Ring contraction versus β -elimination in reactions of alkynyl-substituted bicyclic lactol esters with SmI₂/Pd(0)

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Abstract

Selective addition of alkynyl metal reagents to either carbonyl group of 2-(2-formylethyl)cycloalkanones afforded alkynyl-substituted bicyclic lactols that were further converted into the corresponding acetate or benzoate esters. Reactions of these bicyclic esters with $SmI_2/Pd(PPh_3)_4$ displayed a divergent behavior which was dependent on the degree of substitution at the alkynyl terminus as well as on the bicycle ring size. Thus, 2-oxabicyclo[4.3.0]nonanes with terminal alkynes gave ring contracted bicyclic alcohols whereas the presence of substituents at the alkynyl terminus or the use of higher bicycloalkane homologues led to enol ethers, as the result of Lewis acid-promoted β -elimination.

Keywords: Lactol esters, samarium iodide, palladium(o), β-elimination

Introduction

We have recently reported on the intramolecular propargylation of carbonyl compounds using the umpolung of propargylic esters \mathbf{I} with $SmI_2/Pd(0)$. This reaction is thought to involve the initial formation of an allenylpalladium complex \mathbf{II} that is rapidly reduced to an equilibrium mixture of allenic (\mathbf{V}) and propargylic (\mathbf{VI}) organosamarium intermediates that finally add to the carbonyl group (Scheme 1).

Therefore, in this reaction the propargylic acetate acts as a synthetic equivalent of the propargyl anion synthon. As an extension of this chemistry, we have described the alternative use of acetals (III, $R^1 - R^3 = H$) and esters (IV, $R^1 - R^3 = H$), derived from structurally related lactols, to convert a carbohydrate to carbocycle via a ring contraction of the carbohydrate that presumably proceeds through the same intermediates. We aimed to apply this last reaction to bicyclic substrates 1,2 and 4. These would presumably afford the bicyclic alcohols 3 and 5 (Scheme 2) and this paper reports on the results of that study.

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Results and Discussion

Synthesis of starting materials. The representative substrates **1,2** and **4** were prepared using ketoaldehydes **6** as common starting materials (Scheme 3, Table 1). The method initially selected involved *in situ* protection of the al dehyde carbonyl using tetrakis(diethylamino)titanium according to the procedure developed by Reetz, followed by addition to the ketone to give lactols **7**. This procedure worked well for additions to cyclohexanone **6a** (Table 1, entries 1–3) but failed when cyclopentanone **6b** was used.

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$$R1$$
 O
 OR_2
 SmI_2
 $Pd(0)$
 $R3$
 $R3$
 $R1$
 OH
 $R3$

$$\begin{array}{c|c} COOR_2 & R1 \\ \hline O & \\ \hline SmI_2 \\ \hline Pd(0) & \\ \hline R3 & \\ \hline \end{array}$$

Table 1. Preparation of Acetates 1 from Ketoaldehydes 6

	6	R-≡-M	7 (%)	1 (%)	Cis/trans ^a Ratio
1	6a ^b	H-≡-MgBr	7a	$1a (71)^{c}$	Cis only
2	6a ^b	TMS-≡-Li	7b (84)	1b (69)	63:37
3	6a ^b	Ph-≡-Li	7c	$1c (54)^{c}$	79:21
4	$6b^d$	H-≡-Li	7d (44)	1d (87)	Trans only
5	6b ^e	n-Hex-≡-Li	7e (33)	1e (89)	f

^a *Cis/trans* refers to ring fusion. Ratio determined by ¹H NMR integration or from isolated weights of individual isomers. ^bProtection method: Ti(NEt₂)₄, -78 °C, then -52-(-43) °C. ^c Yield for two steps starting from 6a. ^d Protection method: dibenzylamine, benzotriazole, 4 Å molecular sieves, 25 °C, work-up. ^eProtection method: cyclohexylamine, 25 °C, work-up. ^fAn approximate 20:1 diastereoisomer ratio.

In this case the starting ketoaldehyde was recovered unchanged. Alternatively, protection of the aldehyde carbonyl of **6b** as either a *N*-(dibenzylaminoalkyl)benzotriazole⁹ or *N*-cyclohexyl imine followed by addition of appropriate alkynyl lithium or magnesium derivatives to the remaining ketone carbonyl afforded moderate yields of lactols **7d,e** (Table 1, entries 4,5). Lactols **7** were obtained with moderate to high diastereoselectivity. Acetylation of the lactols using standard conditions afforded in all cases good yields of acetates **1**. Alternatively, treatment of lactol **7a** with HCl/MeOH led to the corresponding acetal **2** as a single diastereoisomer (Scheme 3).

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For the synthesis of **4**, the ketoaldehyde **6b** was directly treated with the required alkynyl lithium and the resulting alkoxide intermediate was trapped with benzoyl chloride to afford moderate yields of benzoates **4**. Two out of four isomers were observed at most for **1**, **2** and **4**. The stereochemistry of ring fusion has been determined for both isomers of **1c** and the assignments were extended by analogy to **1a**, **1b**, **1d** and **2**. Thus, the major isomer of **1c**, derived from cyclohexanone **6a**, was assigned a *cis*-ring fusion based on the observed coupling constants (*J*) for its H-4 proton in the ¹H NMR spectrum (Table 2).

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Table 2. Coupling constants *J* (in Hz) for H-4 in 1 and 2

n (1,2)	R	X	Major isomer	Minor isomer
2 (1a)	Н	Ac	m	_
2 (1b)	TMS	Ac	m	dd
				(10.2, 2.9)
2 (1c)	Ph	Ac	dd	dd
			(7.3, 5.4)a	(10.1, 3.1)
1 (1d)	Н	Ac	dd	
			(9.7, 2.7)	
1 (1e)	<i>n</i> -Bu	Ac	dd	_
			(9.8, 2.6)	
2 (2)	Н	Me	m	

These values were intermediate between those expected for axial and equatorial dispositions of that proton. This assignment was confirmed by the temperature dependence shown by the apparent J values of H-4 in that isomer (see Table 2). This indicated a flexible conformation only compatible with a *cis* ring junction. This analysis did not allow, however, an unambiguous configurational assignment for C-4. On the other hand, H-4 in minor **1c** appeared in the ¹H NMR spectrum as doublet of doublets with J = 10.1, 3.1 Hz. This signal remained unchanged over a temperature interval between -30 and 58 °C and this was taken as an indication of a rigid *trans* ring fusion with an axial disposition for H-4. According to this analysis, the major isomers of cyclopentanone-derived **1d,e** would present *trans* ring junctions as indicated by their H-4 J values (~ 10 and 3 Hz). No stereochemical assignment was made for **4a,b**.

Reactions with SmI₂/Pd(0). The major isomers of lactol esters **1** and **4** were independently treated with 2.2 equivalents of SmI₂ and a catalytic amount of Pd(PPh₃)₄ (5 mol%) at room temperature. The results were drastically dependent on the type of substrate employed and the presence or absence of substituents at the terminal alkynyl position (Scheme 4, Table 3).

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Thus, for cyclopentanone-derived substrates **1d** and **4a**, with a terminal alkyne, a slow reaction was observed that led to the formation of bicyclic alcohols **3d** and **5a**, respectively, with good yields and excellent stereoselectivities. However, the reaction took a completely different course when a terminal alkynyl substituent was present in the substrate, as in **1e** and **4b**. In those cases the only reaction products isolated were the enol ethers **8e** and **9b**, respectively. Similarly, the reactions of the cyclohexanone-derived **1a-c** led predominantly to enol ethers **8a-c**, accompanied occasionally by lactols **7**, probably the result of hydrolysis of **8** during work-up. The formation of enol ethers **8** does not require the presence of Pd(0) in the medium, as indicated by the result of entry 3 in Table 3. The acetal **2** was inert in the presence of SmI₂/Pd(PPh₃)₄ even under refluxing conditions.

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	1 ^a	R	Product	Yield (%)
1	1a	Н	$8a^b$	65
2	1b	TMS	8b	63
3	$1c^{c}$	Ph	8c	78
4	1d	Н	3d	84
5	1e	<i>n</i> -Bu	8e	88
6	4a	Н	5a	73
7	4 b	<i>n</i> -Bu	9 b	37^{d}

Table 3. Reactions of 1 and 4 with SmI₂/Pd(PPh₃)₄

The stereochemical assignment of 3d followed from the chemical shift differences found in the carbinolic protons of the diastereomeric mixture of alcohols 3d and 3d' obtained after NaBH₄ reduction of the ketone 10 that resulted from the PDC oxidation of 3d (Scheme 5). Thus, H-4 of 3d resonated at δ 4.37 whereas the corresponding proton in 3d', buried in the concave face of the bicyclic structure, was found further upfield at δ 3.8–3.9. The bicycle 5a obtained from 4a was identical in all physical properties to the previously reported material prepared using the route $I \rightarrow VII$ (Scheme 1).

Scheme 5

Results and Discussion

Carbohydrate-derived monocyclic acetals and esters related to 1, 2 and 4 have been employed in ring raction reactions leading to 2-alkynylcyclopentanols, 5,6 a transformation that is readily understood in the mechanistic terms shown in Scheme 1. This reaction is facilitated by Lewis acidic Sm(III) species, inevitably present in the reaction medium, that activate the leaving group (OMe or OCOR) and bring about the initial oxidative addition step that begins the catalytic cycle. Always latent in these reactions is the possibility of β -elimination promoted by the same

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^aMajor isomers. ^bAlso obtained was **7a** (6%). ^cPd(PPh₃)₄ was not used. ^dStarting material was recovered (32%).

Lewis acids or even by SmI_2 .¹⁰ In monocyclic systems with terminal alkynes the oxidative addition is reasonably fast and, as a result, ring opening is favored over β -elimination.^{5,6} This also seems to be the case with cyclopentanone-derived substrates 1d and 4a. If, on the other hand, the oxidative addition step becomes slower due to the presence of terminal alkynyl substituents, then β -elimination takes over as it is observed in the reactions of 1e and 4b. However, it is also apparent from the strikingly different results observed for 1a and 1d, that the size of the rings containing the bridgehead propargylic position, as well as the axial or equatorial orientation of the leaving group, are also probably important to the outcome of the reaction. Thus, β -elimination from the equatorial leaving group of the conformationally rigid bicycle 1d is expected to be slower than from the much more flexible bicycle 1a where both axial and equatorial orientations of the leaving group are possible at any given time. Additionally, the hybridization change in cyclopentanone-derived 1d in going to an intermediate related to 1d (Scheme 1d) should be more favorable than that taking place in cyclohexanone-derived 1a. Taken together, all these factors give as a result a preferred 1e-elimination pathway for 1a-c.

Conclusions

The SmI₂/Pd(0)-promoted ring contraction of bicyclic lactol esters leads to the expected bicyclic homopropargyl alcohols only when the oxidative addition that initiates the reaction competes efficiently with the alternative Lewis acid-promoted β -elimination. The synthetic potential of the ring contraction process for the preparation of bicyclic systems is significantly restricted by factors that adversely affect oxidative addition, namely (*i*) the presence of terminal alkynyl substituents and (*ii*) the development of torsional strain upon ring opening. At the same time, β -elimination is facilitated by axial leaving groups with an antiperiplanar β -hydrogen.

Experimental Section

General Procedures. All reactions involving air- and moisture-sensitive materials were performed using standard bench-top techniques. Diiodoethane was purified as reported. Tetrahydrofuran (THF) was freshly distilled from sodium/benzophenone and, for reactions with SmI₂, it was deoxygenated prior to use. Other solvents were routinely purified using literature procedures. Flash column chromatography was performed on silica gel (230–400 mesh). HPLC purifications were carried out with either a LiChrosorb Si60 (7 μm, 25 x 2.5 cm, column 1) or a μ Porasil (10 μm, 19 x 1.5 cm, column 2) column using a refractive index detector. Routine H and To NMR spectra were obtained at 250 MHz and 62.9 MHz, respectively, using CDCl₃ as solvent and internal reference (δ 7.26 for H and δ 77.0 for To Di R data include only characteristic absorptions. Mass spectra were obtained at 70 eV.

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Ethvl 4-acetoxy-6-ethynyl-cis-5-oxabicyclo[4.4.0]decanecarboxylate (1a). To a stirred solution of 6a¹⁵ (543 mg, 2.40 mmol) in THF (19 mL) at -78 °C was added dropwise Ti(NEt₂)₄ (880 uL, 2.43 mmol). The mixture was allowed to reach -43 °C over a period of 1 h, and ethynylmagnesium bromide (0.5 M in THF, 9.8 mL, 4.90 mmol) was added dropwise. The resulting solution was stirred at the same temperature for 14 h and guenched with 1 M HCl (19 mL). The aqueous layer was extracted with EtOAc (3 x 45 mL) and the combined organic extracts were washed with brine (6 mL) and dried (Na₂SO₄). The crude after evaporation of the solvent was dissolved in triethylamine (1.20 mL, 8.61 mmol) and treated with acetic anhydride (680 µL, 7.20 mmol) and then DMAP (90 mg, 0.73 mmol). The mixture was stirred at room temperature for 15 h, diluted with Et₂O (80 mL) and poured over ice-water (8 mL). The aqueous layer was extracted with Et₂O (3 x 20 mL) and the combined organic layers were successively washed with 1 M HCl (10 mL), 1 M NaOH (9 mL), H₂O (6 mL) and brine (6 mL). After drying (Na₂SO₄) and evaporation of the solvents the resulting residue was purified by flash chromatography (silica gel, 15% EtOAc in hexanes) to afford 1a (473 mg, 71%) as a colorless oil: ¹H NMR δ 1.25 (t, J = 7.1 Hz, 3H, CO₂CH₂CH₃), 1.38–2.24 (m, 13H), 2.10 (s, OCOCH₃, included in m at 1.38-2.24), 2.24-2.37 (m, 1H), 2.49 (td, J = 13.5, 4.8 Hz, 1H), 2.60 (s, 1H, H-2'), 4.16 (q, J = 7.1 Hz, 2H, CO₂CH₂CH₃), 6.21–6.26 (m, 1H, H-4); ¹³C NMR δ 14.0, 20.0, 21.2, 21.6, 28.2, 28.6, 35.4, 47.3, 60.5, 73.9, 75.6, 82.7, 92.6, 169.1, 173.4; IR (neat) v 3260, 2109, 1753, 1725 cm⁻¹; HRMS calcd for $C_{16}H_{22}O_5$ 294.1467, found 294.1478.

Alternatively, the crude obtained in the reaction with ethynylmagnesium bromide was purified by flash chromatography (silica gel, 15% EtOAc in hexanes) to yield **ethyl 4-hydroxy-6-ethynyl-***cis***-5-oxabicyclo**[**4.4.0**]**decanecarboxylate** (**7a**) (42%, one isomer). The analytical sample was obtained after additional purification by HPLC (column 1, 30% EtOAc in hexanes, 10 mL/min): t_R 20 min; ¹H NMR δ 1.22–1.95 (m, 12H), 1.25 (t, J = 7.1 Hz, CO₂CH₂CH₃, included in m at 1.22–1.95), 2.03–2.31 (m, 2H), 2.43–2.56 (m, 2H), 2.53 (s, H-2', included in m at 2.43-2.56), 3.33 (d, J = 5.7 Hz, 1H, OH), 4.15 (q, J = 7.1 Hz, 2H, CO₂CH₂CH₃), 5.35 (ddd, J = 9.7, 5.7, 3.0 Hz, 1H, H-4); ¹³C NMR δ 14.0, 20.1, 21.7, 27.9, 28.2, 28.9, 35.8, 47.3, 60.5, 73.0, 75.0, 83.7, 93.3; IR (KBr) υ 3370, 3240, 2085, 1735 cm⁻¹. Anal. Calcd for C₁₄H₂₀O₄: C, 66.63; H, 7.99. Found: C, 66.51; H, 7.87.

Ethyl 4-acetoxy-6-(trimethylsilylethynyl)-5-oxabicyclo[4.4.0]decanecarboxylate (1b). The procedure described for the synthesis of 1a was followed from $6a^{15}$ (521 mg, 2.30 mmol) and lithium (trimethylsilyl)acetylide (~ 0.27 M, 2.85 mmol). Reaction time : 1.2 h at -45 °C. The intermediate lactols were purified by flash chromatography (silica gel, 20% EtOAc in hexanes, then 30% EtOAc in hexanes) and then acetylated individually as described for 1a. The acetate derived from the major lactol was purified by flash chromatography (silica gel, 5% EtOAc in hexanes) to afford *cis*-1b (295 mg, 35% from 6a) as a colorless oil. The acetate derived from the minor lactol was purified by flash chromatography (silica gel, 8% EtOAc in hexanes, then 10% EtOAc in hexanes) and HPLC (column 1, 10% EtOAc in hexanes, 8 mL/min, t_R 32 min) to yield *trans*-1b (173 mg, 23% from 6a). Data for *cis*-1b: 1 H NMR δ 0.15 (s, 9H, Si(CH₃)₃), 1.26 (t, J = 7.1 Hz, 3H, CO₂CH₂CH₃), 1.50-2.13 (m, 14H), 2.09 (s, OCOCH₃, included in m at 1.50-2.13),

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2.24–2.51 (m, 2H), 4.14 (q, J = 7.1 Hz, 2H, CO₂CH₂CH₃), 6.20–6.26 (m, 1H, H-4); ¹³C NMR δ –0.2, 14.1, 20.0, 21.2, 21.5, 28.2, 28.7, 35.3, 47.3, 60.4, 74.3, 92.3, 92.6, 104.4, 169.1, 173.5; IR (neat) υ 2160, 1760, 1730 cm⁻¹. Anal. Calcd for C₁₉H₃₀O₅Si: C, 62.26; H, 8.26. Found: C, 62.32; H, 8.14. Data for *trans*-1b: ¹H NMR δ –0.21 (s, 9H, Si(CH₃)₃), 1.08–1.30 (m, 4H), 1.27 (t, J = 7.1 Hz, CO₂CH₂CH₃, included in m at 1.08–1.30), 1.37–2.17 (m, 13H), 2.08 (s, OCOCH₃, included in m at 1.37–2.17), 2.63–2.75 (m, 1H), 4.10-4.24 (m, 2H, CO₂CH₂CH₃), 6.40 (dd, J = 10.2, 2.9 Hz, 1H, H-4); ¹³C NMR δ –0.31, 14.1, 21.1, 22.0, 22.9, 27.9, 30.2, 33.0, 49.3), 60.1, 78.4, 92.4, 94.9, 103.9, 169.0, 172.3; IR (KBr) υ 2150, 1750 cm⁻¹. Anal. Calcd for C₁₉H₃₀O₅Si: C, 62.26; H, 8.26. Found: C, 62.07; H, 8.28.

Ethyl 4-acetoxy-6-(phenylethynyl)-5-oxabicyclo[4.4.0]decanecarboxylate (1c). Prepared from 6a¹⁵ (955 mg, 4.22 mmol) and lithium phenylacetylide (~ 0.35 M, 5.08 mmol) following the procedure described for the synthesis of 1a. The addition of the lithium reagent was done at -78°C and the reaction was allowed to proceed at -52 °C for 2.2 h. The crude product was purified by flash chromatography (silica gel, 15%, 25% and finally 35% EtOAc in hexanes) to afford cis-1c (546 mg) and trans-1c (140 mg, after additional purification by HPLC). Dat a for cis-1c: ¹H NMR δ 1.27 (t, J = 7.1 Hz, 3H, CO₂CH₂CH₃), 1.53–2.22 (m, 13H), 2.12 (s, OCOCH₃, included in m at 1.53–2.22), 2.33–2.62 (m, 2H), 4.18 (q, J = 7.1 Hz, 2H, $CO_2CH_2CH_3$), 6.35 (dd, J = 7.3, 5.4 Hz, 1H, H-4), 7.27–7.29 (m, 3H, Ar-H), 7.41–7.45 (m, 2H, Ar-H); ¹³C NMR δ 14.0, 2 0.0, 21.0, 21.5, 25.3, 28.2, 28.8, 35.3, 47.5, 60.3, 74.4, 87.3, 88.2, 92.5, 122.4, 128.0, 128.1, 131.5, 168.9, 173.4; IR (neat) ν 2225, 1753, 1725 cm⁻¹; HRMS calcd for $C_{22}H_{26}O_5$ 370.1780, found 370.1778. Data for *trans-1c*: HPLC (column 1, 25% EtOAc in hexanes, 7 mL/min) t_R 24 min; ¹H NMR δ 1.19–1.31 (m, 4H), 1.28 (t, J = 7.1 Hz, $CO_2CH_2CH_3$, included in m at 1.19–1.31), 1.44-2.13 (m, 8H), 2.05-2.10 (m, 5H), 2.09 (s, OCOC H_3 , included in m at 2.05-2.10), 2.16 (td, J = 13.6, 9.8 Hz, included in m at 2.05–2.10), 2.72–2.85 (m, 1H), 4.22 (q, J = 7.1 Hz, 2H, $CO_2CH_2CH_3$), 6.48 (dd, J = 10.1, 3.1 Hz, 1H, H-4), 7.30–7.35 (m, 3H, Ar-H), 7.44–7.50 (m, 2H, Ar-H); ¹³C NMR δ 14.2, 21.2, 22.2, 23.1, 28.1, 30.5, 33.3, 33.3, 49.8, 60.3, 78.7, 87.7, 89.9, 92. 122.3, 128.3, 128.5, 131.6, 169.3, 172.5; IR (neat) ν 2225, 1753, 1725 cm⁻¹; HRMS calcd for C₂₂H₂₆O₅ 370.1780, found 370.1775.

Ethyl 4-methoxy-*cis*-**5-oxabicyclo**[**4.4.0**]**decanecarboxylate** (**2**). A solution of **7a** (273 mg, 1.08 mmol) in MeOH/HCl (98:2, 4.3 mL) was refluxed for 3 h. Saturated K₂CO₃ (5 mL) was added until basic pH and MeOH was evaporated at reduced pressure. The aqueous layer was extracted with EtOAc (100, 25 and 25 mL) and the combined organic layers were washed with brine (10 mL) and dried (Na₂SO₄). The crude after evaporation of the solvent was purifieby flash chromatography (silica gel, 7% EtOAc in hexanes) to yield **2** (230 mg, 79%, one isomer): 1 H NMR δ 1.21 (t, J = 7.1 Hz, 3H, CO₂CH₂CH₃), 1.28–2.24 (m, 11H), 2.44 (td, J = 13.3, 4.5 Hz, 1H), 2.50 (s, 1H, H-2'), 3.42 (s, 3H, OCH₃), 4.10 (q, J = 7.1 Hz, 2H, CO₂CH₂CH₃), 4.85–4.90 (m, 1H, H-4); 13 C NMR δ 14.0, 20.0, 21.6, 26.2, 28.1, 28.8, 35.5, 47.4, 55.9, 60.3, 72.5, 74.6, 83.9, 99.9, 173.8; IR (neat) ν 3255, 2101, 1733 cm⁻¹. Anal. Calcd for C₁₅H₂₂O₄: C, 67.63; H, 8.33. Found: C, 67.72; H, 8.40.

Ethyl 6-ethynyl-4-hydroxy-5-oxabicyclo[4.3.0]nonanecarboxylate (7d). A mixture of 6b¹⁶

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(0.223 g, 1.07 mmol), benzotriazole (0.127 g, 1.07 mmol), dibenzylamine (0.20 mL, 1.07 mmol) and 4 Å molecular sieves (4 g) in benzene (30 mL) was stirred for 12 h at 25 °C and filtered through Celite. The solvent was removed at reduced pressure and the residue was dissolved in THF (30 mL). To the well stirred solution at -78 °C was added *via* cannula lithium acetylide ¹⁷ (0.042 M, 1.26 mmol) and the mixture was stirred at -78 °C for 30 min. The solution was allowed to reach room temperature and stirred further 3 h. After adding 3 M HCl (15 mL) the layers were separated, the organic layer was washed with brine (15 mL) and dried (Na₂SO₄). The crude after evaporation was purified by flash chromatography (silica gel, 35% EtOAc in hexanes) to yield 0.123 g (44%) of lactol **7d** (~ 9:1 diastereomeric mixture) as an oil: ¹H NMR δ 1.26 (t, J = 7.1 Hz, 3H, CH_3 , major isomer, overlapped with signals from the minor isomer), 1.47–1.64 (m, 1H), 1.80–2.14 (m, 7H), 2.29–2.55 (m, 3H), 2.53 (s, acetylenic, major diast., included in m at 2.29–2.55), 2.98 (br s, $W_{I/2} = 10.2$ Hz, 1H, OH), 4.09–4.21 (m, 2H, OCH_2), 5.13 (d, J = 9.5 Hz, H-4, major diast.) and 5.23–5.25 (m, H-4, minor diast.) (total 1H); ¹³C NMR (major diast.) δ 13.9, 21.5, 24.7, 27.3, 30.6, 40.6, 55.9, 60.8, 75.3, 80.9, 81.0, 92.8, 174.4; IR (neat) 3430, 3260, 2100, 1730, 1720 cm⁻¹.

Ethyl 6-(hex-1-ynyl)-4-hydroxy-5-oxabicyclo[4.3.0]nonanecarboxylate (7e). To a stirred solution of hex-1-vnvl lithium (0.11 M, 11.39 mmol)¹ in THF (100 mL) at -78 °C under Ar was 3-(N-cyclohexylimino)propyl-2added via cannula solution of ethyl oxocyclopentanecarboxylate [from cyclohexylamine (1.20 mL, 10.85 mmol) and 6b¹⁶ (2.30 g, 10.85 mmol)] in THF (300 mL). The mixture was stirred for 2 h and then was allowed to reach room temperature. The solution was extracted successively with 1 M HCl (15 mL), 1 M NaOH (5 mL) and brine (10 mL). The aqueous layers were back extracted with ether (3 x 20 mL) and the combined organic extracts were dried (Na₂SO₄). The crude after evaporation was purified by flash chromatography (silica gel, 20% EtOAc in hexanes) to yield 1.05 g (33%, 17:1 diastereomeric mixture) of lactol 7e as an oil: ¹H NMR δ 0.88 (t, J = 7.1 Hz, 3H, CH_3), 1.24 (t, J= 7.2 Hz, 3H, OCH_2CH_3 , major isomer, overlapped with signals from the minor isomer), 1.32–1.60 (m, 4H), 1.77–2.09 (m, 8H), 2.17 (t, J = 6.8 Hz, 2H, CH_2 -C), 2.27–2.48 (m, 2H), 3.33 (d, J = 6.0 Hz, 1H, OH), 4.01-4.18 (m, 2H, OC H_2 CH₃), 5.10 (ddd, J = 9.6, 6.0, 1.9 Hz, 1H, H-4); 13 C NMR (major diast.) δ 13.5, 14.0, 18.3, 21.7, 21.8, 24.8, 27.7, 30.6, 40.9, 56.1, 60.6, 77.1, 81.6, 88.0, 92.8, 174.7; IR (neat) 3420, 2230, 1730 cm⁻¹; HRMS calcd for C₁₇H₂₆O₄ 277.1804 (M-OH), found 277.1804.

Ethyl 4-acetoxy-6-ethynyl-5-oxabicyclo[4.3.0]nonanecarboxylate (**1d**). The acetylation procedure described for the preparation of **1a** was applied to **7d** (0.260 g, 1.09 mmol). The crude product was purified by flash chromatography (silica gel, 15% EtOAc in hexanes) to yield 0.265 g (87 %) of acetate **1d** as an oil: 1 NMR δ 1.20 (t, J = 7.1 Hz, 3H, OCH₂CH₃), 1.54–2.12 (m, 11H), 2.03 (s, CH₃CO, included in m at 1.54–2.12), 2.28-2.49 (m, 2H), 2.56 (s, 1H, acetylenic), 4.09 (q, J = 7.1 Hz, 2H, OCH₂CH₃), 5.97 (dd, J = 9.7, 2.7 Hz, 1H, H-4); 13 C NMR δ 13.8, 20.9, 21.4, 24.3, 24.9, 30.6, 40.2, 55.8, 60.7, 75.9, 79.9, 81.7, 91.7, 168.8, 173.8; IR (neat) 3260, 2100, 1750, 1725 cm⁻¹; HRMS calcd for C₁₅H₂₀O₅: 280.1311, found 280.1292.

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Ethyl 4-Acetoxy-6-(hex-1-ynyl)-5-oxabicyclo[4.3.0]nonanecarboxylate (**1e**). The acetylation procedure described for the preparation of **1a** was applied to **7e** (0.800 g, 2.72 mmol). The crude product was purified by flash chromatography (silica gel, 10% EtOAc in hexanes) to yield 0.814 g (89 %, ~ 20:1 diastereomeric mixture) of acetate **1e** as an oil: ¹H NMR δ 0.84–0.97 (m, 3H, C H_3), 1.26 (t, J = 7.1 Hz, 3H, OCH₂C H_3 , major isomer, overlapped with signals of the minor isomer), 1.34–2.12 (m, 15H), 2.04 (s, C H_3 CO, minor diast., included in m at 1.34–2.12), 2.09 (s, C H_3 CO₂-, major diast., included in m at 1.34–2.12), 2.21 (t, J = 6.9 Hz, 2H, C H_2 -C), 2.34–2.50 (m, 2H), 4.07-4.19 (m, 2H, OC H_2 CH₃), 6.03 (dd, J = 9.8, 2.6 Hz, 1H, H-4); ¹³C NMR (major diast.) δ 13.4, 14.0, 18.2, 21.1, 21.6, 21.7, 24.5, 25.2, 30.5, 30.7, 40.5, 56.1, 60.6, 76.1, 82.6, 88.6, 92.1, 169.0, 174.3; IR (neat) 2230, 1730 cm⁻¹; HRMS calcd for C₁₉H₂₈O₅ 336.1937, found 336.1942.

Ethyl 6-benzoyloxy-4-ethynyl-5-oxabicyclo[4.3.0]nonanecarboxylate (**4a**). To a solution of **6b**¹⁶ (2.10 g, 10.2 mmol) in THF (40 mL) at -78 °C was added *via* cannula over 1 h a solution of lithium acetylide¹⁷ (0.21 M, 10.7 mmol) in THF (50 mL). The solution was stirred at -78 °C for 90 min and benzoyl chloride (2.4 mL, 20.45 mmol) was added neat. After allowing the mixture to reach room temperature, H₂O (20 mL) was added, the aqueous layer was extracted with ether (2 x 20 mL) and the combined organic layers were dried (Na₂SO₄). The crude was purified by flash chromatography (silica gel, 20% EtOAc in hexanes) to yield 1.20 g (35%) of **4a** as a colorless solid: mp 163–165 °C; ¹H NMR δ 1.21 (t, J = 7.1 Hz, 3H, OCH₂CH₃), 1.87–2.09 (m, 7H), 2.43 (d, J = 2.2 Hz, 1H, acetylenic), 2.47–2.61 (m, 2H), 2.81–2.90 (m, 1H), 4.06–4.25 (m, 2H, OCH₂CH₃), 4.40 (dt, J = 8.3, 2.2 Hz, 1H, H-4), 7.40–7.59 (m, 3H), 7.98–8.02 (m, 2H); ¹³C NMR δ 14.1, 20.4, 24.9, 26.8, 29.0, 36.5, 53.4, 61.0, 62.0, 73.3, 82.1, 109.3, 128.4, 129.8, 130.2, 133.2, 163.3, 174.1; IR (neat) 3300, 1740 cm⁻¹. Anal. Calcd for C₂₀H₂₂O₅: C, 70.16; H, 6.48. Found: C, 70.36; H, 6.62.

Ethyl 6-benzoyloxy-4-(hex-1-ynyl)-5-oxabicyclo[4.3.0]nonanecarboxylate (**4b**). The previous procedure was applied to **6b**¹⁶ (2.30 g, 10.85 m mol) and hex-1-ynyl lithium¹ (11.39 mmol). The crude was purified by flash chromatography (silica gel, 10% EtOAc in hexanes) to yield 1.31 g (30%, 17:1 diastereomeric mixture) of benzoate **4b** as an oil: ¹H NMR δ 0.87 (t, J = 6.8 Hz, 3H, CH_3), 1.08 (t, J = 7.1 Hz, OCH₂CH₃, minor diast.) and 1.17–1.51 (m) (total 7H), 1.20 (t, J = 7.1 Hz, OCH₂CH₃, major diast., included in m at 1.17–1.51), 1.70–2.04 (m, 7H), 2.17 (t, J = 6.9 Hz, 2H, CH_2 -C_), 2.43-2.60 (m, 2H), 2.71–2.87 (m, 1H), 4.05–4.24 (m, 2H, OCH₂CH₃), 4.37–4.42 (m, 1H, H-4), 7.35–7.64 (m, 3H), 8.00 (d, J = 8.0 Hz, major diast.) and 8.13 (d, J = 8.0 Hz, minor diast.) (total 2H); ¹³C NMR (major diast.) δ 13.2, 13.8, 18.2, 20.1, 21.6, 24.7, 27.1, 28.6, 30.1, 36.1, 53.0, 60.6, 62.2, 78.1, 85.4, 109.0, 128.1, 129.4, 130.0, 132.8, 163.0, 173.9; IR (neat) 2230, 1730 cm⁻¹; HRMS calcd for $C_{24}H_{30}O_5$ 398.2093, found 398.2068.

General procedure for reactions of 1, 2, 4 with SmI₂/Pd(PPh₃)₄.

In a typical experiment, a solution of the substrate (0.50 mmol) and $Pd(PPh_3)_4$ (0.025 mmol) in THF (4 mL) was added to a solution of $SmI_2^{2b}(1.1 \text{ mmol})$ in THF (11 mL) at room temperature. The resulting mixture was stirred until total conversion of the substrate (as determined by TLC)

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and saturated K_2CO_3 (5 mL) was added. The aqueous layer was extracted with EtOAc (4 x 35 mL), the combined organic layers were washed with brine (3 mL) and dried (Na_2SO_4). The crude product obtained after evaporation was purified by flash chromatography as specified for the individual cases.

Ethyl 6-ethynyl-cis-5-oxabicyclo[4.4.0]dec-3-enecarboxylate (8a). The general procedure was followed from 1a (136 mg, 0.49 mmol). Reaction time: 62 h. The crude product was purified by flash chromatography (silica gel, 5% EtOAc in hexanes, then 40% EtOAc in hexanes) to afford **8a** (75 mg, 65%) and **7a** (7 mg, 6%). Data for **8a**: ¹H NMR δ 1.24–1.33 (m, 4H), 1.26 (t, J = 7.1Hz, CO₂CH₂CH₃ included in m at 1.24–1.33), 1.52–1.64 (m, 4H), 1.72-1.77 (m, 2H), 1.94–2.03 (m, 2H), 2.42 (s, 1H, H-2'), 2.42–2.62 (m, 2H), 4.17 (q, J = 7.1 Hz, CO₂CHCH₃) and 4.18 (q, J= 7.1 Hz, CO_2CHCH_3) (total 2H), 4.80 (dt, J = 5.7, 2.3 Hz, 1H, H-3), 6.26 (dt, J = 5.7, 1.7 Hz, 1H, H-4); ¹³C NMRδ 14.1, 20.4, 21.9, 29.3, 29.7, 34.1, 47.1, 60.5, 71.5, 72.3, 84.5, 99.1, 140.1, 173.5; IR (neat) v 3290, 1740, 1670 cm⁻¹; HRMS calcd for $C_{14}H_{18}O_3$ 234.1256, found 234.1257. 6-(trimethylsilylethynyl)-cis-5-oxabicyclo[4.4.0]dec-3-enecarboxylate general procedure was followed from 1b(110 mg, 0.30 mmol) for 16 h. The crude product was purified by flash chromatography (silica gel, 2.5% EtOAc in hexanes) to yield **8b** (58 mg, 63%) as an oil: ¹H NMR δ 0.12 (s, 9H, Si(CH₃)₃), 1.25 (t, J = 7.1 Hz, 3H, CO₂CH₂CH₃), 1.34–1.73 (m, 6H), 1.90-2.00 (m, 2H), 2.40-2.56 (m, 2H), 4.15 (q, J = 7.1 Hz, 2H, $CO_2CH_2CH_3$), 4.69–4.75 (m, 1H, H-3), 6.21 (apparent d, J = 6.0 Hz, 1H, H-4); ¹³C NMR δ -0.1, 14.1, 20.5, 21.7, 29.0, 29.9, 34.0, 47.2, 60.4, 72.6, 88.2, 98.7, 106.1, 140.0, 173.5; IR (neat) v 3065, 2169, 1736, 1663 cm $^{-1}$; HRMS calcd for $C_{17}H_{26}O_3Si$ 306.1651, found 306.1646.

Ethyl 6-(phenylethynyl)-*cis*-**5-oxabicyclo[4.4.0]dec-3-enecarboxylate** (**8c**). A solution of **1c** (115 mg, 0.38 mmol) in THF (4 mL) was added to a solution of SmI₂ (0.1 M in THF, 9.5 mL, 0.95 mmol) at room temperature. The mixture was stirred for 22 h at the same temperature and was elaborated as specified in the general procedure. The crude product was purified by flash chromatograph(silica gel, 8% EtOAc in hexanes) to yield **8c** (72 mg, 78%) as an oil: ¹H NMR δ 1.24–2.00 (m, 7H), 1.26 (t, J = 7.1 Hz, CO₂CH₂CH₃, included in m at 1.24–2.00), 2.05–2.11 (m, 2H), 2.47–2.69 (m, 2H), 4.19 (q, J = 7.0 Hz, CO₂CHCH₃) and 4.20 (q, J = 7.1 Hz, CO₂CHCH₃) (total 2H), 4.79 (td, J = 5.9, 2.2 Hz, 1H, H-3), 6.29 (d, J = 5.9 Hz, 1H, H-4), 7.26–7.27 (m, 3H, Ar-*H*), 7.38–7.42 (m, 2H, Ar-*H*); ¹³C NMR δ 14.1, 20.6, 21.8, 29.3, 29.9, 47.5, 60.4, 72.8, 83.6, 90.0, 98.9, 122.9, 128.0, 131.7, 140.2, 173.5; IR (neat) v = 3063, 2230, 1733, 1662 cm⁻¹; HRMS calcd for C₂₀H₂₂O₃ 310.1569, found 310.1574.

(1*R**,4*R**,5*R**)-Ethyl 5-ethynyl-4-hydroxybicyclo[3.3.0]octanecarboxylate (3d). Prepared from acetate 1d (0.714 mmol) and Pd(PPh₃)₄ (0.036 mmol) in THF (8 mL) using the general procedure. Reaction time: 64 h. The crude was purified by flash chromatography (silica gel, 35% EtOAc in hexanes) to yield the alcohol 3d (0.134 g, 84%) as a colorless oil. The spectral sample was obtained after HPLC (column 2, 15% EtOAc in hexanes, 6 mL/min): t_R 15 min; ¹H NMR δ 1.26 (t, J = 7.1 Hz, 3H, OCH₂CH₃), 1.41–2.20 (m, 9H), 2.18 (s, acetylenic, included in m at 1.41–2.20), 2.26–2.55 (m, 3H), 4.12 (q, J = 7.1 Hz, 2H, OCH₂CH₃), 4.37 (dd, J = 10.3, 6.3 Hz,

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1H, H-4); 13 C NMR δ 14.1, 26.2, 31.6, 32.0, 35.3, 38.3, 56.4, 60.8, 63.9, 70.3, 80.2, 88.5, 175.4; IR (neat) 3450, 2100, 1725 cm $^{-1}$; HRMS calcd for $C_{13}H_{19}O_3$ (M+1) 223.1334, found 223.1357.

Ethyl 5-ethynyl-4-oxobicyclo[3.3.0]octanecarboxylate (**10**). To a solution of alcohol **1d** (0.053 g, 0.238 mmol) in CH₂Cl₂ (6 mL) was added PDC (0.30 g, 0.79 mmol) and the mixture was stirred at room temperature for 60 h, diluted with ether (25 mL), filtered through silica gel (70–230 mesh) and evaporated to dryness. The crude after evaporation was purified by flash chromatography (silica gel, 20% EtOAc in hexanes) to yield the ketone **10** (0.049 g, 94%) as a colorless oil: ¹H NMR δ 1.28 (t, J = 7.1 Hz, 3H, OCH₂CH₃), 1.57–2.05 (m, 3H), 2.12–2.33 (m, 4H), 2.29 (s, acetylenic, included in m at 2.12–2.33), 2.37-2.58 (m, 3H), 2.67–2.82 (m, 1H), 4.20 (q, J = 7.1 Hz, 2H, OCH₂CH₃); ¹³C NMR δ 14.1, 24.1, 28.2, 35.4, 35.8, 37.6, 58.8, 61.1, 62.4, 73.1, 81.1, 173.9, 213.1; IR (neat) 3275, 2100, 1750, 1730 cm⁻¹.

Reduction of ketone 10. To a solution of **10** (22 mg, 0.10 mmol) in ethanol (5 mL) at 0 °C was added NaBH₄ (5 mg, 0.13 mmol). The mixture was stirred for 30 min and water (2 mL) was added. The mixture was extracted with ether (3 x 20 mL) and the organic extracts were dried (Na₂SO₄). The crude after evaporation was purified by flash chromatography (silica gel, 5% EtOAc in hexanes and then 40% EtOAc in hexanes) to yield the alcohols **3d**, **3d**' (15 mg, 69%, 1:1 diastereomeric mixture) as a colorless oil. The spectral sample was obtained after HPLC (column 2, 20% EtOAc in hexanes, 6 mL/min): t_R 9 (**3d**) and 10 (**3d**') min; ¹H NMR (mixture of isomers) δ 1.28 and 1.29 (t, J = 7.1 Hz, 3H, OCH₂CH₃), 1.39–2.17 (m, 8H), 2.31–2.68 (m, 2H), 2.20 and 2.37 (s, 1H, acetylenic), 2.83 (d, J = 8.2 Hz, OH, one isomer), 3.81-3.88 (m, H-4, **3d**'), 4.11–4.22 (m, 2H, OCH₂CH₃), 4.40 (br t, H-4, **3d**); ¹³C NMR (mixture of isomers) δ 14.(**3d**), 25.3 (**3d**'), 26.2 (**3d**), 31.6 (**3d**'), 32.1 (**3d**), 33.6 (**3d**'), 34.0 (**3d**'), 35.3 (**3d**), 37.9 (**3d**'), 38.3 (**3d**), 39.8 (**3d**'), 56.5 (**3d**), 59.4 (**3d**'), 60.8 (**3d**), 61.1 (**3d**'), 63.9 (**3d**), 64.3 (**3d**'), 70.4 (**3d**), 74.1 (**3d**'), 79.2 (**3d**'), 80.3 (**3d**), 84.7 (**3d**'), 88.5 (**3d**), 175.3 (**3d**), 176.0 (**3d**'); GC-MS t_R 8.65 (**3d**) and 8.82 (**3d**') min.

Ethyl 6-(hex-1-ynyl)-5-oxabicyclo[4.3.0]non-3-enecarboxylate (**8e**). Prepared from **1e** (0.625 mmol) and Pd(PPh₃)₄ (0.031 mmol) in THF (8 mL) following the general procedure. Reaction time: 77 h. The crude was purified by flash chromatography (silica gel, hexanes) to yield the ether **8e** (0.152 g, 88 %) as a colorless oil: 1 H NMR δ 0.85 (t, J = 7.1 Hz, 3H, CH₃), 1.23 (t, J = 7.1 Hz, 3H, OCH₂CH₃), 1.29–1.48 (m, 4H), 1.66–2.10 (m, 6H), 2.15 (t, J = 6.8 Hz, 2H, CH₂C_), 2.26–2.38 (m, 1H), 2.64 (ddd, J = 17.9, 2.6, 2.5 Hz, 1H, H-2), 4.12 (q, J = 7.1 Hz, 1H, OCHCH₃), 4.13 (q, J = 7.1 Hz, 1H, OCHCH₃), 4.69 (ddd, J = 6.2, 5.0, 2.6 Hz, 1H, H-3), 6.21 (ddd, J = 6.2, 2.5, 2.5 Hz, 1H, H-4); 13 C NMR δ 13.4, 14.0, 18.3, 20.4, 21.7, 23.7, 30.5, 31.7, 38.8, 54.9, 60.6, 77.9, 78.8, 85.8, 97.8, 139.7, 173.9; IR (neat) 2230, 1730 (f, C=O) cm⁻¹; HRMS calcd for C₁₇H₂₄O₃ 276.1735, found 276.1732.

Ethyl 4-(hex-1-ynyl)-5-oxabicyclo[4.3.0]non-6-enecarboxylate (9b). Prepared from 4b (0.606 mmol) and Pd(PPh₃)₄ (0.031 mmol) in THF (8 mL) following the general procedure. Reaction time: 19 h. The crude after evaporation was purified by flash chromatography (silica

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gel, 5% EtOAc in hexanes) to yield in order of elution the ether **9b** (0.063 g, 37%) and recovered starting material **4b** (0.077 g, 32%) as colorless oils. Data for **9b**: 1 H NMR δ 0.88 (t, J = 7.0 Hz,3H, C H_3), 1.23 (t, J = 7.1 Hz, 3H, OCH₂C H_3), 1.26–1.52 (m, 4H), 1.60–2.44 (m, 10H), 4.03–4.29 (m, 2H, OC H_2 CH₃), 4.84 (m, 1H, H-4), 5.17 (br s, $W_{1/2} = 8.0$ Hz, 1H, H-7); 13 C NMR δ 13.5, 14.1, 18.2, 21.7, 25.8, 27.9, 29.9, 30.6, 36.8, 51.5, 60.8, 67.8, 77.5, 87.6, 109.4, 152.7, 175.0; IR (neat) 3060, 2210, 1730 cm⁻¹; HRMS calcd for C₁₇H₂₄O₃ 277.1804 (M+1), found 277.1803.

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References

- 1. Aurrecoechea, J. M.; Fañanás, R.; Arrate, M.; Gorgojo, J. M.; Aurrekoetxea, N. J. Org. Chem. **1999**, *64*, 1893.
- Recent reviews on SmI₂ synthetic applications: (a) Molander, G. A. Chem. Rev. 1992, 92, 29. (b) Molander, G. A. In Organic Reactions; L. A. Paquette, Ed.; John Wiley & Sons, New York: 1994; Vol. 46; p 211. (c) Molander, G. A.; Harris, C. R. Chem. Rev. 1996, 96, 307. (d) Molander, G. A.; Harris, C. R. Tetrahedron 1998, 54, 3321. (e) Krief, A.; Laval, A. M. Chem. Rev. 1999, 99, 745.
- 3. Tabuchi, T.; Inanaga, J.; Yamaguchi, M. Chem. Lett. 1987, 2275.
- 4. Aurrecoechea, J. M.; Lopez, B. Tetrahedron Lett. 1998, 39, 2857.
- 5. Aurrecoechea, J. M.; Arrate, M.; López, B. In 12th International Conference on Organic Synthesis (ICOS-12); Book of Abstracts: Venezia (Italy), 1998, 60.
- 6. López, B. *Doctoral Thesis*, Universidad del País Vasco, 1999.
- 7. A portion of this work has been communicated: Aurrecoechea, J. M.; Fañanás-San Antón, R. J. Org. Chem. **1994**, *59*, 702.
- 8. Reetz, M. T.; Wenderoth, B.; Peter, R. J. Chem. Soc., Chem. Commun. 1983, 406.
- 9. Recent reviews on synthetic applications of benzotriazoles: (a) Katritzky, A. R.; Lan, X. F.; Yang, J. Z.; Denisko, O. V. *Chem. Rev.* **1998**, *98*, 409. (b) Katritzky, A. R.; Henderson, S. A.; Yang, B. Z. *J. Heterocycl. Chem.* **1998**, *35*, 1123. (c) Katritzky, A. R.; Piff l, M.; Lang, H.; Anders, E. *Chem. Rev.* **1999**, *99*, 665.
- 10. See, for example, the discussion in: Curran, D. P.; Gu, X.; Zhang, W.; Dowd, P.

ISSN 1424-6376 Page 126 [©]ARKAT USA, Inc

- Tetrahedron 1997, 53, 9023.
- 11. Brown, H. C.; Kramer, G. W.; Levy, A. B.; Midland, M. M. *Organic Synthesis Via Boranes*; Wiley & Sons: New York, 1975.
- 12. Molander, G. A.; Etter, J. B. J. Org. Chem. 1986, 51, 1778.
- 13. Perrin, D. D.; Armarego, W. F. L. *Purification of Laboratory Chemicals*; 3rd ed.; Pergamon Press: Oxford, 1988.
- 14. Still, W. C.; Kahn, M.; Mitra, A. J. Org. Chem. 1978, 43, 2923.
- 15. Cope, A. C.; Synerholm, M. E. J. Am. Chem. Soc. 1950, 72, 5228.
- 16. Prepared following the procedure described in ref 15 for 6a.
- 17. Midland, M. M.; McLoughlin, J. I.; Werley, R. T. Org. Synth. 1989, 68, 14.

ISSN 1424-6376 Page 127 [©]ARKAT USA, Inc