## A Novel Approach to 1-β-Methylcarbapenem Intermediate Utilizing Lithium Enolate Dianion of 2'-Hydroxypropiophenone

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Received August 16, 1995

Since the discovery of thienamycin, intensive efforts have been devoted for the synthesis of chemically and metabollically more stable cabapenem nucleus with enhanced biological activity, and led to the developement of 1-β-methylcarbapenems 1.1 A number of stereoselective syntheses of 2, a well-established intermediate of 1, have been made mainly from commercially available 4-acetoxy-2-azetidinone 3a.2 Conversion of 4-acetoxy group of 3a into 4-[(R)-1-carboethoxyethyl group of 2 by aldol-type reactions with metal enolates of propionate derivatives has been a very popular approach. Since the stereoselectivity at the C-1 position depends on the structure of enolate, various chiral and achiral auxiliaries were devised, including 3-thiazolidine-2-thione,3 2-oxazolidinone,4 2-picolyl thiol,5 and 2,3-dihydro-4-H-1,3-benzoazin-4ones. 6.7 Although high diastereoselectivities were achieved in introducing the 1-methyl group, these auxiliaries are difficult of access, and such convedrsion requires expensive reagents, like Sn(OTf)2 and Et2BOTf. Thuse, a more practical method to provide 2 is still in demand.

OH H H CH<sub>3</sub> TBSO H H CO<sub>2</sub>H TBSO H H X CO<sub>2</sub>H 
$$\frac{1}{2}$$
 3a: X = OAc 3b: X = OBz

In connection with this, we became interested in the reactions of **3a** or **3b** with metal enolates of readily avaiable 4'-methoxy- and 2'-hydroxypropiophenone. Since the oxygen-substituted benzoyl moiety in the condensation product should be selectively oxidized to the corresponding ester or carboxylic acid, we first studied the stereochemical outcome of this aldol-type condensation of ketone enolate. Trimethylsilyl enol ether of 4'-methoxypropiophenone was reac-

ted 3a in the presence of  $ZnI_2$  to give 5a in 97% yield with a dominant  $\alpha$ -diastereomer (entry 1, Table 1).8 Reaction of 3a with two equivalents of lithium enolate 4'-methoxypropiophenone, generated with LDA in THF at -78 °C, provided an 1:1.9 mixture of  $5a\alpha$  and  $5a\beta$  in 85% yield (entry 2).9 Tin(II) enolate, generated *in situ* from  $\alpha$ -bromo-4'-methoxypropiophenone and tin powder, displayed an improved  $\beta$ -selectivity (entry 3).10 This marginal selectivity could be further

improved by the introduction of enolate dianion of 2'-hydroxypropiophenone. Thus, two equivalents of lithium enolate dianion of 2'-hydroxy propiophenone was reacted with 3a to give an 1:3.7 mixture of 5bα and 5bβ in 56% yield (entry 4). These diastereomers were easily separated by recrystallization and SiO2 chromatography. Even with excess of the dianion, the yield was moderate and unreacted 3a (about 30%) was recovered. We speculated that the increased bascity of the enolate dianion of 2'-hydroxypropiophenone caused transenolization with 3a and was responsible for the recovered 3a. Indeed, when 4-benzovloxy 3b instead of 4acetoxy 3a was used, the yield was improved to 82% (entry 5. Table 1). The enhanced diastereoselectivity in the formation of 5 using the enolate dianion can be explained by the tight coordination in the six-membered transition stayte, like 6, between the acylimine generated from 3a and Z-enolate dianion 4.11

While the diastereomeric mixture of 4'-methoxyphenyl ketone 5a was smoothly oxdized to the corresponding ester with m-CPBA, the same reaction with diastereomerically pure  $5b\beta$  was extremly slow. Direct conversion of 2'-hydroxybenzoyl group of  $5b\beta$  to carboxylic acid 2 was accomplished with 30% hydrogen peroxide and 2 N LiOH in THF in 65% yield after reductive work-up with dimethyl sulfide and recr-

Table 1. Condensations of Metal Enolates of 4'-Methoxy-and 2'-Hydroxypropiophenone with 3a or 3b

entry	M	X	3a/3b	condition	$5\alpha:5\beta$	yield (%)
1	SiMe <sub>3</sub>	4-OCH <sub>3</sub>	3a	ZnI <sub>2</sub> , CH <sub>2</sub> Cl <sub>2</sub> , rt	3.5 : 1	97
2	Li	4-OCH₃	- 3a	THF, $-78$ °C, 1 h	1:1.9	85
3	SnBr	4-OCH <sub>3</sub>	3a	DMF-CH <sub>2</sub> Cl <sub>2</sub> , Sn-AgBF <sub>4</sub> -I <sub>2</sub>	1:2.8	72
4	Li	2-OLi	3a	THF, −78 °C, 2h	1:3.7	56
5	Li	2-OLi	3b	THF, −78 °C, 2h	1:3.8	82

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In summary, we studied diastereoselective reactions between 3a or 3b with metal enolates of readily available propiophenone derivatives. Direct oxidative conversion of the resulting 2'-hydroxybenzoyl derivative  $5b\beta$  to  $1\beta$ -methylcar-bapenem intermediate 2 was developed.

## **Experimental Section**

 $(3S,4R)-3-\lceil (R)-1-(t-Butyldimethylsilyloxy)ethyl \rceil-4 \lceil (R)-1-(2-hvdroxvbenzovi) \rceil-2-azetidinone (5bb)$ . To a solution of diisopropylamine (1.80 mL, 12.8 mmol) in 30 mL of dry THF at -78 °C was added 1.6 M n-BuLi in hexane (8.0 mL, 12.8 mmol). After 1 h, a solution of 2'-hydroxypropiophenone (0.88 mL, 6.4 mmol) in 5 mL of THF was added to LDA solution via cannula and the mixture was stirred at −78 °C for 1 h. A solution of 4-benzoyloxy-β-lactam 3b (1.04 g, 3.15 mmol) in 5 mL of THF was added to the enolate solution, and the resulting solution was stirred for 2 h at the temperature and quenched by adding 150 mL of sat. NH<sub>4</sub>Cl solution. The aqueous layer was extracted twice with 100-mL portions of ethyl acethyl acetate and the combined organic layers were washed with 100 mL of brine, dried over MgSO<sub>4</sub> and concentrated in vacuo. The residue was flash chromatographed twice (ethyl acetate: hexane=1:4, 1:2) to give 203 mg (17%) of  $5b\alpha$  and 771 mg (65%) of **5b**β. Isomer **5b**α: mp 165-167 °C;  $R_f$ =0.42 (ethyl acetate: hexane = 1:1;  $[\alpha]_D = +90.3$  (c=1.0, EtOH); IR (KBr) 1762, 1638 cm<sup>-1</sup>; <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>) δ 0.09 (s, 3H, CH<sub>3</sub>Si), 0.10 (s, 3H, CH<sub>3</sub>Si), 0.90 (s, 9H, t-Bu), 1.29 (d, J=7 Hz, 3H, CH<sub>3</sub>), 1.35 (d, J=6 Hz, 3H, CH<sub>3</sub>), 2.89 (dd, J=6, 2 Hz, 1H, CHCO), 3.57 (dq, J=9, 7 Hz, 1H, C(1)H), 3.95 (dd, J=9, 2 Hz. 1H, CHN), 4.22 (quint., J=6 Hz, CHOSi), 5.80 (br s, 1H, NH), 6.91-7.75 (m, 4H, ArH), 12.2 (s, 1H, OH); MS m/z (relative intensity 320 (M+-t-Bu, 6), 200 (14), 195 (5), 161 (4), 121 (43), 93 (6), 86 (7), 72 (45), 59 (100), 55 (24). Isomer **5b**β: mp 163 °C;  $R_f = 0.35$  (ethyl acetate: hexane = 1:1);  $[\alpha]_D = -77.1$  (c=1.4, EtOH); IR (KBr) 1761, 1630 cm<sup>-1</sup>; <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>) & 0.05 (s, 3H, CH<sub>3</sub>Si), 0.07 (s, 3H, CH<sub>3</sub>Si), 0.86 (s, 9H, t-Bu), 1.15 (d, J=6 Hz, 3H, CH<sub>3</sub>), 1.34 (d, J=5 Hz, 3H, CH<sub>3</sub>), 2.91 (dd, J=5, 2 Hz, 1H, CHCO), 3.74 (dq, J=6, 5 Hz, 1H, C(1)H), 3.99 (dd, J=5, 2 Hz, 1H, CHN), 4.17 (quint., J=6 Hz, CHOSi), 6.60 (br s, 1H, NH), 6.90-7.76 (m, 4H, ArH), 12.3 (s, 1H, OH); MS m/z (relative intensity) 320 (M<sup>+</sup>-t-Bu, 9), 200 (20), 195 (10), 161 (4), 121 (58), 93 (8), 86 (6), 72 (45), 59 (100), 55 (23).

(3S,4R)-3-[(R)-1-(t-Butyldimethylsilyloxy) ethyl]-4[(R)-1-carboxyethyl]-2-azetidinone (2). To a solution of 5bβ (710 mg, 188 mmol) in 20 mL of THF was added 4 mL of 30% hydrogen peroxide and 4 mL of 2 N LiOH followed by stirring for 2h at room temperature under atmosphere. Dimethyl sulfide (4 mL) was added, and the resulting mixture was stirred for 2 h at room temperature followed by addition 100 mL of 1 N HCl solution. The aqueous layer was extracted three times with 100-mL portions of ethyl acetate and the combined organic layers were washed with 100 mL of brine and dried over MgSO<sub>4</sub>. The solvent was removed under reduced pressure and the residue was flash chro-

matographed (ethyl acetate: hexane=1:1 and 2:1) to give 369 mg (65%) of **2** after recrystallization in ethyl acetate and hexane: mp 142-143  $^{\circ}$ C (lit. 140-143  $^{\circ}$ C). <sup>1a</sup> Spectral data (<sup>1</sup>H NMR, IR) of **2** are identical with those reported.

**Acknowledgment.** This study was supported by Korea Science and Engineering Foundation (92-25-00-08) and Korea University. Mass analyses at Organic Chemistry Research Center are also gratefully acknowledged.

## References

- (a) Shih, D. H.; Baker, F.; Cama, L.; Christensen, B. G. Heterocycles 1984, 21, 29.
  (b) Guchikonda, R. N.; Cama. L. D.; Queseda, M.; Woods, M. F.; Salzmann, T. N.; Christensen, B. G. J. Med. Chem. 1987, 30, 871.
- 4-Acetoxy-2-azetidinone 3a is available from Kanegafuchi Chemical Industry Co., Ltd., Osaka, Japan.
- Nagao, Y.; Kumagai, T.; Tamai, S.; Abe, T.; Kuramoto, Y.; Taga, T.; Aoyagi, S.; Nagase, Y.; Ochial, M.; Inoue, Y.; Fujita, E. J. Am. Chem. Soc. 1986, 108, 4673.
- Fuentes, L. M.; Shinkai, I.; Saltzmann, T. N. J. Am. Chem. Soc. 1986, 108, 4675.
- Martel, A.; Daris, J.-P.; Bachand, C.; Corbeil, J.; Menard, M. Can. J. Chem. 1988, 66, 1537.
- Kondo, K.; Seki, M.; Kuroda, T.; Yamanaka, T.; Iwasaki, T. J. Org. Chem. 1995, 60, 1095.
- Alternative methods to prepare 1-β-methylcarbapenem intermediates. (a) Iimori, T.; Shibasaki, M. Tetrahedron Lett. 1986, 27, 2149. (b) Fuentes, L. M.; Shinkai, I.; King, A.; Purick, R.; Reamer, R. A.; Schmitt, S. M.; Cama, L.; Christensen, B. G. J. Org. Chem. 1987, 52, 2563. (c) Kim, C. U.; Luh, B.; Partyka, R. A. Tetrahedron Lett. 1987, 28, 507. (d) Uyeo, S.; Itani, H. Tetrahedron Lett. 1991, 32, 2143. (e) Bender, D. R.; DeMarco, A. M.; Melillo, D. G.; Riseman, S. M.; Shinkai, I. J. Org. Chem. 1992, 57, 2411. (f) Murayama, T.; Yoshida, A.; Kobayashi, T.; Miura, T. Tetrahedron lett. 1994, 35, 2271. (g) Choi, W. B.; Churchill, H. R. O.; Lynch, J. E.; Thompson, A. W.; Humphrey, G. R.; Volante, R. P.; Reider, P. J.; Shinkai, I. Tetrahedron Lett. 1994, 35, 2275.
- 8. Trimethylsilyl enol ether of propiophenone gave similar results, but Baeyer-Villiger reaction of the resulting ketone was not regioselective.
- 9. For comparison, lithium enolate of methyl propionate was reacted with 3a to give mixture  $(\alpha:\beta=2:1)$  of methyl ester of 2 in 66% yield.
- 10. Deziel, R.; Endo, M. Tetrahedron Lett. 1988, 29, 61.
- 11. The assignment of Z-geometry on enolate dianion of 2'-hydroxypropiophenone is tentaytive, and based on geometry of enolate from propiophenone and stereochemical distribution of condensation products in our study. See Heathcock, C. H.; Buse, C. T.; Kleschick, W. A.; Pirrung, M. C.; Sohn, J. E.; Lampe, J. J. Org. Chem. 1980, 45, 1066.
- (a) Dakin, H. D. Org. Synth. Coll. Vol. 1 1941, 149. (b) von Wacek, A.; Eppinger, H. O. Chem. Ber. 1940, 73, 644.