chemical shift (δ 28.2) of the allylic carbon, which was commonly observed in the (*Z*) configurations.¹⁰ On the other hand, stereochemistry of C-2 to C-4 in the phytosphingosine was established in the same way as **1**. The comparison of ^1H NMR data and the optical rotation of tetraacetyl derivative of the base with those of similar compounds allowed us to propose the *2S,3S,4R* configuration¹¹ and the gross structure of **2** was assigned to be (*2R*)-hydroxy-*N*-palmitoyl-*D-erythro*-(*2S,3S,4R*)-docosasphinga-(*13Z*)-monoeneine.

Finally, compound **3** was also obtained as a white solid. Its molecular formula was assigned as $\text{C}_{38}\text{H}_{73}\text{NO}_4$ by HRFAB-MS. The only difference between **2** and **3** was an appearance of one double bond by dehydration of **2**. This double bond was obviously confirmed as (*4E*)-alkene moiety from the COSY correlations from H-2 to H-5 and the large proton coupling constants (J_{45} = 15.1 Hz). On methanolysis of **3**, the methyl ester was recognized as a common methyl (*2R*)-hydroxyhexadecanoate (**4**) and the sphingosine part was given as a docosadiene amino alcohol corresponding to a molecular ion peak at m/z = 354 [$M + H$]⁺ by FAB-MS. Another double bond in the long chain sphingosine was assigned as (*13Z*) configuration by the FAB-MS/MS analyses and carbon chemical shifts in the same way as **2** (Figure 2). Also the ^1H NMR data and the optical rotation of

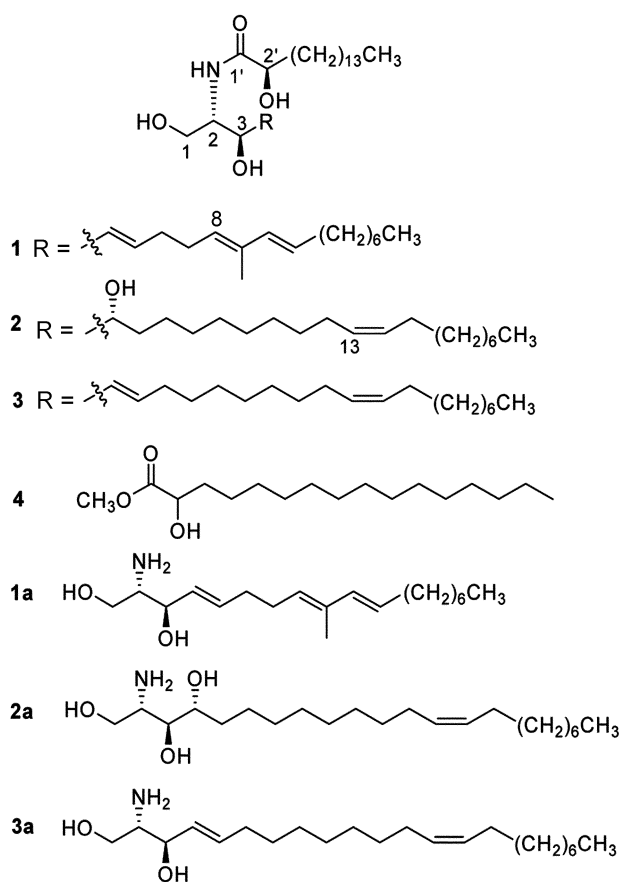


Figure 1. The molecular structure of compounds **1**, **2** and **3** and their derivatives.

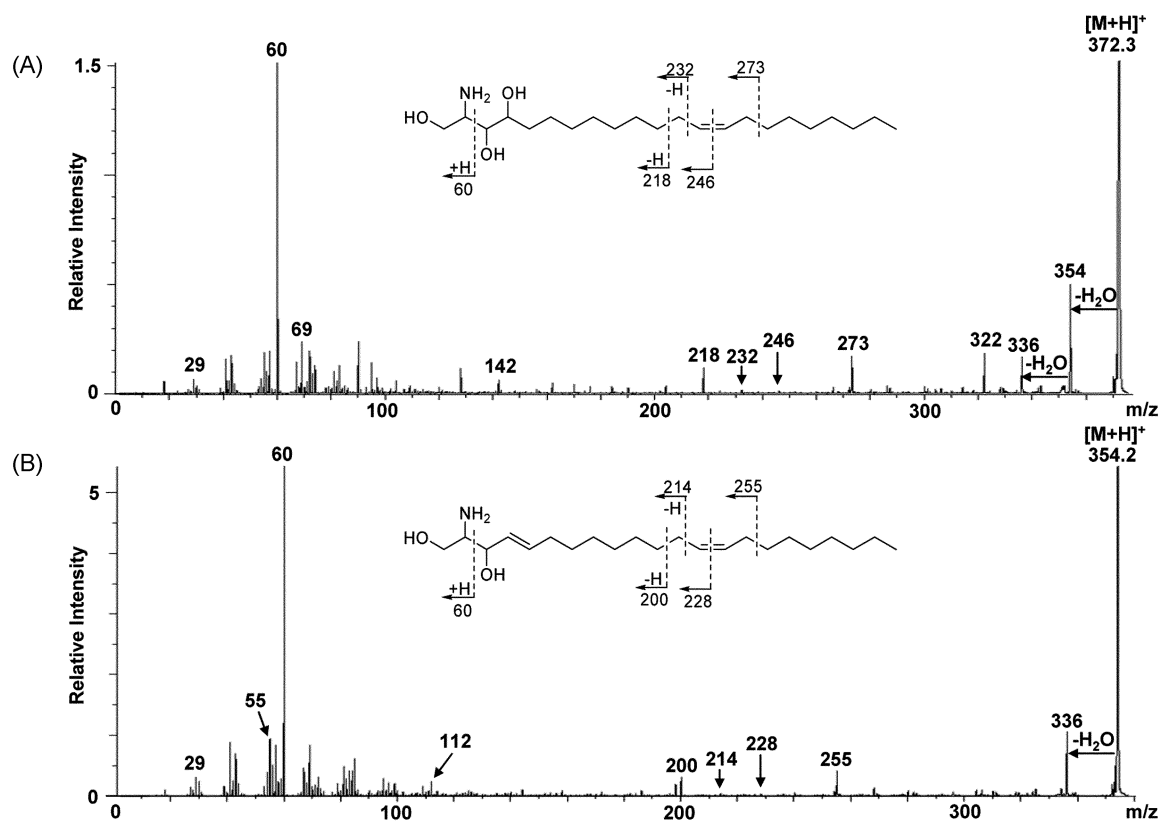


Figure 2. MS/MS spectra and fragmentations of **2a** (A) and **2b** (B).

the acetylated sphingosine derivative agreed well with those of **1**. All these data allowed this compound to be defined as (2'*R*)-hydroxy-*N*-palmitoyl-*D*-erythro-(2*S*,3*R*)-docosasphingina-(4*E*,13*Z*)-dienine.

Compounds **1**, **2**, and **3** are, to the best of our knowledge, new ceramides. For the biological evaluation of them, activity tests are under examination. While several glycosphingolipids from starfish have been studied, the isolation and characterization of ceramides has not been well known.

Experimental Procedures

Optical rotations were measured on a JASCO DIP-1000 digital polarimeter. UV spectra were obtained in MeOH using a Shimadzu UV-1650PC and IR spectra were measured on a Mattson Galaxy spectrometer. All NMR spectra were recorded in CD₃OD solution on a Varian UNITY 500 spectrometer. Chemical shifts of proton and carbon spectra were reported in reference to residual solvent peaks at 3.30 ppm and 49.0 ppm, respectively. GC-MS experiments were done on a Hewlett Packard HP6890. FAB-MS and MS/MS spectra were obtained on JEOL JMS-HX110/110A four sector tandem mass spectrometer. The details about mass spectrometric methods were described elsewhere.¹²

Extraction and Isolation. The starfish *Distolasterias nipon* was collected at Donghae in the East Sea of Korea. The frozen organism (1.2 Kg) was cut into small pieces and

extracted twice with MeOH at room temperature. The methanolic extract (*ca.* 140 g) was partitioned between MeOH and *n*-hexane. The *n*-hexane layer was subjected to silica gel flash column chromatography eluting with the solvents of increasing polarity (hexane, EtOAc, acetone, and MeOH). The acetone-soluble fraction (*ca.* 800 mg) was separated by reversed-phase HPLC (YMC ODS-A column, 250 × 10 mm ID) eluting with 100% MeOH to give eight fractions. The fractions containing compounds **1**, **2**, and **3** were further purified on reversed-phase HPLC using CH₃CN-MeOH (2 : 8) to afford compounds **1** (12 mg), **2** (34 mg), and **3** (14 mg).

Methanolysis and Acetylation. Each compound (3 mg) was heated with 3 mL of 1 N HCl in 80% MeOH at 70 °C overnight in a sealed small vial. The reaction mixture was evaporated and then partitioned with *n*-hexane and MeOH. The hexane layer was concentrated to afford the methyl ester. The sphingosine obtained from the MeOH layer was further acetylated with Ac₂O/pyridine (1 : 4, 0.5 mL) at 80 °C for 2 hrs. After work-up with MeOH and solvent drying procedures, the mixture was partitioned into CH₂Cl₂ and H₂O. The CH₂Cl₂ layer was subjected to silica column to furnish an acetyl derivative of sphingosine.

Compound 1: [α]_D²⁵ +3.30° (c 0.21, MeOH); UV (MeOH) λ_{\max} (log ϵ) 235 nm (4.53); IR (KBr) 3317, 2918, 2849, 1650, 1540, 1450, 1409 cm⁻¹; ¹H and ¹³C NMR data were given in Table 1; HRFAB-MS [M + Na]⁺ *m/z* 586.4808 (Δ -0.4 mmu).

Acetyl derivative of 1a: $[\alpha]_{\text{D}}^{25} -6.6^\circ$ (c 0.13, CHCl_3); ^1H NMR (CDCl_3 , 500 MHz), δ 5.79 (1H, ddd, $J = 15.6, 6.8, 6.8$ Hz, H-5), 5.68 (1H, d, $J = 9.3$ Hz, -NH), 5.38 (1H, dd, $J = 15.6, 6.8$ Hz, H-4), 5.28 (1H, dd, $J = 6.8, 6.8$ Hz, H-3), 4.44 (1H, m, H-2), 4.31 (1H, dd, $J = 11.2, 5.9$ Hz, H-1), 4.04 (1H, dd, $J = 11.2, 3.9$ Hz, H-1), 2.07 (6H, s, CH_3COO), 1.99 (3H, s, CH_3CONH).

Compound 2: $[\alpha]_{\text{D}}^{25} +8.63^\circ$ (c 0.21, MeOH); IR (KBr) 3313, 2919, 2850, 1634, 1545, 1457 cm^{-1} ; ^1H NMR (CD_3OD , 500 MHz), δ 5.33 (2H, t, $J = 4.9$ Hz, H-14, 15), 4.09 (1H, ddd, $J = 4.5, 4.5, 5.9$ Hz, H-2), 4.02 (1H, dd, $J = 7.3, 3.9$ Hz, H-2'), 3.74 (2H, d, $J = 4.5$ Hz, H-1), 3.56 (1H, dd, $J = 5.9, 5.9$ Hz, H-3), 3.51 (1H, ddd, $J = 8.7, 5.9, 2.9$ Hz, H-4), 2.02 (4H, m, H-13, 16), 1.73 (1H, m, H-3'), 1.64 (1H, m, H-5), 1.58 (1H, m, H-3'), 1.41 (1H, m, H-5), 1.42-1.28 (48H, m, - CH_2 -), 0.89 (6H, t, $J = 6.4$ Hz, H-16', 18); ^{13}C NMR (CD_3OD , 125 MHz) δ 176.9 (C, C-1'), 130.8 (CH, C-14, 15), 76.0 (CH, C-3), 73.2 (CH, C-4), 72.9 (CH, C-2'), 35.0 (CH_2 , C-3'), 33.0 (CH_2 , C-5), 30.9-30.5 (CH_2 , - CH_2 -), 28.2 (CH_2 , C-13, 16), 14.4 (CH_3 , C-16', 18); HRFAB-MS $[\text{M} + \text{Na}]^+ m/z$ 648.5543 ($\Delta +0.2$ mmu).

Acetyl derivative of 2a: $[\alpha]_{\text{D}}^{25} = +12.6^\circ$ (c 0.09, CHCl_3); ^1H NMR (CDCl_3 , 500 MHz), δ 5.74 (1H, d, $J = 9.3$ Hz, NH), 5.11 (1H, dd, $J = 8.3, 2.9$ Hz, H-3), 4.94 (1H, dd, $J = 9.7, 2.9$ Hz, H-4), 4.47 (1H, m, H-2), 4.30 (1H, dd, $J = 11.2, 4.9$ Hz, H-1), 4.00 (1H, dd, $J = 11.2, 3.4$ Hz, H-1), 2.08 (3H, s, CH_3COO), 2.05 (6H, s, CH_3COO), 2.03 (3H, s, CH_3CONH).

Compound 3: $[\alpha]_{\text{D}}^{25} +4.98^\circ$ (c 0.30, MeOH); IR (KBr) 3313, 2919, 2850, 1634, 1545, 1457 cm^{-1} ; ^1H NMR (CD_3OD , 500 MHz), δ 5.70 (1H, dt, $J = 15.1, 6.8$ Hz, H-5), 5.46 (1H, dd, $J = 15.1, 7.3$ Hz, H-4), 5.34 (2H, t, $J = 4.9$ Hz, H-14, 15), 4.07 (1H, dd, $J = 7.3, 7.3$ Hz, H-3), 3.98 (1H, dd, $J = 7.8, 3.9$ Hz), 3.84 (1H, ddd, $J = 7.3, 4.9, 3.9$ Hz, H-2), 3.78 (1H, dd, $J = 11.2, 4.9$ Hz, H-1), 3.66 (1H, dd, $J = 11.2, 3.9$ Hz), 2.02 (4H, m, H-13, 16), 2.01 (2H, m, H-6), 1.70 (1H, m, H-3'), 1.54 (1H, m, H-3'), 1.42-1.28 (46H, m, - CH_2 -), 0.89 (6H, t, $J = 5.9$ Hz, H-16', 18); ^{13}C NMR (CD_3OD , 125 MHz) δ 177.1 (C, C-1'), 134.9 (CH, C-5), 131.1 (CH, C-4), 130.8 (CH, C-14, 15), 73.3 (CH, C-3), 73.0 (CH, C-2'), 62.0 (CH_2 , C-1), 56.0 (CH, C-2), 35.9 (CH_2 , C-

3'), 33.5 (CH_2 , C-6), 30.9-30.5 (CH_2 , - CH_2 -), 28.2 (CH_2 , C-13, 16), 14.5 (CH_3 , C-16', 18); HRFAB-MS $[\text{M} + \text{Na}]^+ m/z$ 630.5438 ($\Delta +0.1$ mmu).

Acetyl derivative of 3a: $[\alpha]_{\text{D}}^{25} -7.2^\circ$ (c 0.16, CHCl_3); ^1H NMR (CDCl_3 , 500 MHz), δ 5.79 (1H, ddd, $J = 15.2, 6.8, 6.8$ Hz, H-5), 5.65 (1H, d, $J = 9.5$ Hz, -NH), 5.37 (1H, dd, $J = 15.2, 6.8$ Hz, H-4), 5.29 (1H, dd, $J = 6.8, 6.8$ Hz, H-3), 4.43 (1H, m, H-2), 4.31 (1H, dd, $J = 11.3, 5.8$ Hz, H-1), 4.04 (1H, dd, $J = 11.3, 3.9$ Hz, H-1), 2.07 (6H, s, CH_3COO), 1.99 (3H, s, CH_3CONH).

Acknowledgment. This research was financially supported by the Korea Science and Engineering Foundation (No. R01-2001-000-00211-0) and MarineBio21, Ministry of Maritime Affairs and Fisheries, Korea.

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