

SYNTHESIS AND RADIOPROTECTIVE ACTIVITY OF NEW ORGANOSILICON AND GERMANIUM COMPOUNDS

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Abstract

Silathiazolidine and metalladithioacetals (M = Si, Ge) have been prepared by the interaction of dialkyldichloro- or bis(diethylamino)dialkylsilanes and -germanes with 3-[N-(2-thioethyl)]amino-propanamide (WR-2529) and [1-thioethyl-2-(1-naphthylmethyl)]-2-imidazoline. The study of these compounds in the field of chemical radioprotection has shown a notable decrease in the toxicity and a rather large increase in the radioprotective activity of these new derivatives in comparison with the starting organic compounds.

Introduction

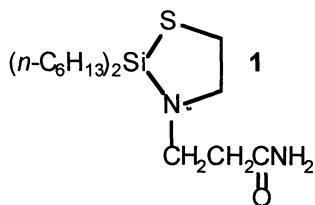
Current interest in the radioprotective activity of several classes of organosilicon and organogermanium derivatives is attested by a growing number of reported syntheses in this area [1-10]. This report concerns the synthesis, toxicity and study of the biological activity of some new silathiazolidine, sila- and germadithioacetals (see scheme).

Materials and methods

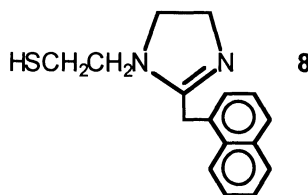
All manipulations were carried out under dry nitrogen. Solvents were freshly distilled from sodium/benzophenone before use. IR spectra were recorded on a Perkin-Elmer 1600FT-IR spectrophotometer. ¹H and ¹³C NMR spectra were recorded on Bruker's AC 80 (80.13 MHz) and AC 200 (50.32 MHz) spectrometers; the multiplicity of the ¹³C NMR signals was determined by the APT technique and quoted (-) for CH₃ and CH, (+) for CH₂ and (C_{quat}) for quaternary carbon atoms. Mass spectra under electron impact (EI) conditions at 70 and 30 eV were recorded on a Hewlett-Packard 5989A spectrometer. Elemental analyses (C, H, N) were performed at the Laboratoire de Microanalyse de l'Ecole Nationale Supérieure de Chimie de Toulouse.

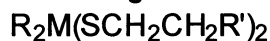
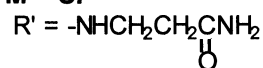
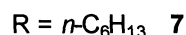
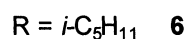
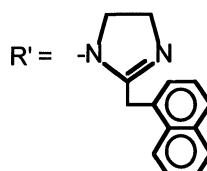
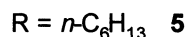
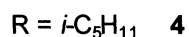
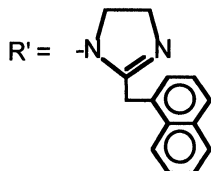
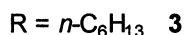
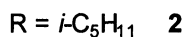
Scheme

Silathiazolidine



[1-Thioethyl-2-(1-naphthylmethyl)]-2-imidazoline



Sila- and germadithioacetals**M = Si****M = Ge****Silathiazolidine 1**

This compound was prepared by two methods: A and B

Method A

Di-*n*-hexyldichlorosilane (3.91 g, 14.51 mmol) in 50 ml of THF was added dropwise to a stirred mixture of 3-[N-(2-thioethyl)]aminopropanamide [11] (2.15 g, 14.51 mmol) and freshly distilled triethylamine (3.23 g, 31.92 mmol) in 70 ml of THF. The reaction mixture was refluxed for 2 h with stirring. After cooling, the mixture was filtered under nitrogen to remove the precipitate Et₃N.HCl. Removal of volatiles (under reduced pressure) from the filtrate, the residue was extracted by 40 ml of dry pentane. Filtration and concentration, afforded **1** (3.88 g, 78 %).

Method B

To a stirred mixture of 3-[N-(2-thioethyl)]aminopropanamide [11] (1.60 g, 10.80 mmol) in 50 ml of THF was added dropwise, a solution of bis(diethylamino)di-*n*-hexylsilane (3.70 g, 10.80 mmol) in 50 ml of THF. The mixture was refluxed under nitrogen for 3 h. The volatile material was removed *in vacuo* to afford the compound **1** (3.57 g, 96 %).

Sila- and germadithioacetals 2-7

These compounds were also synthesized by two methods: C and D

Method C

Diisoamyldichlorosilane (1.82 g, 7.55 mmol) in 30 ml of THF was added dropwise to a stirred mixture of 3-[N-(2-thioethyl)]aminopropanamide (2.24 g, 15.11 mmol) and freshly distilled triethylamine (1.68 g, 16.62 mmol) in 50 ml of THF. After refluxing for 3 h, the resulting mixture was cooled down to room temperature, filtered and evaporated under vacuum. The residue was extracted in 30 ml of dry pentane. Filtration, followed by removal of the solvent under vacuum gave **2** (1.99 g, 57 %).

Method D

Bis(diethylamino)di-*n*-hexylsilane (2.00 g, 5.84 mmol) in 40 ml of THF was added dropwise with stirring to a suspension of HSCH₂CH₂NHCH₂CH₂C(O)NH₂ (1.73 g, 11.67 mmol) in 70 ml of THF. The mixture was refluxed under nitrogen for 3 h. After cooling down to room temperature, the volatiles were removed under vacuum to afford **3** (2.02 g, 70 %).

Compounds **4-7** were prepared analogously from the appropriate dialkyldichlorometallane or bis(diethylamino)dialkylmetallane and 3-[N-(2-thioethyl)]aminopropanamide or [1-thioethyl-2-(1-naphthylmethyl)]-2-imidazoline.

Table I - Physicochemical data and analyses of derivatives 1-8

Compound	Method of synthesis	Yield (%)	Physical Properties and Elemental Analyses
1	A or B	96	¹ H NMR (CDCl ₃ ; δ, ppm): 0.84 (t, 6H, J = 5.56 Hz); 0.91 (t, 4H, J = 7.55 Hz); 1.08-1.52 (m, 16H); 2.29-3.02 (m, 8H). ¹³ C NMR (CDCl ₃ ; δ, ppm): 13.41 (+); 14.09 (-); 14.51 (+); 22.35 (+); 22.55 (+); 31.49 (+); 32.76 (+); 35.40 (+); 44.84 (+); 51.73 (+); 175.25 (C=O). I.R. (CCl ₄ ; cm ⁻¹): ν _{C=O} = 1674; ν _{NH2} = 3182, 3318. Mass spectrum: m/z = 273 [M - 71] ⁺ . Analysis (C ₁₇ H ₃₆ N ₂ OSSi) Calcd %: C, 59.25; H, 10.53; N, 8.13. Found %: C, 59.16; H, 10.51; N, 8.06.
2	C or D	57	¹ H NMR (CDCl ₃ ; δ, ppm): 0.85 (d, 12H, J = 5.8 Hz); 0.97-1.56 (m, 10H); 2.31-2.98 (m, 16H). ¹³ C NMR (CDCl ₃ ; δ, ppm): 10.95 (+); 21.99 (-); 24.76 (+); 30.49 (-); 31.19 (+); 35.42 (+); 44.84 (+); 51.71 (+); 170.76 (C=O). I.R. (CCl ₄ ; cm ⁻¹): ν _{C=O} = 1727; ν _{NH2} = 3202, 3344. Mass spectrum: m/z = 317 [M - 147] ⁺ . Analysis (C ₂₀ H ₄₄ N ₄ O ₂ S ₂ Si) Calcd %: C, 51.68; H, 9.54; N, 12.05. Found %: C, 51.58; H, 9.71; N, 11.95.
3	C or D	70	¹ H NMR (CDCl ₃ ; δ, ppm): 0.85 (t, 6H, J = 5.7 Hz); 0.92 (t, 4H, J = 7.55 Hz); 1.07-1.58 (m, 16H); 2.35-3.35 (m, 16H). ¹³ C NMR (CDCl ₃ ; δ, ppm): 14.17 (-); 15.67 (+); 22.65 (+); 23.02 (+); 29.45 (+); 31.61 (+); 32.99 (+); 37.45 (+); 43.82 (+); 52.07 (+); 170.18 (C=O). I.R. (CCl ₄ ; cm ⁻¹): ν _{C=O} = 1664; ν _{NH2} = 3189, 3259. Mass spectrum: m/z = 345 [M - 147] ⁺ . Analysis (C ₂₂ H ₄₈ N ₄ O ₂ S ₂ Si) Calcd %: C, 53.61; H, 9.81; N, 11.37. Found %: C, 53.31; H, 9.92; N, 11.16.
4	C or D	81	¹ H NMR (CDCl ₃ ; δ, ppm): 0.84 (d, 12H, J = 5.7 Hz); 0.98-1.59 (m, 10H); 2.38-2.90 (m, 4H); 3.06-3.76 (m, 12H); 4.06 (s, 4H); 7.33-7.53 (m, 8H); 7.67-7.89 (m, 4H); 8.01-8.17 (m, 2H). ¹³ C NMR (CDCl ₃ ; δ, ppm): 12.76 (+); 22.16 (-); 24.61 (+); 30.81 (-); 32.23 (+); 32.37 (+); 50.24 (+); 50.37 (+); 52.43 (+); 123.92 (-); 125.51 (-); 125.74 (-); 126.30 (-); 126.43 (-); 127.65 (-); 128.75 (-); 131.85 (C _{quat}); 132.02 (C _{quat}); 133.88 (C _{quat}); 165.22 (C _{quat}). Mass spectrum: m/z = 439 [M - 269] ⁺ . Analysis (C ₄₂ H ₅₆ N ₄ S ₂ Si): Calcd %: C, 71.14; H, 7.96; N, 7.90. Found %: C, 70.97; H, 7.84; N, 7.81.
5	C or D	94	¹ H NMR (CDCl ₃ ; δ, ppm): 0.86 (t, 6H, J = 5.7 Hz); 1.14-1.54 (m, 20H); 2.28-2.90 (m, 4H); 3.08-3.78 (m, 12H); 4.08 (s, 4H); 7.33-7.42 (m, 8H); 7.68-7.90 (m, 4H); 8.06-8.17 (m, 2H). ¹³ C NMR (CDCl ₃ ; δ, ppm): 14.15 (-); 14.73 (+); 16.40 (+); 22.62 (+); 23.43 (+); 31.41 (+); 31.50 (+); 32.41 (+); 50.26 (+); 50.38 (+); 52.50 (+); 122.44 (-); 124.68 (-); 124.99 (-); 126.31 (-); 126.43 (-); 127.76 (-); 128.84 (-); 131.91 (C _{quat}); 132.06 (C _{quat}); 133.88 (C _{quat}); 165.19 (C _{quat}). Mass spectrum: m/z = 651 [M - 85] ⁺ . Analysis (C ₄₄ H ₆₀ N ₄ S ₂ Si) Calcd %: C, 71.69; H, 8.20; N, 6.60. Found %: C, 71.62; H, 8.13; N, 6.67.
6	C or D	98	¹ H NMR (CDCl ₃ ; δ, ppm): 0.88 (d, 12H, J = 5.4 Hz); 1.09-1.42 (m, 10H); 2.34-2.74 (m, 4H); 3.07-3.76 (m, 12H); 4.05 (s, 4H); 7.31-7.56 (m, 8H); 7.69-7.88 (m, 4H); 8.07-8.18 (m, 2H). ¹³ C NMR (CDCl ₃ ; δ, ppm): 17.19 (+); 18.28 (+); 22.08 (+); 30.50 (-); 32.22 (+); 33.41 (+); 50.22 (+); 50.40 (+); 52.59 (+); 123.71 (-); 125.51 (-); 125.78 (-); 126.51 (-); 126.59 (-); 127.66 (-); 128.77 (-); 132.05 (C _{quat}); 132.43 (C _{quat}); 133.88 (C _{quat}); 165.07 (C _{quat}). Mass spectrum: m/z = 485 [M - 269] ⁺ . Analysis (C ₄₂ H ₅₆ N ₄ S ₂ Ge) Calcd %: C, 66.93; H, 7.49; N, 7.43. Found %: C, 66.78; H, 7.32; N, 7.45.

7	C or D	98	¹ H NMR (CDCl ₃ ; δ, ppm): 0.87 (t, 6H, J = 5.6 Hz); 1.09-1.54 (m, 20H); 2.35-2.74 (m, 4H); 3.03-3.77 (m, 12H); 4.05 (s, 4H); 7.19-7.49 (m, 8H); 7.67-7.92 (m, 4H); 8.16-8.33 (m, 2H). ¹³ C NMR (CDCl ₃ ; δ, ppm): 14.11 (-); 19.85 (+); 22.54 (+); 23.36 (+); 24.64 (+); 31.15 (+); 31.57 (+); 32.35 (+); 50.25 (+); 50.44 (+); 52.71 (+); 123.72 (-); 125.52 (-); 125.80 (-); 126.35 (-); 126.56 (-); 127.62 (-); 128.75 (-); 132.10 (C _{quat}); 132.29 (C _{quat}); 133.89 (C _{quat}); 165.03 (C _{quat}). Mass spectrum: m/z = 697 [M - 85] ⁺ . Analysis (C ₄₄ H ₆₀ N ₄ S ₂ Ge) Calcd %: C, 67.61; H, 7.74; N, 7.17. Found %: C, 67.35; H, 7.66; N, 7.54.
8		74	¹ H NMR (CDCl ₃ ; δ, ppm): 2.45 (s, 1H); 2.56-2.85 (m, 2H); 3.07-3.93 (m, 6H); 4.04 (s, 2H); 7.34-7.54 (m, 4H); 7.66-7.89 (m, 2H); 8.04-8.19 (m, 1H). ¹³ C NMR (CDCl ₃ ; δ, ppm): 23.44 (+); 32.31 (+); 50.15 (+); 50.31 (+); 52.60 (+); 123.68 (-); 125.54 (-); 125.85 (-); 126.45 (-); 126.52 (-); 127.77 (-); 128.86 (-); 132.02 (C _{quat}); 132.39 (C _{quat}); 133.88 (C _{quat}); 165.25 (C _{quat}). I.R. (cm ⁻¹): ν _{SH} = 2538. Mass spectrum: m/z = 269 [M - 1] ⁺ . Analysis (C ₁₆ H ₁₈ N ₂ S) Calcd %: C, 71.07; H, 6.71; N, 10.36. Found %: C, 70.86; H, 6.77; N, 9.99.

[1-Thioethyl-2-(1-naphthylmethyl)]-2-imidazoline 8

A solution of 2-(1-naphthylmethyl)-2-imidazoline[#] (5.29 g, 25.16 mmol) in 60 ml of dry toluene was mixed with a solution of ethylene sulfide (1.59 g, 26.45 mmol) (Aldrich-Chemical) in 40 ml of dry toluene (sealed tube, argon flushed). The reaction mixture was then heated (110°C oven) for 15 h. After cooling 100 ml of cold diethyl ether was added with stirring to reaction mixture, and filtration to remove small amount of polyethylene sulfide. The solvent was removed under reduced pressure to give a yellow pasty product **8** (4.78 g, 74 %).

[#]To a solution of 2-(1-naphthylmethyl)-2-imidazoline hydrochloride (Aldrich-Chemical) (30g, 121.58 mmol) in 30 ml of water was added with stirring a solution of NaOH 14 % (34.73 g). After extraction with toluene (500 ml), the organic layer dried on Na₂SO₄. Removal of solvent *in vacuo* and crystallization from THF/diethyl ether (1/9, 400 ml) gave 2-(1-naphthylmethyl)-2-imidazoline in form of white crystals (21.31 g, 87 %).

Physicochemical data of derivatives **1-8** are reported in Table I

Table II - Radioprotective activity of 1-8 compounds

Compound	LD ₅₀ : mg.kg ⁻¹ (mmol)	Injected dose mg.kg ⁻¹	Irradiation Gy (t, min) ^a	Survival rate %	DRF ^b
1	> 1500 (4.35)	1000	8 (15)	70	1.2
		1000	8 (90)	30	
2	> 800 (1.72)	600	7.5 (15)	20	-
3	> 1500 (3.04)	750	8 (15)	60	-
4	~100 (0.14)	50	7.75 (15)	100	1.4
		50	7.75 (90)	80	
		12.5	7.75 (15)	60	
		50	9.75 (15)	10	
5	~100 (0.13)	50	7.75 (15)	100	1.5
		50	7.75 (90)	100	
		12.5	7.75 (15)	50	
6	~80 (0.11)	50	7.75 (15)	100	1.7
		50	7.75 (90)	100	
		50	7.75 (180)	90	
		12.5	7.75 (15)	80	
		50	9.75 (15)	70	
		50	9.75 (90)	0	
7	~150 (0.19)	75	7.75 (15)	90	1.4
		75	7.75 (90)	100	
		18.75	7.75 (15)	70	
		75	9.75 (15)	10	
8	35 (0.13)	17.5	8 (15)	80	1.1
		4.38	8 (15)	47	

a: t = time between administration of compound and irradiation.

b: dose reduction factor = (LD_{50(30 days)} treated/LD_{50(30 days)} untreated).

Pharmacology: evaluation of radioprotection

Male CD1 mice (Charles River, France), 25 g body weight, were used. Compounds were injected intraperitoneally 15, 90 or 180 min before irradiation. The irradiation dose was LD_{100/30 days} for untreated mice (7.5, 7.75 or 8 Gy, according to the irradiation date) or a 2

Gy greater dose. The injected dose of compound was equal to, three-quarter, two-third, one-half or one eighth of the LD₅₀ value which had been determined previously. The radioprotective effect was evaluated by the Dose Reduction Factor (DRF), which is the ratio between the LD₅₀/30 days of treated mice and that of control mice (between 6.5 and 6.75 Gy, according to the date).

Irradiation was applied using a cobalt-60 source at the dose rate of 0.3-0.4 Gy.min⁻¹ according to the date. During irradiation, animals were placed in a Plexiglass box with 30 cells in a homogeneous field, 28.5 x 28.5 cm in area. Dosimetry was checked with an ionisation chamber dosimeter. The different LD₅₀ values were determined by probit analysis.

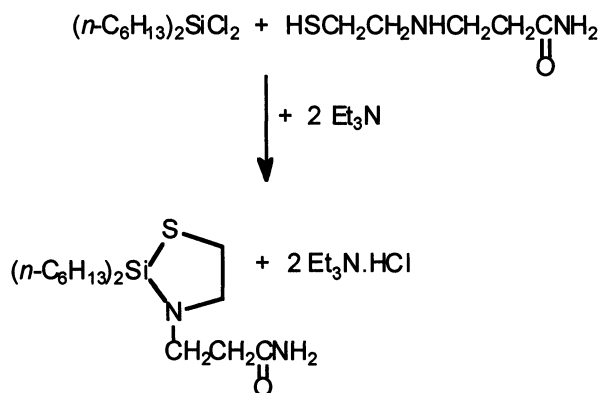
Results and discussion

Silathiazolidine

Silathiazolidine has been prepared according to two methods of heterocyclisation already described in the literature [1, 12, 13].

Method A

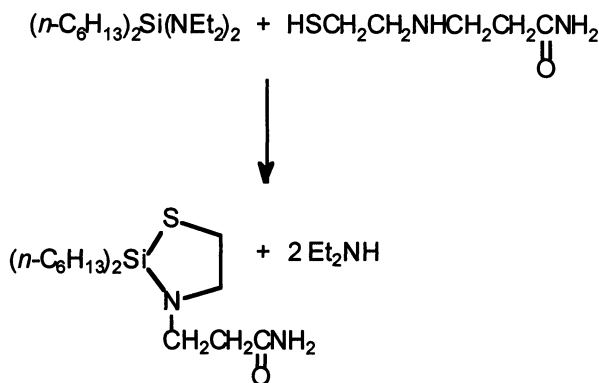
The action of di-*n*-hexyldichlorosilane, in stoichiometric amounts, on 3-[N-(2-thioethyl)]aminopropanamide in refluxing anhydrous THF in the presence of freshly distilled triethylamine gave by a cyclisation reaction, with elimination of hydrochloric acid from Si-Cl and NH groups [13], the corresponding product, Scheme 1:



Scheme 1

Method B

Treatment of bis(diethylamino)-di-*n*-hexylsilane, in stoichiometric amounts, with 3-[N-(2-thioethyl)]aminopropanamide in anhydrous THF resulted in the cleavage of Si-N bonds by the N-H (a transamination reaction) and S-H groups [1,13,15] forming the corresponding silathiazolidine, Scheme 2:



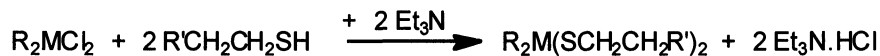
Scheme 2

Sila- and germadithioacetals

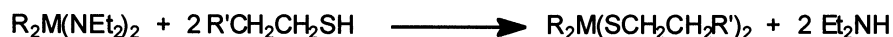
These compounds were also synthesized by two methods, C and D.

Method C

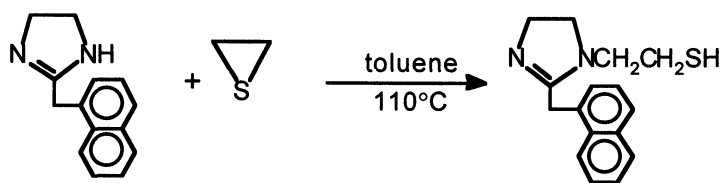
The reaction of dialkyldichlorosilanes and -germanes with two equivalents of 3-[N-(2-thio-ethyl)]aminopropanamide or [1-thioethyl-2-(1-naphtylmethyl)]-2-imidazoline in the presence of tri-ethylamine in refluxing anhydrous THF leads to the acyclic derivatives, Scheme 3:

**Scheme 3****Method D**

The action of two equivalents of 3-[N-(2-thioethyl)]aminopropanamide or [1-thioethyl-2-(1-naphtylmethyl)]-2-imidazoline with bis(diethylamino)dialkylsilanes and -germanes in anhydrous THF, a cleavage reaction of M-N bonds by the N-H (a transamination reaction) and S-H groups [1, 13, 15] leads to the formation of the desired products, Scheme 4:

**Scheme 4****Synthesis of [1-thioethyl-2-(1-naphtylmethyl)]-2-imidazoline**

This compound was obtained by the reaction of 2-(1-naphtylmethyl)-2-imidazoline with ethylene sulfide in a sealed tube at 110°C in anhydrous toluene (i.e. by a cleavage of the C-S bond by the N-H group) [16], Scheme 5:

**Scheme 5****Conclusions**

The experimental evaluation of toxicity and radioprotective activity of metalladithioacetals and silathiazolidine 1-7 in the mice is presented in Table II.

Compounds 1 and 3 showed a lower toxicity ($LD_{50} > 1500 \text{ mg.kg}^{-1}$) compared with the starting organic derivative $HSCH_2CH_2NHCH_2CH_2C(O)NH_2$ (WR-2529) $LD_{50} = 700 \text{ mg.kg}^{-1}$ [10].

With the other sila- and germadithioacetals, derivatives 4-7, we have observed a low decrease of the toxicity but a good radioprotective activity by intraperitoneal administration in mice. For example:

- derivative 4: at $LD_{50}/2 \sim 50 \text{ mg.kg}^{-1}$ this product protects 100 % and 80 % of mice when 4 was injected 15 and 90 minutes before irradiation.

- derivative 5: at $LD_{50}/2 \sim 50 \text{ mg.kg}^{-1}$ this product protects 100 % of mice when 5 was injected 15 and 90 minutes before irradiation. In the example shown, 50 % survival was also observed at $LD_{50}/8$.

Concerning the derivatives 6 and 7 we have noted a low toxicity and a notable increase of radioprotective activity:

- derivative 6: at $LD_{50}/1.6 \sim 50 \text{ mg.kg}^{-1}$ this product protects 100 % of mice when 6 was injected 15 and 90 minutes before irradiation, 90 % survival when 6 was injected 180 minutes before irradiation. At $LD_{50}/6.4$ there was still 80 % survival and at $LD_{50}/1.6$, 70 and 30 % survival at dose of 9.75 and 11.75 Gy.

- derivative 7: at $LD_{50}/2 \sim 75 \text{ mg.kg}^{-1}$ this product protects 90 and 100 % of mice when 7 was injected 15 and 90 minutes before irradiation. At $LD_{50}/8$, 70 % survival when 7 was injected 15 minutes before irradiation.

Chemical radioprotective study of silathiazolidine and metalladithioacetals showed a low toxicity and more potent protection than $HSCH_2CH_2NHCH_2CH_2C(O)NH_2$ (WR-2529) or derivative 8.

The results reported in this paper confirm the positive contribution of silicon and germanium in this field in agreement with previous works [1-10] and the interesting biological activity of organosilicon and organogermanium compounds in different fields [17-25].

Acknowledgments

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