# Regioselective synthesis of $N$-acyl- and $N$-alkyldioxolo[4,5b]phenothiazines 

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#### Abstract

We describe the preparation of new substituted dioxolo[4,5-b]phenothiazines by two slightly different reaction sequences. N-Arylation of [1,3]benzodioxol-5-amine with organolead or organobismuth reagents afforded $N$-aryl[1,3]benzodioxol-5-amines; subsequent Bernthsen thionation gave rise to phenothiazine ring formation and was followed by N -acylation. On the other hand, [1,3]benzodioxol-5-amine was first N -alkylated, the resulting N -alkyl[1,3]benzodioxol-5-amines were N-phenylated, before Bernthsen the final tetracyclic thionation furnished product.


Keywords: Dioxolophenothiazines, arylation, Bernthsen thionation, N-acylation.

## Introduction

The phenothiazines, exemplified by chlorpromazine, are the largest and most widely investigated class of neuroleptic agents. ${ }^{1}$ The important feature of these compounds is that the amino group is separated from the nitrogen atom of the phenothiazine ring by a carbon chain. Although a wide range of derivatives has been described, ${ }^{2}$ the number of polycyclic systems bearing a phenothiazine ring has remained relatively small. We have previously reported the preparation of
new tetracycle derivatives bearing a pyrazole ${ }^{3}$ or a cyclopentane ${ }^{4}$ ring fused to a phenothiazine moiety. We are now involved in the preparation of N -substituted tetracycles featuring a dioxole ring fused to the phenothiazine moiety.

## Results and Discussion

To our knowledge, there are a few syntheses of dioxolo[4,5-b] phenothiazines, ${ }^{5}$ which have been prepared in only four steps using Ullmann coupling of the $N$-aryl $[1,3]$ benzodioxol-5-amines intermediates. Our synthetic approach is based on N -arylation of aromatic primary amines with different organolead or organobismuth reagents in the presence of a copper catalyst. ${ }^{6}$ The resulting $N$-aryl $[1,3]$ benzodioxol-5-amines were subjected to Bernthsen thionation ${ }^{7}$ to yield the corresponding phenothiazines.
The first key step is the synthesis of the diarylamines 3a-d from [1,3]benzodioxol-5-amine 2 by copper catalysis. Ullmann reaction gave only poor yields of desired products. ${ }^{8}$ A modified procedure ${ }^{9}$ using organometallic reagents improved the yield of this N -arylation step. With $p$ tolyllead(IV) triacetate 1 a and (4-methoxyphenyl)lead(IV) triacetate $1 \mathrm{~b}^{10}$ the corresponding coupling products $N$-(4-methylphenyl)[1,3]benzodioxol-5-amine 3 a and N -(4-methoxyphenyl)[1,3]benzodioxol-5-amine 3b were obtained (Scheme 1). Similarly, the $N$ -aryl[1,3]benzodioxol-5-amines 3 c -d were prepared using the arylbismuth reagents 1 c -d. ${ }^{11}$


## Scheme 1

Subsequently, Bernthsen thionation ${ }^{12}$ of diarylamines 3a-d with sulfur and iodine in odichlorobenzene brought about conversion into phenothiazine derivatives. Usually, this cyclization reaction gives rise to mixtures of isomers owing to two possible cyclization sites in the [1,3]benzodioxol moiety; thus, cyclization of compounds 3 is expected togive the linear [b]fused phenothiazine isomer (cyclization at position 6) and the angular [a]fused isomer (cyclization at position 4) (Scheme 2). When applied to compounds 3a-c, the thionation reaction turned out to be regioselective and led to single isomers, the linear dioxolo[4,5-b]phenothiazines $4 a-c$. Under the same conditions the reaction of the chloro derivative 3 d was unsuccessful, leading to many side products of polymerization.


## Scheme 2

The assignment of the linearly fused tetracyclic structure was unambiguously supported by the ${ }^{1} \mathrm{H}$ NMR spectra, in particular, by the evaluation of the multiplet pattern of the C-ring proton signals: In the case of $[b]$ fusion each $4-\mathrm{H}$ and $11-\mathrm{H}$ are expected to resonate as singlets, whereas in the case of [a] fusion, two doublets (AB quartet) would be expected for $4-\mathrm{H}$ and $5-\mathrm{H}$ (Scheme 2). In fact, the former pattern was observed, for example the two singlets at $\delta 6.43$ and 6.61 prove the linear fused structure of $10 \mathrm{H}-[1,3]$ dioxolo $[4,5-b]$ phenothiazine 4 c .

The next step involved the conversion of the phenothiazines 4 into N -acyl and N alkylaminoalkyl derivatives; usually, the preparation of the latter can be achieved with phase transfer catalysis and provides good results when applied to phenothiazine derivatives, ${ }^{13}$ but did not work with tetracycles 4. Acylation of 4a-c with chloroacetylchloride, followed by condensation with diethylamine furnished the corresponding $N$-(2-diethylaminoacetyl) derivatives 5a-c (Scheme 3). Acetic anhydride converted 4c into 10-acetyl-10H-[1,3]dioxolo[4,5b]phenothiazine 6c.


## Scheme 3

Previously described preparative procedures for direct N -amino alkylation of dioxolophenothiazines ${ }^{5 \mathrm{a}, \mathrm{b}}$ using sodium amide in xylene or sodium hydride in DMSO and $\mathrm{N}, \mathrm{N}$ dimethylaminoalkyl halides proved not successful and no recovered material was obtained. We also attempted direct N -alkylation of the $N$-aryl[1,3]benzodioxol-5-amines 3a-d before phenothiazine cyclization but the desired products were not obtained. Therefore, we changed the strategy: In the first step, the aromatic amine 2 was mono-alkylated with different alkyl halides in the presence of sodium hydrogen carbonate in acetonitrile. ${ }^{14} 1$-Bromobutane and 1-bromo-3methylbutane gave $N$-butyl- and $N$-isopentyl[1,3]benzodioxol-5-amines 7 a and 7 b , respectively. The amines 7 c and 7 d were prepared in the same way. Subsequently, the reaction of N -
alkyl $[1,3]$ benzodioxol-5-amines $7 \mathrm{a}-\mathrm{d}$ with triphenylbismuth $(\mathrm{V})$ diacetate 1 c provided the corresponding $N$-alkyl- $N$-phenyl[1,3]benzodioxol-5-amines 8a-d (Scheme 4).


## Scheme 4

Under Bernthsen's condition only two arylamines, 8 a and 8 b were cyclized to linear fused tetracyclic products, 9 a and 9 b . By contrast, the arylamines, 8 c and 8 d only led to degradation products.

## Conclusions

In conclusion, this report describes the preparation of a new class of tetracyclic heterocycles, N -acyl- and $N$-alkyl-10H-[1,3]dioxolo[4,5-b]phenothiazines, employing organometalic reagents for N -arylation and Bernthsen thionation condition for phenothiazine ring closure. Currently, further studies are in progress to explore the scope of this approach for the synthesis of other heterocycles.

## Experimental Section

General Procedures. ${ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ NMR spectra were measured on a BRUKER AC 400 MHz spectrometer. Chemical shifts were recorded as units relative to tetramethylsilane as the internal standard. Separations by chromatography were performed on silica gel (Merck, $70-230$ mesh). 4-Tolyllead(IV) triacetate 1a, 4-methoxyphenyllead(IV) triacetate 1b, triphenylbismuth(V) diacetate 1 c and $\operatorname{tris}(4$-chlorophenyl)bismuth $(\mathrm{V})$ diacetate 1 d were prepared according to reported procedures. ${ }^{10,11}[1,3]$ benzodioxol- 5 -amine 2 was commercially available (JANSSEN) and was used as received.
$N$-(4-Methylphenyl)[1,3]benzodioxol-5-amine (3a). To a solution of [1,3]benzodioxol-5-amine 2 $(1 \mathrm{~g}, 7.3 \mathrm{mmol})$ in dry $\mathrm{CH}_{2} \mathrm{Cl}_{2}(40 \mathrm{~mL})$ at rt were slowly added $p$-tolyllead(IV) triacetate 1a ( 3.8 $\mathrm{g}, 8 \mathrm{mmol}$ ) and copper(II) acetate ( $0.13 \mathrm{~g}, 0.7 \mathrm{mmol}$ ). The mixture was stirred at rt for 4 h . Next, $\mathrm{CH}_{2} \mathrm{Cl}_{2}(20 \mathrm{~mL})$ was added to the solution, and the resulting mixture was filtered. The insoluble part was washed with $\mathrm{CH}_{2} \mathrm{Cl}_{2}(2 \times 20 \mathrm{~mL})$, and the organic layers were dried and evaporated to give a crude reaction product, which was purified by chromatography on silica gel with $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ as the eluant. A white powder 3 a was obtained $(0.66 \mathrm{~g}, 40 \%), \mathrm{mp} 102{ }^{\circ} \mathrm{C} .{ }^{1} \mathrm{H}$-NMR (DMSO- $\mathrm{d}_{6}$ ): $\delta 2.20\left(\mathrm{~s}, \mathrm{CH}_{3}\right), 5.92(\mathrm{~s}, 2 \mathrm{H}), 6.48(\mathrm{dd}, J=8.3,2.2 \mathrm{~Hz}, 1 \mathrm{H}), 6.62(\mathrm{~d}, J=2.2 \mathrm{~Hz}, 1 \mathrm{H}), 6.77(\mathrm{~d}, J=$ $8.3 \mathrm{~Hz}, 1 \mathrm{H}), 6.86(\mathrm{~d}, J=8.1 \mathrm{~Hz}, 2 \mathrm{H}), 6.99(\mathrm{~d}, J=8.1 \mathrm{~Hz}, 2 \mathrm{H}), 7.74(\mathrm{~s}, \mathrm{NH}) ;{ }^{13} \mathrm{C}-\mathrm{NMR}$ (DMSO$\left.d_{6}\right): \delta 20.29,99.97,100.65,108.57,109.60,116.35,127.83,129.61,138.71,140.80,141.98$, 147.75. Anal. Calcd. for $\mathrm{C}_{14} \mathrm{H}_{13} \mathrm{NO}_{2}$ : C, 73.99; H, 5.76; N, 6.16. Found: C, 74.23; H, 5.37; N, 5.90.

## $N$-(4-Methoxyphenyl)[1,3]benzodioxol-5-amine (3b).

As described above, $2(1 \mathrm{~g}, 7.3 \mathrm{mmol})$ and (4-methoxyphenyl)lead(IV) triacetate 1 b ( $3.93 \mathrm{~g}, 8$ mmol ) gave orange needles of $3 \mathrm{~b}(1.2 \mathrm{~g}, 66 \%), \mathrm{mp} 81{ }^{\circ} \mathrm{C} .{ }^{1} \mathrm{H}$-NMR (DMSO- $\mathrm{d}_{6}$ ): $\delta 3.69$ (s, $\left.\mathrm{OCH}_{3}\right), 5.90(\mathrm{~s}, 2 \mathrm{H}), 6.35(\mathrm{dd}, J=8.1,2.2 \mathrm{~Hz}, 1 \mathrm{H}), 6.52(\mathrm{~d}, J=2.2 \mathrm{~Hz}, 1 \mathrm{H}), 6.69(\mathrm{~d}, J=8.2 \mathrm{~Hz}$, $1 \mathrm{H}), 6.86(\mathrm{~d}, J=8.9 \mathrm{~Hz}, 2 \mathrm{H}), 6.99(\mathrm{~d}, J=8.9 \mathrm{~Hz}, 2 \mathrm{H}), 7.60(\mathrm{~s}, \mathrm{NH}) ;{ }^{13} \mathrm{C}-\mathrm{NMR}\left(\mathrm{DMSO}_{6}\right) \delta$ 55.17, 98.66, 100.45, 108.13, 108.50, 114.73, 119.02, 137.38, 139.92, 140.12, 147.71, 153.24. Anal. Calcd. for $\mathrm{C}_{14} \mathrm{H}_{13} \mathrm{NO}_{3}$ : C, 69.12; H, 5.39; N, 5.76. Found: C, 69.51; H, 5.74; N, 5.92.

## $N$-Phenyl[1,3]benzodioxol-5-amine (3c).

As described above, $2(1 \mathrm{~g}, 7.3 \mathrm{mmol})$ and triphenylbismuth $(\mathrm{V})$ diacetate $1 \mathrm{c}(1.5 \mathrm{~g}, 2.7 \mathrm{mmol})$ gave $3 \mathrm{c}(0.84 \mathrm{~g}, 54 \%), \mathrm{mp} 83{ }^{\circ} \mathrm{C} .{ }^{1} \mathrm{H}-\mathrm{NMR}\left(\mathrm{CDCl}_{3}\right): \delta 5.92(\mathrm{~s}, 2 \mathrm{H}), 6.54(\mathrm{dd}, J=8.2,2.2 \mathrm{~Hz}$, $1 \mathrm{H}), 6.69(\mathrm{~d}, J=2.2 \mathrm{~Hz}, 1 \mathrm{H}), 6.73(\mathrm{~d}, J=8.2 \mathrm{~Hz}, 1 \mathrm{H}), 6.85(\mathrm{~m}, 1 \mathrm{H}), 6.92(\mathrm{~m}, 2 \mathrm{H}), 7.22(\mathrm{~m}, 2 \mathrm{H})$; ${ }^{13} \mathrm{C}-\mathrm{NMR}\left(\mathrm{CDCl}_{3}\right): \delta 101.14,102.60,108.62,113.00,116.27,120.06,129.39,137.31,142.92$, 144.70, 148.26. Anal. Calcd. for $\mathrm{C}_{13} \mathrm{H}_{11} \mathrm{NO}_{2}$ : C, 73.22; H, 5.20; N, 6.57. Found: C, 73.30; H, 4.95; N, 6.82.
$N$-(4-Chlorophenyl)[1,3]benzodioxol-5-amine (3d). As described above, $2(1 \mathrm{~g}, 7.3 \mathrm{mmol})$ and $\operatorname{tris}(4$-chlorophenyl)bismuth $(\mathrm{V})$ diacetate $1 \mathrm{~d}(1.8 \mathrm{~g}, 2.7 \mathrm{mmol})$ gave a white powder $3 \mathrm{~d}(0.27 \mathrm{~g}$, $41 \%$ ), mp $78{ }^{\circ} \mathrm{C} .{ }^{1} \mathrm{H}-\mathrm{NMR}\left(\mathrm{DMSO}_{-}\right.$): $\delta 5.96(\mathrm{~s}, 2 \mathrm{H}), 6.54$ (dd, $\left.J=8.3,2.2 \mathrm{~Hz}, 1 \mathrm{H}\right), 6.68$ (dd, $J$ $=8.2,2.2 \mathrm{~Hz}, 1 \mathrm{H}), 6.82(\mathrm{~d}, J=8.3 \mathrm{~Hz}, 1 \mathrm{H}), 6.92(\mathrm{~d}, J=8.2 \mathrm{~Hz}, 2 \mathrm{H}), 7.19(\mathrm{~d}, J=8.8 \mathrm{~Hz}, 2 \mathrm{H})$, 8.05 (s, NH); ${ }^{13} \mathrm{C}-\mathrm{NMR}\left(\mathrm{DMSO}_{6}\right): ~ \delta 100.90,101.49,108.64,111.70,116.54,121.69,128.96$, 137.13, 141.87, 143.96, 147.86. Anal. Calcd. for $\mathrm{C}_{13} \mathrm{H}_{10} \mathrm{NO}_{2} \mathrm{Cl}: \mathrm{C}, 63.04 ; \mathrm{H}, 4.07$; $\mathrm{N}, 5.66$. Found: C, 63.30; H, 4.35; N, 5.96.

## 7-Methyl-10H-[1,3]dioxolo[4,5-b]phenothiazine (4a).

A mixture of $N$-(4-methylphenyl)[1,3]benzodioxol-5-amine $3 \mathrm{a}(0.5 \mathrm{~g}, 2.2 \mathrm{mmol}$ ), sulphur ( 0.15 $\mathrm{g}, 4.6 \mathrm{mmol}$ ), and one iodine crystal was refluxed under nitrogen in dry o-dichlorobenzene ( 4 mL ) during 6 h . The mixture was extracted with $\mathrm{Et}_{2} \mathrm{O}(15 \mathrm{~mL})$, filtered and concentrated. The resulting oil was chromatographed on silica gel with toluene to elute first the solvent ( $o$ dichlorobenzene), and next a red powder $4 \mathrm{a}\left(80 \mathrm{mg}, 15 \%\right.$ ), mp $184{ }^{\circ} \mathrm{C} .{ }^{1} \mathrm{H}-\mathrm{NMR}$ (DMSO-d $\mathrm{d}_{6}$ ): $\delta$ $2.13\left(\mathrm{~s}, \mathrm{CH}_{3}\right), 5.90(\mathrm{~s}, 2 \mathrm{H}), 6.40(\mathrm{~s}, 1 \mathrm{H}), 6.57(\mathrm{~d}, J=8.0 \mathrm{~Hz}, 1 \mathrm{H}), 6.60(\mathrm{~s}, 1 \mathrm{H}), 6.75(\mathrm{br} \mathrm{s}, 1 \mathrm{H})$, 6.80 (br d, $J=7.9 \mathrm{~Hz}, 1 \mathrm{H}), 8.28(\mathrm{~s}, \mathrm{NH}) ;{ }^{13} \mathrm{C}-\mathrm{NMR}\left(\mathrm{DMSO}-d_{6}\right): \delta 20.03,96.95,101.06,106.55$, $106.65,114.19,116.67,126.50,128.01,130.75,137.80,140.49,142.30,147.03$, Anal. Calcd. for $\mathrm{C}_{14} \mathrm{H}_{11} \mathrm{NO}_{2} \mathrm{~S}: \mathrm{C}, 65.35 ; \mathrm{H}, 4.31$; N, 5.44. Found: C, $65.51 ; \mathrm{H}, 4.78 ; \mathrm{N}, 5.63$.

7-Methoxy-10H-[1,3]dioxolo[4,5-b]phenothiazine (4b). As described above, $N$-(4-methoxyphenyl)[1,3]benzodioxol-5-amine $3 \mathrm{~b}(0.5 \mathrm{~g}, 2 \mathrm{mmol})$ gave after chromatography with ethyl acetate as eluant a yellow powder $4 \mathrm{~b}\left(140 \mathrm{mg}, 25 \%\right.$ ), mp $186{ }^{\circ} \mathrm{C} .{ }^{1} \mathrm{H}-\mathrm{NMR}$ (DMSO- $d_{6}$ ): $\delta$ $3.65\left(\mathrm{~s}, \mathrm{OCH}_{3}\right), 5.90(\mathrm{~s}, 2 \mathrm{H}), 6.39(\mathrm{~s}, 1 \mathrm{H}), 6.60(\mathrm{~m}, 2 \mathrm{H}), 6.61(\mathrm{~s}, 1 \mathrm{H}), 6.62(\mathrm{br} \mathrm{s}, 1 \mathrm{H}), 8.19(\mathrm{~s}$, NH), ${ }^{13} \mathrm{C}-\mathrm{NMR}$ (DMSO-d6): $\delta 55.47,96.87,101.04,106.21,106.51,111.58,113.21,114.95$, 118.04, 136.51, 138.31, 142.17, 147.10, 154.69. Anal. Calcd. for $\mathrm{C}_{14} \mathrm{H}_{11} \mathrm{NO}_{3} \mathrm{~S}: \mathrm{C}, 61.52 ; \mathrm{H}, 4.06$; N, 5.12. Found: C, 61.87 ; H, 3.81; N, 5.30.

## 10H-[1,3]Dioxolo[4,5-b]phenothiazine (4c).

As described above, $N$-phenyl[1,3]benzodioxol-5-amine $3 \mathrm{c}(0.5 \mathrm{~g}, 2.3 \mathrm{mmol}$ ), sulphur ( 0.15 g , 4.6 mmol ) gave after chromatography with toluene as eluant a white powder $4 \mathrm{c}(180 \mathrm{mg}, 32 \%)$, $\mathrm{mp} 202{ }^{\circ} \mathrm{C} .{ }^{1} \mathrm{H}-\mathrm{NMR}\left(\mathrm{DMSO}_{6}\right): \delta 5.91(\mathrm{~s}, 2 \mathrm{H}), 6.43(\mathrm{~s}, 1 \mathrm{H}), 6.61(\mathrm{~s}, 1 \mathrm{H}), 6.68$ (dd, $J=7.9,1.0$ $\mathrm{Hz}, 1 \mathrm{H}), 6.76(\mathrm{td}, J=7.6,1.1 \mathrm{~Hz}, 1 \mathrm{H}), 6.92(\mathrm{dd}, J=7.6,1.0 \mathrm{~Hz}, 1 \mathrm{H}), 6.99(\mathrm{td}, J=7.7,1.1 \mathrm{~Hz}$, $1 \mathrm{H}), 8.42(\mathrm{~s}, \mathrm{NH}) .{ }^{13} \mathrm{C}-\mathrm{NMR}\left(\mathrm{DMSO}-d_{6}\right): \delta 97.09,101.11,106.55,106.73,114.35,116.79$, 121.83, 126.27, 127.59, 137.48, 142.54, 143.03, 147.11. Anal. Calcd. for $\mathrm{C}_{13} \mathrm{H}_{9} \mathrm{NO}_{2} \mathrm{~S}: \mathrm{C}, 64.18$; H, 3.73; N, 5.76. Found: C, 64.51; H, 4.02; N, 5.93.

## 2-Diethylamino-1-(7-methyl-10H-[1,3]dioxolo[4,5-b]phenothiazin-10-yl)ethan-1-one

(5a).
To a solution of 7-methyl-10H-[1,3]dioxolo[4,5-b]phenothiazine $4 \mathrm{a}(0.18 \mathrm{~g}, 0.7 \mathrm{mmol}$ ) and toluene ( 7 mL ) was added chloroacetyl chloride ( $5.7 \mathrm{mg}, 0.5 \mathrm{mmol}$ ), and the mixture was kept at $35{ }^{\circ} \mathrm{C}$ under stirring during 45 min . The solution was concentrated, and to the residual viscous oil a solution of $N, N$-diethylamine ( 2 mL ) in toluene ( 4 mL ) was added. The solution was refluxed 2 $h$ under stirring and evaporated to yield a brown oil ( $170 \mathrm{mg}, 65 \%$ ). ${ }^{1} \mathrm{H}-\mathrm{NMR}$ (DMSO- $d_{6}$ ): $\delta 0.92$ $(\mathrm{t}, J=7.1 \mathrm{~Hz}, 6 \mathrm{H}), 2.32\left(\mathrm{~s}, \mathrm{CH}_{3}\right), 2.65(\mathrm{q}, J=7.1 \mathrm{~Hz}, 4 \mathrm{H}), 3.40(\mathrm{~s}, 2 \mathrm{H}), 6.07(\mathrm{br} \mathrm{s}, 2 \mathrm{H}), 7.08(\mathrm{~s}$, $1 \mathrm{H}), 7.18$ (br d, $J=8.1 \mathrm{~Hz}, 1 \mathrm{H}), 7.24(\mathrm{~s}, 1 \mathrm{H}), 7.34(\mathrm{br} \mathrm{s}, 1 \mathrm{H}), 7.48(\mathrm{~d}, J=8.1 \mathrm{~Hz}, 1 \mathrm{H}) ;{ }^{13} \mathrm{C}-\mathrm{NMR}$ (DMSO-d $d_{6}$ : $\delta 11.38,20.44,47.50,54.10,102.24,107.36,108.18,124.74,126.65,127.93$, 127.93, 132.47, 132.76, 136.16, 136.79, 146.27, 147.12, 168.18. Anal. Calcd. for $\mathrm{C}_{20} \mathrm{H}_{22} \mathrm{~N}_{2} \mathrm{O}_{3} \mathrm{~S}$ : C, 64.84; H, 5.99; N, 7.56. Found: C, 65.03; H, 5.74; N, 7.80.

2-Diethylamino-1-(7-methoxy-10H-[1,3]dioxolo[4,5-b]phenothiazin-10-yl)ethan-1-one (5b). As described above, from $4 \mathrm{~b}(0.18 \mathrm{~g}, 0.7 \mathrm{mmol})$ after work up a brown oil $5 \mathrm{~b}(200 \mathrm{mg}, 75 \%)$ was obtained. ${ }^{1} \mathrm{H}-\mathrm{NMR}\left(\mathrm{DMSO}_{6}\right): \delta 0.80(\mathrm{t}, J=7.1 \mathrm{~Hz}, 6 \mathrm{H}), 2.44(\mathrm{q}, J=7.1 \mathrm{~Hz}, 4 \mathrm{H}), 3.31(\mathrm{~s}$, $2 \mathrm{H}), 3.76\left(\mathrm{~s}, \mathrm{OCH}_{3}\right), 6.08(\mathrm{br} \mathrm{s}, 2 \mathrm{H}), 6.91(\mathrm{dd}, J=8.7,2.6 \mathrm{~Hz}, 1 \mathrm{H}), 7.06(\mathrm{~d}, J=2.4 \mathrm{~Hz}, 1 \mathrm{H})$, $7.09(\mathrm{~s}, 1 \mathrm{H}), 7.25(\mathrm{~s}, 1 \mathrm{H}), 7.46(\mathrm{~d}, J=8.7 \mathrm{~Hz}, 1 \mathrm{H}) ;{ }^{13} \mathrm{C}-\mathrm{NMR}\left(\mathrm{DMSO}-d_{6}\right): \delta 12.10,46.88,54.65$, $55.73,102.21,107.36,108.19,112.35,113.21,124.68,127.83,131.83,132.48,133.61,146.15$, 147.10, 157.38, 169.42. Anal. Calcd. for $\mathrm{C}_{20} \mathrm{H}_{22} \mathrm{~N}_{2} \mathrm{O}_{4} \mathrm{~S}: \mathrm{C}, 62.16 ; \mathrm{H}, 5.74 ; \mathrm{N}, 7.25$. Found: C, 62.47; H, 5.85; N, 7.41.

2-Diethylamino-1-(10H-[1,3]dioxolo[4,5-b]phenothiazin-10-yl)ethan-1-one (5c). As described above, $4 \mathrm{c}(0.1 \mathrm{~g}, 0.7 \mathrm{mmol})$ gave a red oil $5 \mathrm{c}(180 \mathrm{mg}, 73 \%) .{ }^{1} \mathrm{H}-\mathrm{NMR}\left(\mathrm{DMSO}-d_{6}\right): \delta 0.81(\mathrm{t}, J$ $=7.0 \mathrm{~Hz}, 6 \mathrm{H}), 2.45(\mathrm{br} \mathrm{q}, J=6.6 \mathrm{~Hz}, 4 \mathrm{H}), 3.36(\mathrm{~s}, 2 \mathrm{H}), 6.05(\mathrm{~s}, 2 \mathrm{H}), 7.13(\mathrm{~s}, 1 \mathrm{H}), 7.28(\mathrm{t}, J=7.5$ $\mathrm{Hz}, 1 \mathrm{H}), 7.29(\mathrm{t}, J=7.5 \mathrm{~Hz}, 1 \mathrm{H}), 7.37(\mathrm{td}, J=7.5,1.2 \mathrm{~Hz}, 1 \mathrm{H}), 7.53(\mathrm{dd}, J=7.7,1.0 \mathrm{~Hz}, 1 \mathrm{H})$, $7.59(\mathrm{~d}, J=7.7 \mathrm{~Hz}, 1 \mathrm{H}) ;{ }^{13} \mathrm{C}-\mathrm{NMR}\left(\mathrm{DMSO}-d_{6}\right): \delta 11.95,46.83,54.57,102.23,107.35,108.28$,
$124.60,126.85,127.21,127.21,127.72,132.50,132.85,138.92,146.12,147.09,169.08$. Anal. Calcd. for $\mathrm{C}_{19} \mathrm{H}_{20} \mathrm{~N}_{2} \mathrm{O}_{3} \mathrm{~S}: \mathrm{C}, 64.02$; H, 5.66; N, 7.86. Found: C, 64.29; H, 5.91; N, 8.01.

## 10-Acetyl-10H-[1,3]dioxolo[4,5-b]phenothiazine (6c).

A mixture of $4 \mathrm{c}(0.2 \mathrm{~g}, 0.8 \mathrm{mmol})$ in acetic anhydride $(5 \mathrm{~mL})$ was stirred at rt during 8 h . The solution was filtrated and evaporated to yield a red oil $6 \mathrm{c}(0.107,47 \%)$ was left. ${ }^{1} \mathrm{H}-\mathrm{NMR}$ (DMSO-d $\mathrm{d}_{6}$ : $\delta 2.11\left(\mathrm{~s}, \mathrm{CH}_{3}\right), 6.06(\mathrm{~s}, 2 \mathrm{H}), 7.13(\mathrm{~s}, 1 \mathrm{H}), 7.26(\mathrm{br} \mathrm{s}, 1 \mathrm{H}), 7.29(\mathrm{t}, J=7.5 \mathrm{~Hz}, 1 \mathrm{H})$, $7.38(\mathrm{t}, J=7.2 \mathrm{~Hz}, 1 \mathrm{H}), 7.54(\mathrm{~d}, J=7.5 \mathrm{~Hz}, 1 \mathrm{H}), 7.56(\mathrm{~m}, 1 \mathrm{H}) ;{ }^{13} \mathrm{C}-\mathrm{NMR}\left(\mathrm{DMSO}_{6}\right) \delta 21.31$, 102.24, 107.27, 108.47, 124.55, 126.68, 127.24, 127.38, 127.70, 132.74, 132.74, 139.08, 146.14, 147.10, 172.23. Anal. Calcd. for $\mathrm{C}_{15} \mathrm{H}_{11} \mathrm{NO}_{3} \mathrm{~S}: \mathrm{C}, 63.14 ; \mathrm{H}, 3.89 ; \mathrm{N}, 4.91$. Found: C, 63.22; H , 4.02; N, 4.96.

## $N$-Butyl[1,3]benzodioxol-5-amine (7a).

To a solution of $2(2 \mathrm{~g}, 14.6 \mathrm{mmol})$ in dry acetonitrile ( 30 mL ) was added 1-bromobutane ( 2.2 g , $16 \mathrm{mmol})$ and $\mathrm{NaHCO}_{3}(1.3 \mathrm{~g}, 15.4 \mathrm{mmol})$. The solution was refluxed under stirring during 8 h , neutralised with $\mathrm{HCl}(2 N, 7 \mathrm{~mL})$ and methylene chloride ( 20 mL ) was added. The organic phase was separated, washed twice with water $(40 \mathrm{~mL})$ and evaporated. The residue was dissolved in $\mathrm{MeOH}(20 \mathrm{~mL})$, acidified with $\mathrm{H}_{2} \mathrm{SO}_{4}(3 \mathrm{~mL})$, filtrated, and the filtrate was concentrated to 5 mL volume. The mixture was neutralized with $\mathrm{NaHCO}_{3}(1.5 \mathrm{~g}, 17.9 \mathrm{mmol})$ and dissolved in methylene chloride ( 20 mL ). The organic phase was separated, washed twice with water ( 20 mL ) and evaporated to give a yellow oil $7 \mathrm{a}(1.6 \mathrm{~g}, 58 \%) .{ }^{1} \mathrm{H}-\mathrm{NMR}\left(\mathrm{DMSO}-d_{6}\right): \delta 0.90(\mathrm{t}, J=7.3 \mathrm{~Hz}$, $3 \mathrm{H}), 1.40(\mathrm{~m}, 2 \mathrm{H}), 1.49(\mathrm{~m}, 2 \mathrm{H}), 2.90(\mathrm{t}, J=7.0 \mathrm{~Hz}, 2 \mathrm{H}), 5.20(\mathrm{br} \mathrm{s}, \mathrm{NH}), 5.81(\mathrm{~s}, 2 \mathrm{H}), 5.95$ (dd, $J=8.3,2.3 \mathrm{~Hz}, 1 \mathrm{H}), 6.24(\mathrm{~d}, J=2.3 \mathrm{~Hz}, 1 \mathrm{H}), 6.63(\mathrm{~d}, J=8.3 \mathrm{~Hz}, 1 \mathrm{H}) ;{ }^{13} \mathrm{C}-\mathrm{NMR}\left(\mathrm{DMSO}-d_{6}\right): \delta$ 13.89, 19.89, 30.95, 43.44, 94.93, 99.89, 102.96, 108.46, 137.73, 145.16, 147.78. Anal. Calcd. for $\mathrm{C}_{11} \mathrm{H}_{15} \mathrm{NO}_{2}$ : C, 68.37; H, 7.82; $\mathrm{N}, 7.25$. Found: C, $68.15 ; \mathrm{H}, 7.94 ; \mathrm{N}, 6.98$.

## $N$-Isopentyl[1,3]benzodioxol-5-amine (7b).

As described above, 1-bromo-3-methylbutane ( $2.4 \mathrm{~g}, 16 \mathrm{mmol}$ ) and $\mathrm{NaHCO}_{3}(1.3 \mathrm{~g}, 15.4 \mathrm{mmol})$ gave a yellow oil $7 \mathrm{~b}(1.8 \mathrm{~g}, 61 \%) .{ }^{1} \mathrm{H}-\mathrm{NMR}\left(\mathrm{DMSO}-d_{6}\right): \delta 0.89(\mathrm{~d}, J=6.6 \mathrm{~Hz}, 6 \mathrm{H}), 1.66(\mathrm{~m}$, $1 \mathrm{H}), 1.42(\mathrm{~m}, 2 \mathrm{H}), 2.98(\mathrm{t}, J=7.3 \mathrm{~Hz}, 2 \mathrm{H}), 5.86(\mathrm{~s}, 2 \mathrm{H}), 6.14(\mathrm{dd}, J=8.4,1.5 \mathrm{~Hz}, 1 \mathrm{H}), 6.40(\mathrm{~d}, J$ $=1.6 \mathrm{~Hz}, 1 \mathrm{H}), 6.70(\mathrm{~d}, J=8.4 \mathrm{~Hz}, 1 \mathrm{H}) ;{ }^{13} \mathrm{C}-\mathrm{NMR}\left(\mathrm{DMSO}-d_{6}\right): \delta 22.34,25.27,37.00,43.23$, $96.49,100.21,105.35,108.38,139.47,142.27,147.73$. Anal. Calcd. for $\mathrm{C}_{12} \mathrm{H}_{17} \mathrm{NO}_{2}$ : C, $69.54 ; \mathrm{H}$, 8.27; N, 6.76. Found: C, 69.17; H, 8.74; N, 6.36.
$N$-Isopropyl[1,3]benzodioxol-5-amine (7c). As described above, 2-bromopropane ( $1.97 \mathrm{~g}, 16$ mmol ) and $\mathrm{NaHCO}_{3}\left(2.2 \mathrm{~g}, 26.3 \mathrm{mmol}\right.$ ) gave an orange oil $7 \mathrm{c}(1.2 \mathrm{~g}, 47 \%) .{ }^{1} \mathrm{H}-\mathrm{NMR}$ (DMSO$\left.d_{6}\right): \delta 1.08(\mathrm{~d}, J=6.3 \mathrm{~Hz}, 6 \mathrm{H}), 3.42(\mathrm{hept}, J=6.3 \mathrm{~Hz}, 1 \mathrm{H}), 4.98(\mathrm{~s}, \mathrm{NH}), 5.83(\mathrm{~s}, 2 \mathrm{H}), 5.96$ (dd, $J$ $=8.4,2.3 \mathrm{~Hz}, 1 \mathrm{H}), 6.24(\mathrm{~d}, J=2.2 \mathrm{~Hz}, 1 \mathrm{H}), 6.63(\mathrm{~d}, J=8.3 \mathrm{~Hz}, 1 \mathrm{H}) ;{ }^{13} \mathrm{C}-\mathrm{NMR}\left(\mathrm{DMSO}-d_{6}\right): \delta$ $22.56,43.81,95.94,99.92,103.86,108.58,137.65,144.17,147.85$. Anal. Calcd. for $\mathrm{C}_{10} \mathrm{H}_{13} \mathrm{NO}_{2}$ : C, 67.02; H, 7.31; N, 7.82. Found: C, 67.25; H, 7.48; N, 8.19.

## $N$-(1-Methylbutyl)[1,3]benzodioxol-5-amine (7d).

As described above, 2-bromopentane ( $2.4 \mathrm{~g}, 16 \mathrm{mmol}$ ) and $\mathrm{NaHCO}_{3}(1.3 \mathrm{~g}, 15.4 \mathrm{mmol})$ gave a yellow oil 7d ( $1.6 \mathrm{~g}, 52 \%$ ). ${ }^{1} \mathrm{H}-\mathrm{NMR}\left(\mathrm{DMSO}-\mathrm{d}_{6}\right): \delta 0.87(\mathrm{t}, J=7.3 \mathrm{~Hz}, 3 \mathrm{H}), 1.04(\mathrm{~d}, J=6.3$ $\mathrm{Hz}, 3 \mathrm{H}), 1.33(\mathrm{~m}, 2 \mathrm{H}), 1.45(\mathrm{~m}, 2 \mathrm{H}), 3.27(\mathrm{~m}, 1 \mathrm{H}), 5.08(\mathrm{br} \mathrm{s}, \mathrm{NH}), 5.81(\mathrm{~s}, 2 \mathrm{H}), 5.97$ (dd, $J=$ $8.2,2.1 \mathrm{~Hz}, 1 \mathrm{H}), 6.24(\mathrm{~d}, J=2.1 \mathrm{~Hz}, 1 \mathrm{H}), 6.62(\mathrm{~d}, J=8.3 \mathrm{~Hz}, 1 \mathrm{H}) ;{ }^{13} \mathrm{C}-\mathrm{NMR}\left(\mathrm{DMSO}-d_{6}\right): \delta$ 14.15, 18.95, 20.39, 38.62, 47.96, 95.47, 99.95, 103.78, 108.62, 137.65, 144.23, 147.87. Anal. Calcd. for $\mathrm{C}_{12} \mathrm{H}_{17} \mathrm{NO}_{2}$ : C, 69.54; H, 8.27; N, 6.76. Found: C, $69.81 ; \mathrm{H}, 7.98 ; \mathrm{N}, 6.90$.

## $N$-Butyl- $N$-phenyl[1,3]benzodioxol-5-amine (8a).

To a solution of $7 \mathrm{a}(1 \mathrm{~g}, 5.2 \mathrm{mmol})$ in dry $\mathrm{CH}_{2} \mathrm{Cl}_{2}(30 \mathrm{~mL})$ at rt were slowly added triphenylbismuth $(\mathrm{V})$ diacetate $1 \mathrm{c}(1 \mathrm{~g}, 1.8 \mathrm{mmol})$ and copper(II) acetate $(0.09 \mathrm{~g}, 0.5 \mathrm{mmol})$. The mixture was stirred at rt during 4 h . Next, $\mathrm{CH}_{2} \mathrm{Cl}_{2}(20 \mathrm{~mL})$ was added, and the resulting mixture was filtered. The insoluble part was washed with $\mathrm{CH}_{2} \mathrm{Cl}_{2}(2 \times 20 \mathrm{~mL})$, and the organic layers were dried and evaporated to give a crude reaction product, which was purified by chromatography on silica gel with toluene as the eluent. A yellow oil was recovered, yielding $0.30 \mathrm{~g}(22 \%)$ of $8 \mathrm{a} .{ }^{1} \mathrm{H}-\mathrm{NMR}\left(\mathrm{DMSO}-\mathrm{d}_{6}\right): \delta 0.87(\mathrm{t}, J=7.4 \mathrm{~Hz}, 3 \mathrm{H}), 1.31(\mathrm{~m}, 2 \mathrm{H}), 1.52(\mathrm{~m}$, $2 \mathrm{H}), 3.56(\mathrm{t}, J=7.6 \mathrm{~Hz}, 2 \mathrm{H}), 6.01(\mathrm{~s}, 2 \mathrm{H}), 6.56(\mathrm{dd}, J=8.3,2.2 \mathrm{~Hz}, 1 \mathrm{H}), 6.68(\mathrm{~d}, J=2.0 \mathrm{~Hz}$, $1 \mathrm{H}), 6.69(\mathrm{~m}, 2 \mathrm{H}), 6.70(\mathrm{~m}, 1 \mathrm{H}), 6.89(\mathrm{~d}, J=8.3 \mathrm{~Hz}, 1 \mathrm{H}), 7.14(\mathrm{~m}, 2 \mathrm{H}) ;{ }^{13} \mathrm{C}-\mathrm{NMR}$ (DMSO-d6): $\delta 13.86,19.67,29.18,51.52,101.21,106.82,108.75,115.95,118.10,118.36,129.89,141.64$, 143.82, 148.11, 148.66. Anal. Calcd. for $\mathrm{C}_{17} \mathrm{H}_{19} \mathrm{NO}_{2}$ : C, 75.81 ; H, 7.11; N, 5.20. Found: C, 76.03; H, 7.45; N, 5.47.

## $N$-Isopentyl- $N$-phenyl[1,3]benzodioxol-5-amine (8b).

As described above, $7 \mathrm{~b}(1 \mathrm{~g}, 4.8 \mathrm{mmol})$ and triphenylbismuth $(\mathrm{V})$ diacetate $1 \mathrm{c}(1.1 \mathrm{~g}, 1.9 \mathrm{mmol})$ gave an orange oil $8 \mathrm{~b}(0.45 \mathrm{~g}, 33 \%) .{ }^{1} \mathrm{H}-\mathrm{NMR}\left(\mathrm{DMSO}-\mathrm{d}_{6}\right): \delta 0.88(\mathrm{~d}, J=6.6 \mathrm{~Hz}, 6 \mathrm{H}), 1.61(\mathrm{~m}$, $1 \mathrm{H}), 1.44(\mathrm{td}, J=7.8,6.7 \mathrm{~Hz}, 2 \mathrm{H}), 3.58(\mathrm{t}, J=7.6 \mathrm{~Hz}, 2 \mathrm{H}), 6.02(\mathrm{~s}, 2 \mathrm{H}), 6.56(\mathrm{dd}, J=8.3,2.2$ $\mathrm{Hz}, 1 \mathrm{H}), 6.68(\mathrm{~m}, 1 \mathrm{H}), 6.69(\mathrm{~d}, J=2.2 \mathrm{~Hz}, 1 \mathrm{H}), 6.71(\mathrm{~m}, 2 \mathrm{H}), 6.90(\mathrm{~d}, J=8.2 \mathrm{~Hz}, 1 \mathrm{H}), 7.15(\mathrm{~m}$, $2 \mathrm{H}) ;{ }^{13} \mathrm{C}-\mathrm{NMR}\left(\mathrm{DMSO}-d_{6}\right): \delta 22.53,25.64,35.73,50.19,101.22,106.84,108.76,115.82$,
118.06, 118.40, 129.02, 141.59, 143.85, 148.11, 148.56. Anal. Calcd. for $\mathrm{C}_{18} \mathrm{H}_{21} \mathrm{NO}_{2}$ : C, 76.29; H, 7.47; N, 4.94. Found: C, 76.51; H, 7.80; N, 5.12.
$N$-Isopropyl- $N$-phenyl-[1,3]benzo-dioxol-5-amine (8c). As described above, 7c (1g, 5.6 mmol ) and triphenylbismuth $(\mathrm{V})$ diacetate $1 \mathrm{c}(1.2 \mathrm{~g}, 2.2 \mathrm{mmol})$ gave a pale yellow oil $8 \mathrm{c}(0.54 \mathrm{~g}$, $38 \%){ }^{1} \mathrm{H}-\mathrm{NMR}\left(\mathrm{DMSO}-d_{6}\right): \delta 1.07(\mathrm{~d}, J=6.5 \mathrm{~Hz}, 6 \mathrm{H}), 4.24$ (hept, $\left.J=6.5 \mathrm{~Hz}, 1 \mathrm{H}\right), 6.05(\mathrm{~s}, 2 \mathrm{H})$, $6.50(\mathrm{dd}, J=8.2,2.1 \mathrm{~Hz}, 1 \mathrm{H}), 6.54(\mathrm{~d}, J=8.1 \mathrm{~Hz}, 1 \mathrm{H}), 6.61(\mathrm{~d}, J=2.1 \mathrm{~Hz}, 1 \mathrm{H}), 6.65(\mathrm{~m}, 1 \mathrm{H})$, $6.94(\mathrm{~m}, 2 \mathrm{H}), 7.11(\mathrm{~m}, 2 \mathrm{H}) ;{ }^{13} \mathrm{C}-\mathrm{NMR}\left(\mathrm{DMSO}-\mathrm{d}_{6}\right): \delta 20.75,47.14,101.43,108.65,110.53$, $115.24,117.30,118.70,128.96,137.12,145.21,148.07,148.65$. Anal. Calcd. for $\mathrm{C}_{16} \mathrm{H}_{17} \mathrm{NO}_{2}: \mathrm{C}$, 75.27 ; H, 6.71; N, 5.49. Found: C, 75.53; H, 6.39; N, 5.79.
$N$-(1-Methylbutyl)- $N$-phenyl[1,3]benzodioxol-5-amine (8d). As described above, 7d (1g, 4.8 $\mathrm{mmol})$ and triphenylbismuth $(\mathrm{V})$ diacetate $1 \mathrm{c}(1.1 \mathrm{~g}, 1.9 \mathrm{mmol})$ gave a brown oil $8 \mathrm{~d}(0.20 \mathrm{~g}$, $15 \%) .{ }^{1} \mathrm{H}-\mathrm{NMR}\left(\mathrm{DMSO}-d_{6}\right): \delta 0.88(\mathrm{t}, J=7.2 \mathrm{~Hz}, 3 \mathrm{H}), 1.08(\mathrm{~d}, J=6.6 \mathrm{~Hz}, 3 \mathrm{H}), 1.22(\mathrm{~m}, 2 \mathrm{H})$, $1.36(\mathrm{~m}, 2 \mathrm{H}), 4.06(\mathrm{~m}, 1 \mathrm{H}), 6.02(\mathrm{~s}, 2 \mathrm{H}), 6.50(\mathrm{dd}, J=8.2,2.1 \mathrm{~Hz}, 1 \mathrm{H}), 6.56(\mathrm{~m}, 2 \mathrm{H}), 6.58(\mathrm{~d}, J$ $=2.0 \mathrm{~Hz}, 1 \mathrm{H}), 6.65(\mathrm{~m}, 1 \mathrm{H}), 6.92(\mathrm{~d}, J=8.2 \mathrm{~Hz}, 1 \mathrm{H}), 7.11(\mathrm{~m}, 2 \mathrm{H}) ;{ }^{13} \mathrm{C}-\mathrm{NMR}\left(\mathrm{DMSO}-d_{6}\right): \delta$ $13.91,18.45,19.52,37.02,51.81,101.20,108.42,109.84,115.57,117.35,121.97,128.76$, 137.59, 144.79, 147.89, 148.74. Anal. Calcd. for $\mathrm{C}_{18} \mathrm{H}_{21} \mathrm{NO}_{2}$ : C, 76.29; H, 7.47; N, 4.94. Found: C, 76.34; H, 7.63; N, 5.22.

## 10-Butyl-10H-[1,3]dioxolo[4,5-b]phenothiazine (9a).

A solution of $8 \mathrm{a}(0.30 \mathrm{~g}, 1.1 \mathrm{mmol})$, sulphur $(0.07 \mathrm{~g}, 2.2 \mathrm{mmol})$, and one iodine crystal was refluxed under nitrogen in dry o-dichlorobenzene ( 2 mL ) during 8 h . The mixture was extracted with $\mathrm{Et}_{2} \mathrm{O}(10 \mathrm{~mL})$, filtered and concentrated. The resulting oil was chromatographed on silica gel with $\mathrm{Et}_{2} \mathrm{O}$ to elute first the solvent (o-dichlorobenzene) followed by a viscous green oil 8 a $(0.07 \mathrm{~g}, 21 \%) .{ }^{1} \mathrm{H}-\mathrm{NMR}\left(\mathrm{DMSO}_{6}\right): \delta 0.89(\mathrm{t}, J=7.4 \mathrm{~Hz}, 3 \mathrm{H}), 1.40(\mathrm{~m}, 2 \mathrm{H}), 1.66(\mathrm{~m}, 2 \mathrm{H}), 3.81$ $(\mathrm{t}, J=6.8 \mathrm{~Hz}, 2 \mathrm{H}), 5.95(\mathrm{~s}, 2 \mathrm{H}), 6.72(\mathrm{~s}, 1 \mathrm{H}), 6.73(\mathrm{~s}, 1 \mathrm{H}), 6.92(\mathrm{brt}, J=7.5 \mathrm{~Hz}, 1 \mathrm{H}), 6.98(\mathrm{br} \mathrm{d}$, $J=7.9 \mathrm{~Hz}, 1 \mathrm{H}), 7.12(\mathrm{dd}, J=7.7,1.5 \mathrm{~Hz}, 1 \mathrm{H}), 7.18(\mathrm{td}, J=7.8,1.5 \mathrm{~Hz}, 1 \mathrm{H}) ;{ }^{13} \mathrm{C}-\mathrm{NMR}$ (DMSO$\left.d_{6}\right): \delta 13.64,19.44,28.95,46.70,98.97,101.46,107.18,114.58,115.93,122.36,124.94,127.01$, 127.52, 140.17, 142.88, 145.84, 147.67. Anal. Calcd. for $\mathrm{C}_{17} \mathrm{H}_{17} \mathrm{NO}_{2} \mathrm{~S}: \mathrm{C}, 68.20 ; \mathrm{H}, 5.72 ; \mathrm{N}$, 4.68. Found: C, 68.33; H, 5.87; N, 4.49.

## 10-Isopentyl-10H-[1,3]dioxolo[4,5-b]phenothiazine (9b).

A solution of $8 \mathrm{~b}(0.45 \mathrm{~g}, 1.6 \mathrm{mmol})$, sulphur $(0.1 \mathrm{~g}, 3.2 \mathrm{mmol})$ and one iodine crystal was refluxed under nitrogen in dry o-dichlorobenzene ( 3 mL ) during 8 h . The mixture was extracted as described above and chromatographed to yield a red oil $9 \mathrm{~b}(0.14 \mathrm{~g}, 28 \%) .{ }^{1} \mathrm{H}-\mathrm{NMR}$ (DMSO$\left.d_{6}\right): \delta 0.89(\mathrm{~d}, J=6.3 \mathrm{~Hz}, 6 \mathrm{H}), 1.55(\mathrm{q}, J=7.8,6.7 \mathrm{~Hz}, 2 \mathrm{H}), 1.68(\mathrm{~m}, 1 \mathrm{H}), 3.78(\mathrm{t}, J=7.1 \mathrm{~Hz}$,
$2 \mathrm{H}), 5.90(\mathrm{~s}, 2 \mathrm{H}), 6.53(\mathrm{~s}, 1 \mathrm{H}), 6.66(\mathrm{~s}, 1 \mathrm{H}), 6.92(\mathrm{~m}, 1 \mathrm{H}), 6.98(\mathrm{~d}, J=8.2 \mathrm{~Hz}, 1 \mathrm{H}), 7.12(\mathrm{dd}, J=$ $7.6,1.5 \mathrm{~Hz}, 1 \mathrm{H}), 7.18(\mathrm{td}, J=7.7,1.5 \mathrm{~Hz}, 1 \mathrm{H}) ;{ }^{13} \mathrm{C}-\mathrm{NMR}\left(\mathrm{DMSO}_{\mathrm{d}}\right): \delta 22.66,26.31,36.18$, 46.12, $98.54,101.50,107.68,115.56,115.56,122.45,126.06,127.31,127.45,140.62,143.17$, 146.30, 147.82. Anal. Calcd. for $\mathrm{C}_{18} \mathrm{H}_{19} \mathrm{NO}_{2} \mathrm{~S}: \mathrm{C}, 68.98$; H, 6.11; N, 4.47. Found: C, 69.18; H, 6.35; N, 4.24.

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