# Internet EGEFONIG Journal of Molecular Design

October 2004, Volume 3, Number 10, Pages 622–650

Editor: Ovidiu Ivanciuc

Proceedings of the Internet Electronic Conference of Molecular Design, IECMD 2003 November 23 – December 6, 2003 Part 5

# Quantitative Structure–Activity Relationship Study of Bisphosphonates

Aihua Xie,<sup>1,2</sup> Chenzhong Liao,<sup>1,2</sup> Zhibin Li,<sup>1</sup> Zhiqiang Ning,<sup>1</sup> Weiming Hu,<sup>1</sup> Xianping Lu,<sup>1</sup> Leming Shi,<sup>1</sup> and Jiaju Zhou<sup>2</sup>

 <sup>1</sup> Chipscreen Biosciences, Ltd., Research Institute of Tsinghua University, Suite C301, Shenzhen, Guangdong 518057, China
 <sup>2</sup> Institute of Process Engineering, Chinese Academy of Sciences, P.O. Box 353, Beijing 100080, China

Received: October 10, 2003; Revised: April 5, 2004; Accepted: May 27, 2004; Published: October 31, 2004

#### Citation of the article:

A. Xie, C. Liao, Z. Li, Z. Ning, W. Hu, X. Lu, L. Shi, and J. Zhou, Quantitative Structure– Activity Relationship Study of Bisphosphonates, *Internet Electron. J. Mol. Des.* **2004**, *3*, 622–650, http://www.biochempress.com. Internet BEFFONE Journal of Molecular Design BIOCHEM Press http://www.biochempress.com

# Quantitative Structure–Activity Relationship Study of Bisphosphonates<sup>#</sup>

Aihua Xie,<sup>1,2,\*</sup> Chenzhong Liao,<sup>1,2</sup> Zhibin Li,<sup>1</sup> Zhiqiang Ning,<sup>1</sup> Weiming Hu,<sup>1</sup> Xianping Lu,<sup>1</sup> Leming Shi,<sup>1</sup> and Jiaju Zhou<sup>2</sup>

<sup>1</sup> Chipscreen Biosciences, Ltd., Research Institute of Tsinghua University, Suite C301, Shenzhen, Guangdong 518057, China

<sup>2</sup> Institute of Process Engineering, Chinese Academy of Sciences, P.O. Box 353, Beijing 100080, China

Received: October 10, 2003; Revised: April 5, 2004; Accepted: May 27, 2004; Published: October 31, 2004

Internet Electron. J. Mol. Des. 2004, 3 (10), 622–650

#### Abstract

**Motivation.** Bisphosphonates (BPs) are most widely used as agents for treating osteoporosis. They have also been used for other purposes such as herbicides, anticancer agents, and antiparasitics. Here we report QSAR models of four BPs datasets based on the 118 structural and biological data we have collected from various literature sources.

**Method.** The structures were energy minimized with the MMFF94 force field, then 181 2D and 3D descriptors were calculated with MOE (Molecular Operating Environment). The step by step multiple regression and principle analysis were used to generate QSAR models. The predictive power of each QSAR model was estimated with the leave–one–out (LOO) cross–validation method.

**Results.** The QSAR model for a dataset of 28 GGPPSase inhibitors ( $r^2 = 0.86$ ,  $r^2_{LOO} = 0.82$ , s = 0.45,  $s_{LOO} = 0.51$ , F = 77.56) is made up of two descriptors. Another dataset of 28 compounds with bioactivities against the growth of *T. Brucei rhodesiense* was studied using PCA and reached a model ( $r^2 = 0.85$ ,  $r^2_{LOO} = 0.79$ , s = 0.30,  $s_{LOO} = 0.35$ , F = 32.03) with four principle components (PCs). Both the above models have comparable predictive ability with CoMFA model reported by Szabo *et al.* The 86 BPs provided by Novartis with *in vivo* bio–data of TPTX rats were divided into two datasets. A six PCs model ( $r^2 = 0.80$ ,  $r^2_{LOO} = 0.72$ , s = 0.44,  $s_{LOO} = 0.53$ , F = 24.83) elucidated the dataset of 44 compounds in which containing aliphatic linked nitrogen atoms. The other dataset includes 42 BPs containing a heterocyclic moiety with at least one nitrogen atom. Its PCA model ( $r^2 = 0.80$ ,  $r^2_{LOO} = 0.71$ , s = 0.46,  $s_{LOO} = 0.57$ , F = 19.99) consists of seven PCs.

**Conclusions.** A leave–four–out test procedure shows that though the QSAR models based on *in vivo* bone resorption  $pED_{50}$  values cannot provide explicit indications for drug design, their predictive ability for related compounds is quite good.

**Keywords.** Bisphosphonate; principal component analysis (PCA); GGPPSase inhibitor; quantitative structureactivity relationships (QSAR); IC<sub>50</sub>; ED<sub>50</sub>.

Abbreviations and notations	
Bps, bisphosphonates	$r_{\rm LOO}^2$ , leave–one–out cross validated correlation
ED <sub>50</sub> , the dose of compound administratered sc, which	
results in a 50% reduction of the hypercalcemia	

<sup>&</sup>lt;sup>#</sup> Presented in part at the Internet Electronic Conference of Molecular Design, IECMD 2003.

<sup>\*</sup> Correspondence author; phone: 86–10–82612204; fax: 86–10–62561822; E-mail: ahxie@home.ipe.ac.cn.

induced in TPTX rats by 1,25–dihydroxyvitamin D <sub>3</sub> FPPSase, farnesyl pyrophosphate synthase GGPPSase, geranylgeranyl diphosphate synthase	coefficient. $r_{LOO}^{2} = 1 - \frac{\sum (y_{pred} - y_{act})^{2}}{\sum (y_{act} - y_{mean})^{2}}$
IC <sub>50</sub> , experimental concentration required to reduce activity/proliferation of enzymes/cells/parasites by 50% PCA, principal component analysis PRED, leave-one-out cross validated prediction QSAR, quantitative structure-activity relationships $r^2$ , correlation coefficient. $r^2 = 1 - \frac{\sum (y_{calc} - y_{act})^2}{\sum (y_{act} - y_{mean})^2}$	<i>s</i> , root mean square error. $s = \sqrt{\frac{\sum (y_{calc} - y_{act})^2}{n-k-1}}$ , <i>n</i> , number of observations; <i>k</i> , number of descriptors TPTX, thyroparathyroidectomy $s_{LOO}$ , leave–one–out cross validated correlation root mean square error. $s_{LOO} = \sqrt{\frac{\sum (y_{pred} - y_{act})^2}{n-k-1}}$ , <i>n</i> , number of objects; <i>k</i> , number of descriptors

#### **1 INTRODUCTION**

Bisphosphonates (BPs) are the most widely used inhibitors of bone resorption. They all contain two phosphonate groups attached to a single carbon atom, forming a P–C–P structure. Bisphosphonates are stable analogs of naturally occurring pyrophosphate–containing compounds, which now helps to explain their intracellular as well as their extracelluar modes of action. Several bisphosphonates, *e.g.*, etidronate, clodronate, pamidronate, alendronate, tiludronate, risedronate, and ibandronate, have been established as effective treatments in clinical disorders such as Paget's disease of bone, tumour–associated bone disease, and osteoporosis [1]. Bisphosphonates have also been repeated for uses as herbcides [2], anticancer agents [3], and antiparasitics [4,5].

Recent studies suggest that bisphosphonates inhibit bone resorption by cellular effects on bone– resorbing osteoclasts, rather than by purely physicochemical mechanisms. It is likely that BPs are internalized by osteoclasts and interfere with specific biochemical process and induce apoptosis [6]. In recent work, the site of action has been narrowed down to the mevalonate pathway and the isoprene pathway.

The exact enzymes of the mevalonate pathway that are inhibited by BPs have not yet been fully identified. However, incadronate and ibandronate are known inhibitors of squalene synthase, an enzyme in the mevalonate pathway required for cholesterol biosynthesis [6]. Alendronate and pamidronate are less potent inhibitors of squalene synthase but can also inhibit sterol biosynthesis, suggesting that these bisphosphonates may inhibit up stream enzymes of the mevalonate pathway other than squalene synthase [7].

Several enzymes of the mevalonate pathway such as isoprenoid diphosphate isomerase (IPP isomerase), farnesyl diphosphate synthase (FPPSase), geranylgeranyl diphosphate synthase (GGPPSase), and squalene synthase, utilize an isoprenoid diphosphate as a substrate and thus are likely to have similar substrate binding sites. Thus if nitrogen–containing BPs act as substrate analogs of an isoprenoid diphosphate, it is likely that these BPs actually inhibit several enzymes of the mevalonate pathway. FPPSase are the most reported target for many BPs. For example,

Cromartie and Fisher demonstrated that herbicidal bisphosphonates were potent, low-nanomolar inhibitors of a daffodil FPPSase [2,8], and Grove *et al.* reported that BPs were growth and FPPSase inhibitors of the primitive eukaryote *Dictyostelium discoideum* [9].

Several groups [1,3,10,11] have reported that FPPSase was the target of the nitrogen–containing bisphosphonates in bone, leading to the apoptosis of osteoclasts. The group of Eric Oldfield, which did a lot of jobs on chemotherapy of parasitic protozoa diseases, reported that bisphosphonates were in vitro inhibitors of the growth of the causative agents of Chagas' disease, human East African trypanosomiasis, visceral leishmaniasis, toxoplasmosis, malaria, and cryptosporidiosis, *T. Cruzi, Trypanosoma brucei rhodesiense, Leishmania donovani, Toxoplasma gondii, Plasmodium falciparum, and Cryptosporidium parvum* [4,5]. They also showed that in some of the parasites, such as *T. b. rhodesiense* and *D. discoideum*, the molecular target of some bisphosphonate drugs, in the inhibition of bone resorption as well as the growth of *D. discoideum* and the bloodstream form of *T. b. rhodesiense* [5,12]. Though there are fewer reports, other enzymes, *e.g.* IPP isomerase, GGPP synthase, and squalene synthase of the mevalonate pathway, may be also potential targets for different bisphosphonates.

Eric Oldfield group has investigated the inhibition of a human recombinant GGPPSase by 23 bisphosphonates and six azaptenyl diphosphates. In addition to CoMFA analysis of structure–activity relationship, the pharmacophore of these GGPPSase inhibitors obtained from Catalyst was also provided [13].

Though the actual conformations of the bisphosphonates in the FPPSase and GGPPSase active sites are not yet known, good predictive CoMFA models were obtained using the molecular mechanics-derived lowest-energy conformers [5,12,13].

Widler *et al.* reported an extensive structure–activity relationship (SAR) study of bisphosphonates [14]. Small changes of the structure of pamidronate (compound 2) lead to marked improvements of the inhibition of osteoclastic resorption potency. Alendronate (compound 3 in Table 1), with an extra methylene group in the *N*–alkyl chain, and olpadronate (compound 7), the *N*,*N*–dimethyl analogue, are about 10 times more potent than pamidronate (compound 2).

Extending one of the *N*-methyl groups of olpadronate to a pentyl substituent leads to ibandronate (compound 10), which is the most potent close analogue of pamidronate. Even slightly better antiresorptive potency is achieved with derivatives having a phenyl group linked via a short aliphatic tether of three to four atoms to nitrogen, the second substituent being preferentially a methyl group. The most potent bisphosphonate, zoledronate (compound 65), is found in the series containing a heteroaromatic moiety with at least one nitrogen atom, which is linked via a single methylene group to the geminal bisphosphonate unit [14].

Quantitative Structure-Activity Relationship Study of Bisphosphonates
Internet Electronic Journal of Molecular Design 2004, 3, 622–650

	Table 1. Stucture and bioactivity of bisphosphonates									
$R1$ $PO_3H_2$										
			I	N—( R2	сн <sub>2</sub> )п— F	<sup>—</sup> ∧ ⁰О <sub>3</sub> Н₂				
No	Cmpd code	R <sub>1</sub>	R <sub>2</sub>	Х	n	$ED_{50} (\mu g/kg)^a$	$IC_{50}(\mu M)^{b}$	IC <sub>50</sub> (µM) <sup>c</sup>		
1	Novartis 1a	Н	Н	OH	1	150				
2	Pamidronate	Н	Н	OH	2	61	177	180		
3	Alendronate	Н	Н	OH	3	8		440		
4	Novartis 1d	Н	Н	OH	4	20				
5	Neridronate	Н	Н	OH	5	60	31.7	690		
6	Novartis 1g	Me	Н	OH	2	15				
7	Olpadronate	Me	Me	OH	2	12	5.4			
8	T.B. 009	propyl	Me	OH	2	3	7.8	330		
9	Novartis 1j	Et	Et	OH	2	3				
10	Ibandronate	pentyl	Me	OH	2	1.1	0.96	83		
11	Novartis 11	Me	Me	Η	2	100				
				R1		O <sub>3</sub> H <sub>2</sub>				
					R P	— ОН О <sub>2</sub> Н <sub>2</sub>				
No	Cmpd code	R <sub>1</sub>	R	2	R	$\frac{ED_{50} (\mu g / kg)^a}{ED_{50} (\mu g / kg)^a}$	$IC_{50}(\mu M)^{b}$	IC <sub>50</sub> (µM) <sup>c</sup>		
12	Novartis 1n	Н	H	Ι	Me	3.4				
13	Novartis 10	Me	Ν	ſe	Me	18				
14	Novartis 1p	pentyl	Ν	/le	Me	65				
			(	$\frown$						
			(		(CH <sub>2</sub> )n—					
N	Cross	11.	(			D		( /1 )8		
NO	Cmpo	a code		_N		K	n ED	<sub>50</sub> (µg /kg)"		
15	Nova	rtis 2a		N-	.}	Н	2	10		
			R	$\sim$	<					
16	Nova	rtis 2b				Н	3	25		
17	Nova	rtis 2c				Н	5	250		
18	Nova	rtis 2d				Ph	2	70		
19	Nova	rtis 2e				4ClPh	2	3.5		
20	Nova	rtis 2f	R—	N-		Н	2	5.6		
21	Nova	rtis 2a				Ph	2	11		
21	Nova	rtis 2g				Ph	2	100		
22	Nova	artis 211				3_F_Ph	2	30		
23	11072	u tis 2j				5-1-111	2	50		
24	Nova	rtis 2k	(	) N—			2	25		
				<u> </u>						
25	Nova	rtis 2m	R-N	_N—		Me	2	400		
			```							

$PO_{i}H_{2}$ SerialCmpd codeImage: PO_{i}H_{2}ED <sub>50</sub> (µg /kg) <sup>3</sup> 26Novartis 3aImage: PO_{i}H_{2}5027Novartis 3bImage: PO_{i}H_{2}25028Novartis 3cHm Image: PO_{i}H_{2}2500Rt Novartis 3cHm Image: PO_{i}H_{2}SerialCmpd codeR1R2R3ED <sub>50</sub> (µg /kg) <sup>8</sup> 29Novartis 4aGreat H30030Novartis 4aMeR3R3R3R3R3R3R3Novartis 4aMeH1014R3R3ED <sub>50</sub> (µg /kg) <sup>8</sup> 2500R1R2R3ED <sub>50</sub> (µg /kg) <sup>8</sup> 29Novartis 4aMe1430Novartis 4aMe14143030Novartis 4iMe1433 <td< th=""></td<>								
NH PO <sub>1</sub> H2SerialCmpd codeNHED50 (µg /kg) <sup>8</sup> 26Novartis 3a $\int_{-}^{H}$ 5027Novartis 3b $\int_{-}^{H}$ 25028Novartis 3c $Hh$ 250R1 Novartis 3cR2 N-CHCH2H2 OHR2 N-CHCH2H2 OHPO <sub>2</sub> H2SerialCmpd codeR1R2R3ED50 (µg /kg) <sup>8</sup> 29Novartis 4a $\int_{-}^{+}$ 30Novartis 4aMe31Novartis 4cR3R3Novartis 4cEtR4H1534Novartis 4fMe35Novartis 4f36Novartis 4i37Novartis 4i38Novartis 4l39Novartis 4l								
SerialCmpd code $H_{H_{H_{H_{H_{H_{H_{H_{H_{H_{H_{H_{H_{H$								
26       Novartis 3a $\int_{h}^{h}$ 50         27       Novartis 3b $\int_{h}^{h}$ 250         28       Novartis 3c $Hh$ 250         R1 Novartis 3c $Hh$ 250         Serial       Cmpd code       R1       R2       R3       ED <sub>50</sub> (µg /kg) <sup>4</sup> 29       Novartis 4a $\int_{h}^{h}$ H       300         30       Novartis 4a $\int_{h}^{h}$ H       H       20         31       Novartis 4c       R3       Me       H       1         32       Novartis 4c       R4       H       15         34       Novartis 4f       Me       3-Me       1.5         35       Novartis 4g       Me       4-C1       0.7         36       Novartis 4i $\int_{h}^{h}$ Me       0.4         38       Novartis 4k       Me       20       20         39       Novartis 41       Me       1500								
27       Novartis 3b $H_{1}$ 250         28       Novartis 3c $H_{1}$ 2500         R1 N-CHUCH2-DH H       2500         R1 N-CHUCH2-DH H       2500         Serial       Cmpd code       R1       R2       R3       ED <sub>50</sub> (µg /kg) <sup>a</sup> 29       Novartis 4a       Me       1.4         30       Novartis 4b       Me       1.4         31       Novartis 4b       Me       1.4         31       Novartis 4d       Me       1.4         31       Novartis 4d       Me       1.4         32       Novartis 4d       Me       1.4         33 <th< th=""></th<>								
28       Novartis 3c       HN       2500 $R_1^{PO_3H_2}_{PO_3H_2}$ $PO_3H_2^{PO_3H_2}_{PO_3H_2}$ $PO_3H_2^{PO_3H_2}_{PO_3H_2}$ Serial       Cmpd code $R_1$ $R_2$ $R_3$ $ED_{50}(\mu g / kg)^a$ 29       Novartis 4a $\mu$ H       300         30       Novartis 4a       Me       1.4         31       Novartis 4c $\mu$ H       H       20         32       Novartis 4d       Me       H       1       5         33       Novartis 4d       Me       H       1.5         34       Novartis 4g       Me       4-C1       0.7         36       Novartis 4g       Me       0.4       0.4         38       Novartis 4k       Me       20         39       Novartis 41       Me       20								
R1 R2 $PO_3H_2$ $PO_3H_2$ SerialCmpd codeR1R2R3 $ED_{50} (\mu g / kg)^a$ 29Novartis 4a $f$ H30030Novartis 4bMe1.431Novartis 4cMeH132Novartis 4cMeH133Novartis 4cMeH134Novartis 4dMeH1.535Novartis 4fMe3-Me1.536Novartis 4jMe0.437Novartis 4kMe0.438Novartis 4lMe1500								
SerialCmpd code $R_1$ $R_2$ $R_3$ $ED_{s0}(\mu g / kg)^a$ 29Novartis 4aIH30030Novartis 4bMe1.431Novartis 4cIH2032Novartis 4dMeH133Novartis 4dMeH134Novartis 4fMe3-Me1.535Novartis 4gMe4-Cl0.736Novartis 4jMe0.438Novartis 4kMe2039Novartis 4lMe1500								
29Novartis 4a $\checkmark$ H30030Novartis 4bMe1.431Novartis 4c $\checkmark$ HH2032Novartis 4dMeH133Novartis 4dMeH1534Novartis 4fMe3-Me1.535Novartis 4gMe4-Cl0.736Novartis 4i $\checkmark$ H1.037Novartis 4jMe0.438Novartis 4k $\checkmark$ Me2039Novartis 4l $\checkmark$ Me1500								
30Novartis 4bMe $1.4$ $31$ Novartis 4c $4$ HH $20$ $32$ Novartis 4dMeH1 $33$ Novartis 4dMeH1 $34$ Novartis 4fMe $3-Me$ $1.5$ $35$ Novartis 4gMe $4-C1$ $0.7$ $36$ Novartis 4i $1$ Me $0.4$ $37$ Novartis 4jMe $0.4$ $38$ Novartis 4k $1$ Me $20$ $39$ Novartis 41 $1$ $1500$								
31Novartis 4c $H$ H2032Novartis 4dMeH133Novartis 4eEtH1534Novartis 4fMe3-Me1.535Novartis 4gMe4-Cl0.736Novartis 4i $H$ 1.037Novartis 4jMe0.438Novartis 4k $Me$ 2039Novartis 4l $Me$ 1500								
32Novartis 4dMeH133Novartis 4eEtH1534Novartis 4fMe3-Me1.535Novartis 4gMe4-Cl0.736Novartis 4i $\checkmark$ H1.037Novartis 4jMe0.438Novartis 4k $\checkmark$ Me2039Novartis 4l $\checkmark$ Me1500								
36       Novartis 4i       H       1.0         37       Novartis 4j       Me       0.4         38       Novartis 4k       Me       20         39       Novartis 4l       Me       1500								
37         Novartis 4j         Me         0.4           38         Novartis 4k         Me         20           39         Novartis 4l         Me         1500								
38         Novartis 4k         Me         20           39         Novartis 4l         Me         1500								
<b>39</b> Novartis 41 Me 1500								
$R2 \xrightarrow{PO_{3}H_{2}} X - (CH_{2})m - N - (CH_{2})n - OH \\ R1 \qquad PO_{3}H_{2}$								
Serial Cmpd code X $R_1$ $R_2$ m n $ED_{50} (\mu g / kg)^a$								
40         Novartis 5a         O         Me         H         2         2         1.5								
<b>41</b> Novartis 5b O Me 4–Cl 2 2 1.7								
<b>42</b> Novartis 5c O H H 3 2 1.2								
$\begin{array}{cccccccccccccccccccccccccccccccccccc$								
$\begin{array}{cccccccccccccccccccccccccccccccccccc$								
<b>45</b> Novatus 31 U Me $4-r$ 5 2 0.0 <b>46</b> Novartis 5 $\sigma$ O Me $4-Cl$ 2 2 1.2								
<b>47</b> Novatis 5h $O$ Ft $4-CI$ 3 2 1.5								
$48 \qquad \text{Novartis 5i} \qquad O \qquad \text{propvl} \qquad H \qquad 3 \qquad 2 \qquad 20$								
<b>49</b> Novartis 5j O butyl H 3 2 10								
50         Novartis 5k         O         Me         H         4         2         500								

Quantitative Structure–Activity Relationship Study of Bisphosphonates Internet Electronic Journal of Molecular Design **2004**, *3*, 622–650

Table 1. (Continued)												
Serial	Cmpd code	Х	$R_1$	I	$R_2$	m	n	$ED_{50} \left(\mu g / kg\right)^{a}$				
51	Novartis 51	0	Me	I	Η	6	2	4				
52	Novartis5m	0	Me	I	H	3	2	7500				
53	Novartis 5n	0	Me	I	H	2	3	100				
54	Novartis 5p	S	Me	I	Η	2	2	0.7				
55	Novartis 5q	S	Н	I	Η	3	2	7				
56	Novartis 5r	S	Me	1	H	3	2	0.33				
57	Novartis 5s	S	Me		I-Cl	3	2	7.8				
$Het - (CH_2)n - OH$												
	PO <sub>3</sub> H <sub>2</sub>											
Serial	Cmpd code	Het		R1	R2	R3	n	ED50 $(\mu g / kg)^a$				
58	Novartis 6a	$\left[ \begin{array}{c} N \\ N \\ N \\ R1 \end{array} \right]$		Н			1	5				
59	Novartis 6b			Me			1	0.6				
60	Novartis 6c			Bz			1	25				
61	Novartis 6d		-}	Н	Н		1	0.3				
62	Novartis 6e			Н	Н		2	20				
63	Novartis 6f			Me	Н		1	15				
64	Novartis 6h			Н	Me		1	1.5				
65	Zoledronate	R3 N N R1	_}	Н	Н	Н	1	0.07				
66	Novartis 6j			Н	Н	Н	2	45				
67	Novartis 6k			Me	Н	Н	1	3				
68	Novartis 61			Н	Me	Me	1	1.5				
69	Novartis 6n						1	600				
			R1 P N-+ R2 P	O <sub>3</sub> H <sub>2</sub> —H O <sub>3</sub> H <sub>2</sub>								
Serial	Cmpd code	$R_1R_2N$				EI	D <sub>50</sub> (μg	/kg) <sup>a</sup>				
70	Novartis 7c		$\bigcirc$	n—{		80	0					
71	Novartis 7d		$\bigcirc$	»_{		40						
72	Novartis 7e		$\bigcirc$	)-N, H		7						

Het $H_{PO_3H_2}$ SerialCmpd codeHet $R_1$ $R_2$ EDs73Novartis 8a $\stackrel{R^2}{\underset{R1}{\longrightarrow}}$ HH574Novartis 8bHMe10075Novartis 8cMeH1.576Novartis 8dEtH1.577Novartis 8dBuH0.978Novartis 8fBuH0.979Novartis 8gPrH20080Novartis 8hPhCH_2CH_2H2.781Novartis 8j $\stackrel{N}{\underset{R1}{\longrightarrow}}$ H50082Novartis 8kMe583Novartis 8hPhCH_27584Novartis 8mPh200SerialCmpd codeXRED <sub>50</sub> (µg85Novartis 9aCH2OH20086Novartis 9bSOH700	<sub>50</sub> (μg /kg) <sup>a</sup>
Het $H_{PO_3H_2}$ SerialCmpd codeHet $R_1$ $R_2$ ED <sub>3</sub> 73Novartis 8a $R^2                                     $	<sub>50</sub> (μg /kg) <sup>a</sup>
FO3F2SerialCmpd codeHetR1R2ED373Novartis 8a $\stackrel{R2}{\hspace{1em}\hspace{1em}}, \stackrel{N}{\hspace{1em}\hspace{1em}}, \stackrel{N}{\hspace{1em}\hspace{1em}}, \stackrel{N}{\hspace{1em}}, \stackrel{N}{\hspace{1em}\hspace{1em}}, \stackrel{N}{\hspace{1em}\hspace{1em}}, \stackrel{N}{\hspace{1em}}, \stackrel{N}{$	<sub>50</sub> (μg /kg) <sup>a</sup>
SerialCmpd codeHet $R_1$ $R_2$ ED373Novartis 8a $R^2 \downarrow N \downarrow$ HH574Novartis 8bHMe10075Novartis 8cMeH1.576Novartis 8dEtH1.577Novartis 8cPrH20078Novartis 8fBuH0.979Novartis 8gPrH2.781Novartis 8j $\bigvee_{R1}$ H50082Novartis 8jPhCH27584Novartis 8lPh200SerialCmpd codeXRED30 (µg85Novartis 9aCH2OH86Novartis 9bSOH	<sub>50</sub> (μg /kg) <sup>a</sup>
73Novartis 8a $R^2 \rightarrow N$ $R_1 \rightarrow S$ HH574Novartis 8bHMe10075Novartis 8cMeH1.576Novartis 8dEtH1.577Novartis 8dPrH20078Novartis 8fBuH0.979Novartis 8gPrH20080Novartis 8hPhCH2CH2H2.781Novartis 8j $\bigwedge_{R1}$ H50082Novartis 8kMe583Novartis 8lPhCH27584Novartis 8mPh200 $\checkmark_{R1}^{P-OH}$ SerialCmpd codeXRED <sub>50</sub> (µg85Novartis 9aCH2OH20086Novartis 9bS $OH$ 700 $OH$	
74Novartis 8bHMe10075Novartis 8cMeH1.576Novartis 8dEtH1.577Novartis 8dPrH2.978Novartis 8fBuH0.979Novartis 8gPrH20080Novartis 8hPhCH2CH2H2.781Novartis 8j $\bigwedge_{R1}^{N}$ H50082Novartis 8jPhCH27584Novartis 8hPhCH27584Novartis 8mPh200SerialCmpd codeXRED <sub>50</sub> (µg85Novartis 9aCH2OH86Novartis 9bSOH	
74Novartis 80HH1.075Novartis 8cH1.576Novartis 8dEtH1.577Novartis 8ePrH278Novartis 8fBuH0.979Novartis 8gPrH20080Novartis 8hPhCH <sub>2</sub> CH <sub>2</sub> H2.781Novartis 8j $\bigwedge_{R1}$ H50082Novartis 8kMe583Novartis 8lPhCH <sub>2</sub> 7584Novartis 8mPh200SerialCmpd codeXRSerialCmpd codeXRED <sub>50</sub> (µg85Novartis 9aCH <sub>2</sub> OH20086Novartis 9bSOH700	1 1 1
ToNovartis 8dHHH76Novartis 8dEtH1.577Novartis 8ePrH278Novartis 8fBuH0.979Novartis 8gPrH20080Novartis 8hPhCH2CH2H2.781Novartis 8j $\bigwedge_{R1}$ H50082Novartis 8j $\bigwedge_{R1}$ H50083Novartis 8lPhCH27584Novartis 8lPh200 $\bigvee_{R}$ SerialCmpd codeXRED <sub>30</sub> (µg85Novartis 9aCH286Novartis 9bS0HTo	
77Novartis 8 kPrH278Novartis 8 fBuH0.979Novartis 8 gPrH20080Novartis 8 hPhCH2CH2H2.781Novartis 8 j $\bigwedge_{R1}$ H50082Novartis 8 kMe583Novartis 8 hPhCH27584Novartis 8 mPh200 $\bigvee_{R1}^{S_{P1}}$ SerialCmpd codeXRED <sub>50</sub> (µg85Novartis 9aCH2OH86Novartis 9bSOH	
78Novartis 8fBuH0.979Novartis 8gPrH20080Novartis 8hPhCH2CH2H2.781Novartis 8j $\bigwedge_{R1}$ H50082Novartis 8kMe583Novartis 8lPhCH27584Novartis 8mPh200SerialCmpd codeXRSerialCmpd codeXRED <sub>50</sub> (µg85Novartis 9aCH2OH20086Novartis 9bSOH700	
79Novartis 8gPrH20080Novartis 8hPhCH2CH2H2.781Novartis 8j $\bigwedge_{R1}$ H50082Novartis 8kMe583Novartis 8lPhCH27584Novartis 8mPh200 $\bigvee_{R1}$ $\bigvee_{R1}$ SerialCmpd codeXRED <sub>50</sub> (µgSerialCmpd codeXRED <sub>50</sub> (µg85Novartis 9aCH2OH20086Novartis 9bSOH700	
80Novartis 8hPhCH2CH2H2.781Novartis 8j $\bigwedge_{R1}$ H50082Novartis 8kMe583Novartis 8lPhCH27584Novartis 8mPh200 $\bigvee_{R}$ SerialCmpd codeXRED <sub>50</sub> (µg85Novartis 9aCH2OH200Novartis 9bSOH200	1
81Novartis 8j $\bigwedge_{R1}^{N}$ H50082Novartis 8kMe583Novartis 8lPhCH27584Novartis 8mPh200 $\qquad \qquad $	1
82Novartis 8kMe583Novartis 8lPh $CH_2$ 7584Novartis 8mPh200 $\begin{tabular}{lllllllllllllllllllllllllllllllllll$	
83Novartis 81PhCH27584Novartis 8mPh200 $S \rightarrow P \rightarrow OH$ S $P \rightarrow OH$ Ph200S $OH$ 700S $OH$ 700	1
84Novartis 8mPh200 $\stackrel{\end{black}}{\label{eq:spectral_product}}$ $\stackrel{\end{black}}{\label{eq:spectral_product}}$ $\stackrel{\end{black}}{\end{black}}$ SerialCmpd codeXRED <sub>50</sub> (µg85OHNovartis 9aCH2OH200SerialCmpd codeXRED <sub>50</sub> (µg85Novartis 9aCH2OH20086Novartis 9bSOH700	
$\begin{array}{c c} & & & & \\ & & & \\ & & & \\ Serial & Cmpd code & X & R & ED_{50} (\mu g \\ \hline \textbf{85} & Novartis 9a & CH_2 & OH & 200 \\ \textbf{86} & Novartis 9b & S & OH & 700 \\ \hline \end{array}$	
SerialCmpd codeXR $ED_{50}$ ( $\mu g$ 85Novartis 9aCH2OH20086Novartis 9bSOH700	<u>, u ) a</u>
85         Novartis 9a         CH2         OH         200           86         Novartis 9b         S         OH         700	/kg)"
80 Novarus 90 S OH 700	
$R \xrightarrow{HO, P < O, P < O} O \xrightarrow{T > OH}$	
Serial Cmpd code R n	$IC_{50} (\mu M)^c$
87 3-azaGGPP 2	0.14
88 3–azaFPP 2	0.74
<b>89</b> 3–azaGPP Me 2	240
90 3–azahomoGGPP 3	0.37
91 3-azahomoFPP 3	0.31
92 15-azaGGPP $N = 15 + azaGGPP$	>>100
О, ОН	
Serial Cmpd code n $IC_{50} (\mu M)^b$ $IC_{50} (\mu M)^b$	ιM) <sup>c</sup>
<b>93</b> T.B. 024 2 92.0 620	/
<b>94</b> T.B. 025 3 99.8 200	
<b>95</b> T.B. 023 4 62.4 53	
<b>96</b> GGPP018 5 11.0	

**ВюСнем** Press

Quantitative Structure–Activity Relationship Study of Bisphosphonates Internet Electronic Journal of Molecular Design **2004**, *3*, 622–650

	Table 1. (Continued)									
Serial	Cmpd code	n	$IC_{50} (\mu M)^{b}$	$IC_{50} (\mu M)^{c}$						
97	GGPP017	6		4.3						
98	T.B. 014	8	20.5	0.72						
99	T.B. 010	9	8.0	1.4						
100	T.B. 007	10	2.0	0.92						
				L						
Serial	Cmpd code	Struct	ure	$IC_{50} (\mu M)^{0}$	$IC_{50} (\mu M)^{c}$					
101	T.B. 006	$\sim$	Ч С Р ОН 0=Р ОН ОН	1.7	2.2					
102	GGPP031	$\sim$	0 он → Р он 0 ₽ он 0 ₽ он он		19.0					
103	T.B. 021	$\sim$	н О, ОН - - - - - - - - - - - - -	50.6						
104	T. B. 026		0, он <sup>Р</sup> он <sup>2</sup> Р∼он он	102						
105	T.B. 016		Р ОН ОН ОН	21.3	220.0					
106	Т.В. 020	H <sub>2</sub> N		40.0	180.0					
107	T.B. 012	(N) H	окрании Страни он он он	8.6	220.0					
108	NE97220		н Окраника И У Рон 0 - Р Он 0 - Он	0.7	220.0					
109	N-(2-(4-picolyl))ADMP	Ţ	н органия N органия N органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органия Органи Орсан Орсания Орсан Орсан Орсан Орсан Орсан Орсан Орсан Орсан Ор	0.61	260.0					
110	T.B. 013	N	<sup>О</sup> , <sub>Р</sub> OH s HOH o <sup>Р</sup> OH O OH O OH	19.8	550.0					
111	Homorisedronate		O P OH OH OH	1.7	410.0					
112	Risedronate		окрании	8.6	350.0					

Table 1. (Continued)								
Serial	Cmpd code	Structure	$IC_{50} (\mu M)^{b}$	$IC_{50} (\mu M)^{c}$				
113	NE58018	N OH OH OH OH	0.22					
114	N–(2–(5–chloro)– pyridyl)AMDP		53.3					
115	T.B. 015		20.9					
116	T.B. 018	N N O O O H O H	34.4					
117	T.B. 019	Br N O OH OH OH	39.5					
118	T.B. 2–13	$H \xrightarrow{N} V \xrightarrow{O} C-CH_2 \cdot CH_3$ $-CH_2 \cdot CH_3$ $-CH_2 \cdot CH_3$ $-CH_2 \cdot CH_3$ $-CH_2 \cdot CH_3$ $-CH_2 \cdot CH_3$ $-CH_2 \cdot CH_3$	27.9					

 $^{a}$  the dose of compound administratered sc, which results in a 50% reduction of the hypercalcemia induced in TPTX rats by 1,25–dihydroxyvitamin D<sub>3</sub> [14]

<sup>b</sup> experimental concentration required to reduce proliferation of *T. Brucei rhodesiense* by 50% [4,5]

<sup>c</sup> experimental concentration required to reduce activity of GGPPSase by 50% [13].

The comprehension of BPs mechanism gives us indications to investigate the quantitative structure–activity relationship of the bisphosphonates provided by Novartis [14] with *in vivo* ED<sub>50</sub> against hypercalcemia induced in thyroparathyroidectomy (TPTX) rats. The total of 86 compounds were divided into two datasets, one for the series containing a heterocyclic moiety, which contains at least one nitrogen atom; the other for bisphosphonates that contains a nitrogen atom in aliphatic link and do not possess a heterocyclic substitute. The two datasets were analyzed using QSAR module of MOE and achieved two predictive models through principal component analysis.

We also investigated the BPs with  $IC_{50}$  against *T. Brucei Trypomastigotes* [4,5] and BPs with  $IC_{50}$  for GGPPSase inhibition [13] using the molecular modeling package MOE respectively, and achieved more simple and lightening models.

#### 2 MATERIALS AND METHODS

#### 2.1 Chemical Data

The structures we have collected here are listed in the Table 1 along with compound code and bioactivities. Some of the compound codes were assigned following their traditional name such as

Alendronate and Pamidronate, or codes from original references such as NE58018 and NE97220. The others were assigned according original activity source such as T.B.006 and GGPP031, or data provider such as Novartis1a and Novartis1d.

Dataset1 are made up of 28 BPs with  $IC_{50}$  values against GGPPSase. The library covers many diverse structural features: ionic bisphosphonate and diphosphate groups; alkyl, alkenyl (prenyl), aryl, and heteroaryl side chains; 1–OH– and 1–H–bearing bisphosphonates; and nitrogen– containing or nitrogen–free side chains, together with different location of the side chain nitrogens. The pIC<sub>50</sub> values of this dataset vary from 3.16 to 6.85, with a mean value of 4.49 and a SD of 1.23. The distribution of activity of this dataset is shown in Figure 1.



**Figure 1.** pIC<sub>50</sub> distribution of dataset1.

Dataset2 include 28 bisphosphonates and their  $IC_{50}$  values against the growth of *T. Brucei rhodesiense* that is one of the causative agents of human African trypanosomiasis (sleeping sickness) [5]. The FPPSase is considered at least the main target of nitrogen–containing bisphosphonates in *T. Brucei rhodesiense* [4,5]. The pIC<sub>50</sub> values of this dataset vary from 3.75 to 6.66; with a mean value of 4.91 and a SD of 0.79. The distribution of activity of this dataset is shown in Figure 2.

The structures and activity data of dataset3 and dataset4 are both from Novartis pharma research [14]. The  $ED_{50}$  values in the two datasets are the doses of compound administratered sc, which results in a 50% reduction of the hypercalcemia induced in TPTX rats by 1,25–dihydroxyvitamin  $D_3$ .



Figure 2. pIC<sub>50</sub> distribution of dataset2.

We have known from the Introduction that the *in vivo* effect of bisphosphonates involves several enzymes of the mevalonate pathway e.g. IPP isomerase, FPP synthase, GGPP synthase, and squalene synthase. Therefore, the total of 86 compounds from Novartis were divided into two datasets according their structural features and rough speculation on their mode of action.

Dataset3 includes 44 bisphosphonates that contain a nitrogen atom in aliphatic link and do not possess nitrogen–containing heterocyclic substitutes. These compounds are less potent inhibitors of FPPSase and are speculated to act mainly with GGPPSase. The pED<sub>50</sub> values of this dataset vary from 5.12 to 9.48, with a mean value of 8.13 and a SD of 0.98. The distribution of activity of this dataset is shown in Figure 3.



**Figure 3.** pED<sub>50</sub> distributions of dataset3.

Dataset4 includes 42 bisphosphonates containing a heterocyclic moiety, which contains at least one nitrogen atom. Some of these BPs are more potent antiresorptive agents in the *in vivo* experiment and more potent FPPSase inhibitors *in vitro*. The pED<sub>50</sub> values of this dataset vary from 5.60 to 10.16; with a mean value of 7.66 and a SD of 1.06. The distribution of activity of this dataset is shown in Figure 4.



Figure 4. pED<sub>50</sub> distributions of dataset4

The  $IC_{50}$  or  $ED_{50}$  values and the respective negative logarithm (pIC<sub>50</sub> or pED<sub>50</sub>) for all compounds are listed in the tables of supplementary materials along with model predictions. The stronger inhibitor a compound is, the greater the pIC<sub>50</sub> or pED<sub>50</sub> is.

#### **2.2 Previous QSAR Models**

Using the comparative molecular field analysis (CoMFA), Szabo *et al.* [13] obtained a fairly good model for dataset1:

$$pIC_{50} = 3.55 + 0.44 \times "CH_3/473" - 0.37 \times "CH_3/330" - 0.01 \times "H^+/318"$$

$$n = 28, \quad r^2 = 0.938, \quad r^2_{LOO} = 0.900, \quad F = 86.6$$
(1)

Martin et al. [5] obtained another CoMFA model for dataset2:

$$pIC_{50} = 4.60 - 0.06" H^{+}/187" + 0.02" H^{+}/237" + 0.05" H^{+}/186" + 0.01" H^{+}/185"$$

$$n = 26, \quad r^{2} = 0.87, \quad r_{LOO}^{2} = 0.79, \quad F = 34.80$$
(2)

where  $H^+/i$  represents the interaction energy between a proton probe and the molecule at the gridpoint *i*, and CH<sub>3</sub>/*j* represents the interaction energy between a methyl probe and the molecule at the grid point *j*.

## 2.3 Molecular Modeling

The structures and biological activity data were stored in an ISIS/Base database from which an SD file was exported. The SD file was imported into a molecular modeling package (MOE) for subsequent calculations. The molecular structures were optimized using MMFF94 force field. All the 181 2D and inner 3D descriptors available in MOE [15] were calculated for every molecule. The QuaSAR–Contingency module was used to prune the descriptors in order to select an optimum subset for QSAR. The Qua–cluster module of MOE was used to evaluate the diversity of the collection of our molecules based on the table of selected molecular descriptors and assigned weights to molecules if necessary. JMP4.5 (SAS Institute) [16] was used to perform most of the statistical analyses reported in this study.

## 2.4 Structure Descriptors

The 181 descriptors calculated in MOE include 2D and internal 3D descriptors.

2D descriptors only use atom and connection information for the calculation, and no 3D coordinates or individual conformations are needed. The 2D descriptors include physical properties such as atom counts and bond counts, mr, logP and vdw\_area etc.; subdivided surface areas that are based on an approximate accessible van der Waals surface area calculation for each atom, v<sub>i</sub>, along with some other atomic property, p<sub>i</sub> (the v<sub>i</sub> are calculated using a connection table approximation) [17]; Kier and Hall connectivity and Kappa Shape indices [18]; adjacency and distance matrix descriptors [19–21]; pharmacophore features (*e.g.* donor, acceptor, polar, positive, negative, hydrophobic.) descriptors; and two sets of partial charges; one set was calculated from Partial Equalization of Orbital Electronegativities method [22] and the other set was previously stored forcefield (MMFF94) partial charges.

Internal 3D descriptors use 3D coordinate information but independent on the rotations and translations of the conformation. The internal 3D descriptors include potential energy descriptors, surface area, volume and shape descriptors, and conformation–dependent charge descriptors. The potential energy descriptors such as value of the potential energy (E), electrostatic component of potential energy (E\_ele), and solvation energy (E\_sol) etc. were calculated using MMFF94 forcefield and corresponding default potential set up in MOE. Surface area, volume, and shape descriptors such as water accessible surface area (ASA); Van der Waals volume (vol) and Van der Waals surface area (VSA) etc. depend on the structure connection and conformation. Conformation–dependent charge descriptors such as water accessible surface area of all atoms with positive charges (ASA+), water accessible surface area of all atoms with negative charges (ASA-), and water accessible surface area of all hydrophobic atoms (ASA\_H) etc. depend upon partial charges and conformations. Water accessible surface area was calculated using a radius of 1.4 angstroms for the water molecule.

#### **2.5** Computational Methods

The QSAR modeling process consists of the following steps: structure optimization using MMFF94 force field; evaluation of chemical structure descriptors; descriptor pruning through QSAR–contingency, correlation analysis of descriptors, step–forward and step–backward selection of descriptors; structural diversity analysis of the dataset based on pruned descriptor set and assigned weight to molecules if necessary; multiple regression analysis between pIC<sub>50</sub> and selected descriptors; evaluation of the significance level of the model and each determined descriptor; validation and cross–validation (leave–one–out procedure) of the model; detection of outliers and modification of QSAR–model; interpretation of the model equation.

MOE detects outliers with Grubbs test. The first step is to quantify how far away the experimental  $pIC_{50}$  is from the model value, by calculating the ratio Z–SCORE, defined as the difference between the  $pIC_{50}$  and model value divided by the SD of the whole dataset. MOE provides Z–SCORE values for all molecules and considers molecules with a Z–SCORE of 2.5 or more to be possible outliers.

Grubbs and others have tabulated critical values for Z–SCORE which are tabulated below for p = 0.05/0.02 (two tails) [24]. The critical value increases with sample size. Thus instead of simply taking the MOE criteria of outlier detection, we consulted the Grubbs table of Z–SCORE for different sample sizes for detecting outliers, and considered the complex influence of the PCA method, take the values of p = 0.02 as criteria.

Model adequacy was measured as the square of correlation coefficient ( $r^2$ ), root mean square error (s), cross-validated  $r^2$  ( $r_{LOO}^2$ ) and cross-validated s ( $s_{LOO}$ ).

## **3 RESULTS AND DISCUSSION**

#### **3.1 QSAR Model for dataset1**

After structure optimization, 181 descriptors were selected and evaluated from MOE descriptor selection panel. After descriptor pruning procedures, two descriptors were selected to build the final QSAR model for the data set. ASA denotes the water accessible area calculated using a radius of 1.4 angstroms for the water molecule, while PEOE VSA-1 denotes the sum of van der Waals surface areas of the atoms whose PEOE partial charge is in the range of [-0.10, -0.05]. PEOE (Partial Equalization of Orbital Electronegativities) [22] method of calculating atomic partial charges is a method in which charge is transferred between bonded atoms until equilibrium. Diversity analysis based on the two descriptors showed that there was no need to assign weight to the molecules. The two-descriptor linear model is shown in Eq. (3):

$$pIC_{50} = 0.51396 + 0.00675 \times (ASA) + 0.01742 \times (PEOE\_VSA-1)$$
  

$$r^{2} = 0.86, \ s = 0.45, \ r_{LOO}^{2} = 0.82, \ s_{LOO} = 0.51, \ n = 28, \ F = 77.56, \ k = 2$$
(3)

ASA and PEOE VSA-1 are all positively correlated with pIC<sub>50</sub> values, thus increasing ASA and PEOE VSA-1 will lead to the improvement of pIC<sub>50</sub>. The parameter effect tests for the model show that ASA is the determined descriptor in the model (Table 2). The 3D-QSAR/CoMFA analysis carried out by Szabo et al. [13] indicates that van der Waals interactions are very important in GGPPSase inhibition. Our model revealed the importance of water accessible surface area, which is mainly responsible for the van der Waals interactions between BPs and GGPPSase enzyme. Though our model did not provide 3D information like the CoMFA model, it offers a much simple equation and fast method to gain insight into the GGPPSase inhibitor system.

Table 2. Effect tests of the descriptors for Eq. (3)									
Descriptor	Correlation to $\text{pIC}_{50}(r^2)$	Sum of Squares	F Ratio	Prob > F					
ASA	0.76	14.61	64.78	< 0.0001					
PEOE_VSA-1	0.50	4.18	18.53	0.0002					

The leave-one-out cross-validated predictive pIC<sub>50</sub> values (PRED) were listed in Table 1 of supplementary material and plotted in Figure 5.

To test the predictive ability of our model, we also removed three compounds from the training set and performed the whole QSAR procedure on the reduced training set; then using the resulting model to predict the activities of the three excluded compounds. This procedure was repeated three times using different test sets, and the predicted pIC<sub>50</sub> values are listed in bold in Table 1 of supplementary material along with individual training sets and all statistical data for QSAR equations.



Figure 5. Leave-one-out cross-validated prediction versus experimental pIC<sub>50</sub> values for dataset1.

The three compounds in each test set were chosen following the Ref. [13] in order to compare the model predictive ability with that of the CoMFA model performed by Szabo *et al.*. The graphical result of the total nine compounds test set is shown in Figure 6. The rms error in predicted  $pIC_{50}$  of the test set compounds is 0.44, the correlation coefficient between experimental and predicted values is  $r^2 = 0.80$ .

The QSAR equations for the three training sets with reduced size are as follows:

$$pIC_{50} = 0.5489 + 0.006720 \times (ASA) + 0.01747 \times (PEOE\_VSA-1)$$
  

$$r^{2} = 0.85, \quad s = 0.47, \quad r_{LOO}^{2} = 0.81, \quad s_{LOO} = 0.54, \quad n = 25, \quad F = 64.53, \quad k = 2$$
(4)

$$pIC_{50} = 0.8793 + 0.005812 \times (ASA) + 0.02083 \times (PEOE\_VSA-1)$$

$$^{2} = 0.88 \times 0.40 \times 0^{2} = 0.85 \times 0.25 \times 0.25$$

$$P = 0.003, \quad S = 0.476, \quad P_{LOO} = 0.033, \quad n = 233, \quad P = 0.0033, \quad n = 23$$
  
 $pIC_{50} = 0.4741 + 0.006827 \times (ASA) + 0.01726 \times (PEOE VSA - 1)$ 

$$r^{2} = 0.85, \quad s = 0.46, \quad r_{LOO}^{2} = 0.81, \quad s_{LOO} = 0.54, \quad n = 25, \quad F = 64.81, \quad k = 2$$
 (6)



Figure 6. Predicted pIC<sub>50</sub> values versus experimental pIC<sub>50</sub> values for 9 GGPPSase inhibitors test set.

The comparison of Eq. (3) and the CoMFA model (Eq. (1)) reported by Szabo *et al.* [13] of the dataset1 is listed in Table 3. The rms error value between predicted and experimental values of the

test set is 0.39 for the CoMFA model and 0.44 for Eq. (3). Then, to compare the predictive ability of the two models, we can calculate the  $F_{9,9} = 1.27$  from the rms error values and look up the  $F_{0.05; 9,9} = 3.18$  from the F distribution. The result of the F test tells us the predictive ability of the two models has no significant difference at  $\alpha = 0.05$ .

Table 3. Statistical comparison of model (3) from the current study and model (1) reported by Szabo *et al.* [13]

Model	$r^2$	$r_{LOO}^2$	n <sup>a</sup>	$k^{b}$	F	rms error	Test $r^2$
Model (1)	0.938	0.90	28	3	86.8	0.39	0.88
Model (3)	0.86	0.82	28	2	77.56	0.44	0.80

<sup>*a*</sup> number of observations. <sup>*b*</sup> number of descriptors for certain model.

#### 3.2 QSAR Model for Dataset2

Firstly, the QSAR–contingency, correlation analysis, step–forward and step–backward selection procedures recommended 11 descriptors for the model of dataset2. Some of the descriptors such as a\_nH, apol, and KierFlex are correlated with (coefficient > 0.8) and irreplaceable by each other in the model. So many descriptors make the model complicated and difficult to interpret. And a model of 11 descriptors for a 28–observation dataset is sure over–fitting. In order to obtain a more robust and concise model, we performed principal components analysis (PCA) to reduce the dimensions of the descriptor subset, but failed.

We tried to select another subset among 181 descriptors. The element of the subset was measured mainly by its contribution to  $r^2$ . Finally we obtained a 32-descriptor subset, which keeps most interpretive information for pIC<sub>50</sub> and have the fewest number of descriptors at the same time. The statistical parameters of the model based on the 32 descriptors are:  $r^2 = 1.00$ , s = 0.00. The names of the 32 descriptors are listed in Table 5 of supplementary materials.



Figure 7. Leave-one-out cross-validated prediction versus experimental pIC<sub>50</sub> values for dataset2.

Then we transformed the 32 descriptors into a set of uncorrelated and normalized variables using PCA. To capture 100% of the variance in the previous 32–descriptor subset, 26 principal components (PCs) are needed. The accumulative percentage of variance explained by the first five PCs is 81.38%; with the 1<sup>st</sup> PC explaining 34.87\%, the 2<sup>nd</sup> 16.23\%, 3<sup>rd</sup> 13.22\%, 4<sup>th</sup> 6.01\%, and 5<sup>th</sup> 5.05\%.

After stepwise selection, four PCs (PC<sub>2</sub>, PC<sub>3</sub>, PC<sub>4</sub>, PC<sub>5</sub>) were determined to best describe the tendency of  $pIC_{50}$ . We obtained the following linear model:

$$pIC_{50} = 4.9108 + 0.2218 \times PC_2 - 0.4067 \times PC_3 - 0.5010 \times PC_4 + 0.1535 \times PC_5$$
  

$$r^2 = 0.85, \quad r = 0.30, \quad r_{LOO}^2 = 0.79, \quad s_{LOO}^2 = 0.35, \quad n = 28, \quad F = 32.03, \quad k = 4$$
(7)

The leave–one–out cross–validated predictive  $pIC_{50}$  values were listed in Table 2 of the supplementary materials and plotted in Figure 7.

We then carried out the leave-three-out procedure just as we did on model (3) to test whether the PCA model have predictive value. The selection of test compounds followed Martin *et al.* [5] on the CoMFA model (Eq. (2)). The results for three training-test sets of calculations are given in Table 2 of supplementary materials. The graphical representation of the results is shown in Figure 8. The rms error for the test set compounds was 0.66, and the correlation coefficient between experimental and predicted pIC<sub>50</sub> values was  $r^2 = 0.7$  (Test  $r^2$ ). The results indicate that Eq. (7) predicts the test set quite well and is not over fitting for the training set.



Figure 8. Predicted pIC<sub>50</sub> values versus experimental pIC<sub>50</sub> values for 9–compound test set of dataset2.

The comparison of PCA model (Eq. (7)) and the CoMFA model (Eq. (2)) [5] of the dataset2 is listed in Table 4. The rms error value of the test set (Test RMSE) is 0.32 for CoMFA (Eq. (2)) and 0.66 for Model (7). Then,  $F_{9,9} = 4.25$  is larger than the boundary value  $F_{0.05; 9,9}=3.18$ . It seems that model (7) is inferior to CoMFA model in predictive ability. However, compared to  $r^2$  and  $r^2_{LOO}$ , Test  $r^2$  value for the CoMFA model seems artificially high. General trend should be Test  $r^2 < r^2_{LOO} < r^2$  according to statistical principle. This may be resulted from chance correlation of the test

compounds to the CoMFA model (Eq. (2)). Therefore we cannot claim that predictive ability of the two models has significant difference at  $\alpha = 0.05$ .

Table 4. Statistical comparison of Eq. (7) and CoMFA model (Eq. (2))							
Model	$r^2$	$r_{LOO}^2$	п	k	F	rms error	Test $r^2$
Model (2)	0.87	0.79	26	4	34.80	0.32	0.87
Model (7)	0.85	0.79	28	4	32.03	0.66	0.70

**Note:** Dataset2 includes pamidronate (compound **2**) and T.B.2–13 (compound **118**) from Ref. [4], which did not include in CoMFA dataset [5].

#### 3.3 QSAR model of dataset3

The biological complexity of dataset3 is much greater than those of dataset1 and dataset2. The normal descriptor selection procedure suggested 16 descriptors for the dataset, and the statistical parameters of the model based on the 16 descriptors are:  $r^2 = 0.84$ , s = 0.40,  $r_{LOO}^2 = 0.50$ ,  $s_{LOO} = 0.85$ , n = 44, F = 8.36, k = 16. The names of the 16 descriptors are listed in Table 5 of supporting materials. Principal component analysis was carried out on the 16 descriptors. 16 PCs are required to capture the 100% variance in the previous descriptor subset. The accumulative percentage of variance explained by the first five PCs is 91.06%; with the 1<sup>st</sup> PC explaining 62.48%, the 2<sup>nd</sup> 10.29%, 3<sup>rd</sup> 8.27%, 4<sup>th</sup> 5.56%, and 5<sup>th</sup> 4.47%. After stepwise selection, six PCs (PC<sub>2</sub>, PC<sub>6</sub>, PC<sub>9</sub>, PC<sub>12</sub>, PC<sub>14</sub>, PC<sub>15</sub>) were selected to build the final model:

 $pED_{50} = 8.10 - 0.35 \times PC_2 - 0.25 \times PC_6 + 0.23 \times PC_9 - 0.65 \times PC_{12} + 0.22 \times PC_{14} - 0.26 \times PC_{15}$   $r^2 = 0.80, \quad s = 0.44, \quad r_{LOO}^2 = 0.72, \quad s_{LOO}^2 = 0.53, \quad n = 44, \quad F = 24.83, \quad k = 6$ (8)



Figure 9. Leave-one-out cross-validated prediction versus experimental *pED*<sub>50</sub> values for dataset3.

The percentage of variance explained by the 6 descriptors was listed in Table 5 respectively along with the result of parameter effect test of Eq. (8). The most correlative PC of Eq. (8) is  $PC_{12}$ 

(F = 80.98), which only explains 0.12% variance of original descriptor subset. The PCA procedure succeeded in extracting useful information and getting rid of noisy information from original dataset.

	Table 5. Effect tests of the descriptors for Eq. (8)					
Source	Correlation to pED <sub>50</sub> ( $r^2$ )	Sum of Squares	F Ratio	Prob > F	Percentage of variance (%)	
$PC_{12}$	0.44	18.76	80.98	<.0001	0.12	
$PC_2$	0.13	5.42	23.39	<.0001	10.29	
$PC_{15}$	0.07	3.09	13.33	0.0008	0.01	
$PC_6$	0.07	2.81	12.13	0.0013	3.56	
PC <sub>9</sub>	0.05	2.33	10.07	0.0030	1.31	
$PC_{14}$	0.05	2.12	9.13	0.0045	0.02	

Table 5.	Effect	tests	of t	the	descrij	ptors	for	Eq.	(8)	

The leave-one-out cross-validated predictive pIC<sub>50</sub> values were listed in the Table 3 of supplementary materials and plotted in Figure 9.

To further investigate the predictive ability of this model, we removed four compounds from the training set randomly before recomputing the QSAR equation on the reduced dataset. The  $pED_{50}$ values of the removed compounds were predicted using the QSAR model derived from the reduced training set. The procedure was repeated four times and the predicted 16 pED<sub>50</sub> values are given in the Table 3 of supplementary materials in bold and plotted in Figure 10. The graphical representation of the results is shown in Figure 10. The rms error between predicted pED<sub>50</sub> and the experimental pED<sub>50</sub> of the test set compounds was 0.34, and the correlation coefficient between experimental and predicted values is  $r^2 = 0.91$ . The quite good predictive result indicates that the PCA model (8) is robust and not seriously over fitting for the training set. Of course, general trend should be Test  $r^2 < r_{LOO}^2 < r^2$ ; the particularly high Test  $r^2$  should be attributed to chance correlation.



Figure 10. Predicted pED<sub>50</sub> values versus experimental pED<sub>50</sub> values for 16-compound test set of dataset3

#### 3.4 QSAR for dataset4

15 descriptors were selected through normal descriptor selection procedure. The statistical parameters of the model based on the 15 descriptors are:  $r^2 = 0.86$ , s = 0.39,  $r_{LOO}^2 = 0.68$ ,

 $s_{100} = 0.60$ , n = 42, F = 10.54, k = 15, The names of the 15 descriptors are listed in Table 5 of supplementary materials. Principal component analysis was carried out on the 15 descriptors. 15 PCs are required to capture the 100% variance of the previous descriptor subset. The accumulative percentage of variance explained by the first five PCs is 83.58%; with the 1<sup>st</sup> PC explaining 39.92%, the 2<sup>nd</sup> 18.85%, 3<sup>rd</sup> 9.86%, 4<sup>th</sup> 8.00%, and 5<sup>th</sup> 6.95%. After stepwise selection, seven PCs (PC<sub>3</sub>, PC<sub>7</sub>, PC<sub>9</sub>, PC<sub>10</sub>, PC<sub>12</sub>, PC<sub>14</sub>, PC<sub>15</sub>) were selected to build the final model:

$$pED_{50} = 7.69 + 0.53 \times PC_3 + 0.15 \times PC_7 + 0.17 \times PC_9 - 0.35 \times PC_{12} - 0.24 \times PC_{14} + 0.24 \times PC_{10} + 0.56 \times PC_{15}$$

$$r^2 = 0.80, \quad s = 0.46, \quad r_{LOO}^2 = 0.71, \quad s_{LOO} = 0.57, \quad n = 42, \quad F = 19.99.$$
(9)

The percentage of variance explained by the seven descriptors was listed in Table 6 respectively, along with the result of parameter effect test of Eq. (9). The most interpretive PC of the Eq. (9) is  $PC_{15}$  (F = 49.94), which only explains 0.01% variance of original descriptor subset. The PCA procedure also succeeded in extracting useful information and getting rid of noisy information from original dataset.

	I able	e 6. Effect Tests of th	ne descripto	rs for Eq. (9)	
Source	Correlation to pED <sub>50</sub>	Sum of Squares	F Ratio	Prob > F	Percentage of variance (%)
$PC_{15}$	0.29	13.14	49.94	<.0001	0.01
$PC_3$	0.25	11.59	44.08	<.0001	9.86
$PC_{12}$	0.11	5.21	19.82	<.0001	0.47
$PC_{14}$	0.05	2.35	8.95	0.0051	0.04
$PC_{10}$	0.05	2.32	8.84	0.0054	0.92
$PC_9$	0.03	1.22	4.66	0.0381	1.68
PC <sub>7</sub>	0.02	0.95	3.63	0.0654	4.61

The leave-one-out cross-validated predictive pED<sub>50</sub> values were listed in the Table 4 of supplementary materials and plotted in Figure 11.



Figure 11. Leave-one-out cross-validated prediction versus experimental pED<sub>50</sub> values for dataset4.



Figure 12. Predicted pED<sub>50</sub> values versus experimental pED<sub>50</sub> values for 16–compounds test set of dataset4.

A QSAR model with seven descriptive variables for a dataset of 42 compounds may have a tendency of over-fitting. The leave-four-out procedure was carried out to test the predictive ability and robustness of the model. The predicted pED<sub>50</sub> values for the 16 test compounds are listed in bold in Table 4 of supplementary materials and plotted in Figure 12. The rms error between predicted pED<sub>50</sub> and the experimental pED<sub>50</sub> of the test set compounds was 0.65, and the correlation coefficient between experimental and predicted values is  $r^2 = 0.71$ .

#### **4 CONCLUSIONS**

We have collected over 118 bisphosphonates with different bioactivities from various literature sources and performed QSAR studies on datasets according different bioactivities. For the GGPPSase inhibitor dataset (dataset1), we built a simple and explicit QSAR model based on the enzymatic activity of 28 compounds. This model (Eq. (3),  $r^2 = 0.86$ ,  $r_{LOO}^2 = 0.82$ , s = 0.45,  $s_{LOO} = 0.51, F = 77.56$ ) has comparable predictive ability with that of the CoMFA model (Eq. (1)) reported by Szabo et al. [13] for the same dataset. The QSAR of Dataset2 of 28 compounds with bioactivities against the growth of T. Brucei rhodesiense was studied using principal component analysis followed by stepwise variable selection. The PCA model (Eq. (7),  $r^2 = 0.85$ ,  $r^2_{LOO} = 0.79$ , s = 0.30,  $s_{LOO} = 0.35$ , F = 32.03) based on the dataset also has nearly equal predictive ability with that of the CoMFA model (Eq. (2)) built by Martin et al. [5] We divided the 86 bisphosphonates reported by Novartis with *in vivo* activity data in TPTX rats into two sub datasets according their structural features and rough speculations of their mode of action. A six PCs model (Eq. (8),  $r^{2} = 0.80, r_{LOO}^{2} = 0.72, s = 0.44, s_{LOO} = 0.53, F = 24.83$ ) elucidated the dataset of 44 compounds in which containing aliphatic linked nitrogen atoms. The other dataset includes 42 BPs containing a heterocyclic moiety with at least one nitrogen atom. Its PCA model (Eq. (9),  $r^2 = 0.80$ ,  $r_{LOO}^2 = 0.71$ , s = 0.46,  $s_{LOO} = 0.57$ , F = 19.99) consists of seven PCs. A leave-four-out test procedure shows that though the QSAR models based on in vivo bone resorption pED<sub>50</sub> values cannot provide explicit indications for drug design, their predictive ability for related compounds is quite good.

#### **Supplementary Material**

Table S1. Experimental  $IC_{50}$ ,  $pIC_{50}$  and predicted  $pIC_{50}$  values for GGPPSase inhibitors (dataset1) and statistical parameters for QSAR models

	Cmpd	Experimental	activity	QSAR 1	model pred	licted pIC <sub>5</sub>	0
Serial	Cmpd code	IC <sub>50</sub> (µM)	pIC <sub>50</sub>	Training set	3 c	compd test	set
111	Homorisedronate	410	3.39	3.67	3.80	3.76	3.77
2	Pamidronate	180	3.74	2.91	2.93	2.94	2.89
3	Alendronate	440	3.36	3.36	3.26	3.22	3.23
10	Ibandronate	83	4.08	4.83	4.72	4.70	4.70
112	Risedronate	350	3.46	3.46	3.63	3.61	3.60
108	NE97220	220	3.66	3.74	3.80	3.76	3.77
109	N-(2-(4-picolyl))AMDP	260	3.59	3.40	3.56	3.48	3.53
101	T.B. 006	2.2	5.66	5.50	5.41	5.44	5.38
100	T.B. 007	0.92	6.04	6.20	6.24	6.30	6.21
8	T.B. 009	330	3.48	3.16	3.10	3.09	3.07
99	T.B. 010	1.4	5.85	6.02	6.00	6.05	5.97
107	T.B. 012	220	3.66	3.33	3.26	3.22	3.22
110	T.B. 013	550	3.26	3.41	3.68	3.59	3.65
98	T.B. 014	0.72	6.14	5.62	5.55	5.61	5.52
105	T.B. 016	220	3.66	3.94	4.51	4.66	4.47
5	Neridronate	690	3.16	3.90	3.79	3.73	3.76
106	Т.В. 020	180	3.74	3.61	4.01	4.02	3.98
95	Т.В. 023	53	4.28	4.25	4.14	4.19	4.10
93	T.B. 024	620	3.21	3.58	3.41	3.46	3.37
94	T.B. 025	200	3.70	3.90	3.78	3.83	3.75
87	3–azaGGPP	0.14	6.85	6.57	6.69	6.44	6.69
91	3–azahomoFPP	0.31	6.51	5.93	5.91	5.68	5.90
90	3-azahomoGGPP	0.37	6.43	6.92	6.96	6.67	6.97
88	3–azaFPP	0.74	6.13	5.60	5.59	5.40	5.57
97	GGPP017	4.3	5.37	4.94	4.84	4.90	4.81
96	GGPP018	11	4.96	4.61	4.50	4.56	4.47
102	GGPP031	19	4.72	4.79	4.74	4.76	4.71
89	3–azaGPP	240	3.61	4.52	4.44	4.33	4.42
$r^2$				0.86	0.85	0.88	0.85
S				0.45	0.47	0.40	0.46
$r_{LOO}^2$				0.82	0.81	0.85	0.81
S <sub>LOO</sub>				0.51	0.54	0.46	0.54
F				77.56	64.53	83.83	64.81
k				2	2	2	2
п				28	25	25	25

	Cmpd	Experimental	activity	QSAR m	nodel prec	licted pIC	50
Serial	Cmpd code	IC <sub>50</sub> (μM)	pIC <sub>50</sub>	Training set	3 c	ompd test	set
111	Homorisedronate	1.7	5.77	5.89	5.89	5.93	5.89
2	Pamidronate	177	3.75	4.22	4.24	4.23	4.21
10	Ibandronate	0.96	6.01	5.88	5.87	5.86	5.96
112	Risedronate	8.6	5.06	5.52	5.51	5.57	5.51
108	NE97220	0.70	6.15	5.87	5.85	5.88	5.96
109	N-(2-(4-picolyl))AMDP	0.61	6.21	5.83	5.81	5.83	5.94
113	NE58018	0.22	6.66	6.52	6.51	6.53	6.59
114	N-(2-(5-chloro)-pyridyl)AMDP	53.30	4.27	5.16	5.12	5.16	5.27
101	T.B. 006	1.70	5.77	5.20	5.22	5.23	5.18
100	T.B. 007	2.0	5.70	5.75	5.73	5.78	5.75
7	Olpadronate	5.4	5.27	5.47	5.50	5.45	5.53
8	T.B. 009	7.8	5.11	4.89	4.92	4.91	4.90
99	T.B. 010	8.0	5.10	5.16	5.15	5.20	5.13
107	T.B. 012	8.6	5.07	4.64	4.62	4.67	4.66
110	T.B. 013	19.8	4.70	5.00	4.99	5.03	5.02
98	T.B. 014	20.5	4.69	4.74	4.69	4.76	4.76
115	T.B.015	20.9	4.68	4.59	4.62	4.64	4.50
105	T.B. 016	21.3	4.67	4.77	4.73	4.82	4.75
5	Neridronate	31.7	4.50	4.68	4.71	4.71	4.66
116	T.B. 018	34.4	4.46	4.50	4.53	4.49	4.54
117	T.B. 019	39.5	4.40	4.38	4.34	4.39	4.47
106	T.B. 020	40.0	4.39	4.39	4.37	4.49	4.31
103	T.B. 021	50.6	4.30	3.98	3.99	4.00	3.93
95	T.B. 023	62.4	4.20	4.02	3.97	4.02	4.07
93	T.B. 024	92.0	4.04	3.87	3.83	3.86	3.92
95	T.B. 025	99.8	4.00	3.98	3.95	3.99	3.99
104	T.B. 026	102.0	3.99	4.19	4.19	4.21	4.20
118	T.B. 2–13	27.9	4.55	4.39	4.35	4.41	4.53
$r^2$				0.85	0.83	0.85	0.91
S				0.30	0.30	0.29	0.24
$r_{LOO}^2$				0.79	0.76	0.79	0.88
$S_{LOO}$				0.35	0.37	0.34	0.28
F				32.03	24.88	27.43	50.13
k				4	4	4	4
п				28	25	25	25

**Table S2.** Experimental IC50, pIC50 and predicted pIC50 values for bisphosphonates against T. Brucei Trypomastigotes (dataset2) and statistical parameters for QSAR models

models	0 1	<b>D</b> 1 (1	,	6		1. 1 1	10	<u> </u>
~	Cmpd	Experimental a	activity	(	SAR mode	predicted	pIC <sub>50</sub>	
Serial	Cmpd code	ED <sub>50</sub> (µg /kg)	$pED_{50}$	Training set		4 compd	test set	
2	Pamidronate	61	7.21	7.48	7.50	7.50	7.45	7.45
3	Alendronate	8	8.10	8.04	8.05	8.05	8.08	8.10
10	Ibandronate	1.1	8.96	8.10	8.16	8.25	8.24	8.30
7	Olpadronate	12	7.92	8.10	8.03	8.12	8.05	8.08
8	T.B. 009	3.4	8.47	8.25	8.23	8.26	8.27	8.30
5	Neridronate	60	7.22	7.65	7.69	7.60	7.61	7.61
1	Novartis 1a	150	6.82	7.31	7.20	7.29	7.25	7.25
4	Novartis 1d	20	7.70	7.86	7.92	7.84	7.87	7.88
6	Novartis 1g	15	7.82	8.03	7.95	8.04	8.04	8.07
8	T.B. 009	3	8.52	8.61	8.49	8.60	8.59	8.66
9	Novartis 1i	3	8.52	8.17	8.09	8.23	8.23	8.29
11	Novartis 11	100	7	6 80	6 90	6.87	6.87	6.84
13	Novartis 10	18	7 74	8 57	8 30	8 36	8 29	8 31
14	Novartis 1n	65	7 1 9	7.08	7 21	7 11	7.08	7.06
20	Novarris 4a	300	6.52	6.58	6.63	6.62	6.62	6 59
30	Novartis 4h	14	8.85	9.47	9.33	9.37	9.32	9.39
31	Novartis Ac	20	7 70	8 3 3	8 20	8 30	9.52 8.27	8 31
22	Novartis 4d	20	0	8.55	8.29	8.30 9.97	8.27	8.51
32	Novartis 4a	1	7 87	0.07	0.05	7.80	7.06	0.09
33	Novartis 46	15	7.82	7.87 9.72	7.87 8.67	7.03 9.72	7.00 9.67	7.07 9.72
25	Novartis 4g	1.5	0.02	8.73 8.00	8.07	0.75	0.07 8.02	0.72 8.00
35	Novartis 4g	0.7	9.13	0.99	8.98 8.46	0.90	0.95	0.99
30	Novartis 41	1	9	8.45	8.40	8.30	8.02 9.70	8.08
3/	Novartis 4j	0.4	9.40	8.74	8.0/	8.81	8.79	8.8/
38	Novartis 4k	20	/./0	/.09	/.10	1.24	/.1/	/.18
39	Novartis 41	1500	5.82	6.26	6.09	6.25	6.21	6.18
40	Novartis 5a	1.5	8.82	9.08	9.05	9.04	8.98	9.04
41	Novartis 5b	1.7	8.77	9.45	9.38	9.35	9.36	9.43
42	Novartis 5c	1.2	8.92	8.01	8.22	8.17	8.14	8.17
43	Novartis 5d	0.5	9.30	8.45	8.56	8.50	8.52	8.55
44	Novartis5e	1.7	8.77	8.01	8.10	8.07	8.04	8.05
46	Novartis 5g	1.3	8.89	8.34	8.42	8.35	8.36	8.38
45	Novartis 5f	0.6	9.22	9.72	9.52	9.63	9.61	9.70
47	Novartis 5h	1.2	8.92	8.55	8.56	8.63	8.64	8.67
48	Novartis 5i	20	7.70	7.75	7.84	7.77	7.76	7.76
49	Novartis 5j	10	8	7.78	7.80	7.82	7.83	7.86
50	Novartis 5k	500	6.30	7.03	6.93	6.94	6.98	6.97
51	Novartis 51	4	8.40	7.57	7.67	7.64	7.64	7.64
52	Novartis 5m	7500	5.12	5.15	5.17	5.28	5.21	5.14
53	Novartis 5n	100	7	7.71	7.66	7.72	7.70	7.71
54	Novartis 5p	0.7	9.15	8.97	9.05	8.95	8.98	9.01
55	Novartis 5q	7	8.15	8.91	8.72	8.78	8.82	8.87
56	Novartis 5r	0.33	9.48	8.82	8.85	8.90	8.84	8.89
57	Novartis 5s	7.8	8.11	8.95	8.76	8.65	8.66	8.66
72	Novartis 7e	7	8.15	9.13	8.61	8.59	8.60	8.66
$r^2$				0.80	0.75	0.77	0.79	0.79
S				0.44	0.44	0.46	0.45	0.44
$r_{LOO}^2$				0.72	0.65	0.67	0.70	0.68
100				0.53	0.55	0.55	0.54	0.55
<b>у</b> <sub>LOO</sub>				24.83	16.61	18.42	21.24	20.47
k				6	6	6	6	6
n				44	40	40	40	40

Table S3. Experimental  $ED_{50}$ ,  $pED_{50}$  and predicted  $pED_{50}$  values for dataset3 and statistical parameters for QSAR models

	mnd	Experimental	otivity	09	AR model	predicted	nIC.	
Serial num	Cmnd code		nED	Qo Training set	AK HIOUEI	4 comp	test set	
65	Zaladranata	$ED_{50}(\mu g/\kappa g)$	10.15	0.50	0.52	0.55	0.75	0.66
05	Novertia 2a	0.07	10.15	9.30	9.53 7.92	9.33	9.73	9.00
15	Novartis 2a	25	0 7.60	1.19 7.63	7.64 7.66	7.00 7.65	7.00 7.68	7.05
10	Novartis 20	25	7.00	6.65	6.73	6.63	6.66	6.63
17	Novartis 2d	230	7.15	0.05	7 35	7 30	7.36	7.28
10	Novartis 2e	35	8.46	8 49	8 46	8.48	8.55	8 50
20	Novartis 26	5.5	8 25	7 50	7 58	7.62	7.64	0.50 7 47
20	Novartis 2g	11	7.96	8 64	8 56	8.54	8.68	8.58
21	Novartis 2h	100	7.90	7.02	7.05	6.98	7.09	7 09
23	Novartis 2i	30	7 52	7.13	7.00	7.18	7.05	7 38
23	Novartis 2k	25	7.60	8 24	8 14	8 16	8.25	8.09
25	Novartis 2m	400	6 40	7 13	6.82	6.84	6 67	7.31
26	Novartis 3a	50	7 30	7 70	7.59	7.66	7 70	7 41
27	Novartis 3b	250	6 60	5 94	6.11	6.12	6.12	5.92
28	Novartis 3c	2500	5.60	5.52	5.62	5.67	5.65	5.54
58	Novartis 6a	5	8.30	8.83	8.75	8.70	8.87	8.90
59	Novartis 6b	0.6	9.22	9.04	8.99	9.05	9.15	9.32
60	Novartis 6c	25	7.60	7.17	7.31	7.31	7.40	7.44
61	Novartis 6d	0.3	9.52	8.21	8.35	8.21	8.38	8.52
62	Novartis 6e	20	7.70	7.59	7.65	7.48	7.65	7.82
63	Novartis 6f	15	7.82	8.45	8.36	8.25	8.32	8.50
64	Novartis 6h	1.5	8.82	8.41	8.47	8.39	8.47	8.65
66	Novartis 6j	45	7.35	8.08	7.92	7.86	8.03	7.77
67	Novartis 6k	3	8.52	8.64	8.52	8.63	8.70	8.79
68	Novartis 61	1.5	8.82	7.92	8.02	7.98	8.03	8.17
69	Novartis 6n	600	6.22	6.60	6.60	6.56	6.59	6.52
70	Novartis 7c	800	6.10	6.65	6.57	6.53	6.60	6.42
71	Novartis 7d	40	7.40	6.91	7.04	7.01	7.11	7.17
73	Novartis 8a	5	8.30	7.85	7.83	7.91	7.97	7.87
74	Novartis8b	100	7	7.24	7.20	7.24	7.23	7.21
75	Novartis 8c	1.5	8.82	7.98	8.20	8.07	8.18	8.54
76	Novartis 8d	1.5	8.82	8.69	8.66	8.65	8.73	8.90
77	Novartis 8e	2	8.70	8.29	8.31	8.23	8.32	8.56
78	Novartis 8f	0.9	9.05	8.91	8.95	8.80	8.94	9.07
79	Novartis 8g	200	6.70	6.63	6.64	6.66	6.65	6.75
80	Novartis 8h	2.7	8.57	10.22	9.66	9.53	9.74	10.32
81	Novartis 8j	500	6.30	7.21	7.15	7.13	7.15	7.12
82	Novartis 8k	5	8.30	7.83	7.88	8.00	7.94	8.06
83	Novartis 81	75	7.12	7.25	7.21	7.19	7.19	7.20
84	Novartis 8m	200	6.70	7.40	7.24	7.27	7.24	7.34
85	Novartis 9a	200	6.70	6.62	6.69	6.60	6.71	6.41
86	Novartis 9b	700	6.15	6.38	6.34	6.23	6.30	6.29
$r^2$				0.80	0.76	0.81	0.80	0.83
S				0.46	0.46	0.44	0.46	0.45
$r_{LOO}^2$				0.71	0.65	0.69	0.69	0.75
S <sub>LOO</sub>				0.55	0.60	0.56	0.58	0.51
F				19.99	13.84	18.47	17.44	20.59
k				7	7	7	7	7
п				42	38	38	38	38

Table S4. Experimental  $ED_{50}$ ,  $pED_{50}$  and predicted  $pED_{50}$  values for dataset4 and statistical parameters for QSAR models

Datasets	Original descriptors
Dataset2	a_nH, zagreb, PEOE_VSA+0, PEOE_VSA+1, PEOE_VSA+2, PEOE_VSA-1, Q_VSA_POS,
	Q_VSA_HYD, Q_VSA_PPOS, E_sol, E_stb, E_strain, E_tor, E_vdw, KierFlex, apol, vsa_don,
	vsa_other, SlogP_VSA1, SlogP_VSA4, SlogP_VSA7, SlogP_VSA8, SlogP_VSA9, SMR_VSA1,
	SMR_VSA2, SMR_VSA3, SMR_VSA4, SMR_VSA5, SMR_VSA6, SMR_VSA7, vol, VSA
Dataset3	VSA, DASA, vol, SlogP_VSA8, E_tor, E_ang, Q_VSA_POS, Zagreb, ASA+, SMR_VSA6,
	Q_VSA_PNEG, Q_VSA_HYD, PEOE_VSA_HYD, PEOE_VSA-1, weinerPath
Dataset4	weinerPol, PEOE_VSA+3, E_ang, SlogP_VSA5, DASA, DCASA, E_vdw, apol, SlogP_VSA6,
	SMR VSA2, ASA H. PEOE VSA+4, PEOE VSA–3, O VSA HYD, SMR VSA5

Table S5. Original descriptors adopted for PCA procedure in each dataset

#### Appendix 1

Denotations of original descriptors adopted for PCA procedure in each dataset

I. Physical Properties that can be calculated from the connection table (with no dependence on conformation) of a molecule:

apol Sum of the atomic polarizabilities (including implicit hydrogens) with polarizabilities taken from ref [2]	Code	Description	
aper Sum et me wenne permileennee (merwang mpreterity wegens) with permileennees when nom ten [	apol	Sum of the atomic polarizabilities (including implicit hydrogens) with polarizabilities taken from ref. [24	4].

II.Subdivided Surface Areas

The Subdivided Surface Areas are descriptors based on an approximate accessible van der Waals surface area calculation for each atom,  $v_i$  along with some other atomic property,  $p_i$ . The  $v_i$  are calculated using a connection table approximation. Each descriptor in a series is defined to be the sum of the  $v_i$  over all atoms *i* such that  $p_i$  is in a specified range (a,b).

In the descriptions to follow,  $L_i$  denotes the contribution to logP(o/w) for atom *i* as calculated in the SlogP descriptor [17].  $R_i$  denotes the contribution to Molar Refractivity for atom *i* as calculated in the SMR descriptor [17]. The ranges were determined by percentile subdivision over a large collection of compounds.

Code	Description	
SlogP_VSA0	Sum of $v_i$ such that $L_i \leq -0.4$ .	
SlogP_VSA1	Sum of $v_i$ such that $L_i$ is in (-0.4,-0.2].	
SlogP_VSA2	Sum of $v_i$ such that $L_i$ is in (-0.2,0].	
SlogP_VSA3	Sum of $v_i$ such that $L_i$ is in (0,0.1].	
SlogP_VSA4	Sum of $v_i$ such that $L_i$ is in $(0.1, 0.15]$ .	
SlogP_VSA5	Sum of $v_i$ such that $L_i$ is in (0.15,0.20].	
SlogP_VSA6	Sum of $v_i$ such that $L_i$ is in (0.20,0.25].	
SlogP_VSA7	Sum of $v_i$ such that $L_i$ is in (0.25,0.30].	
SlogP_VSA8	Sum of $v_i$ such that $L_i$ is in (0.30,0.40].	
SlogP_VSA9	Sum of $v_i$ such that $L_i > 0.40$ .	
SMR_VSA0	Sum of $v_i$ such that $R_i$ is in [0,0.11].	
SMR_VSA1	Sum of $v_i$ such that $R_i$ is in (0.11,0.26].	
SMR_VSA2	Sum of $v_i$ such that $R_i$ is in (0.26,0.35].	
SMR_VSA3	Sum of $v_i$ such that $R_i$ is in (0.35,0.39].	
SMR_VSA4	Sum of $v_i$ such that $R_i$ is in (0.39,0.44].	
SMR_VSA5	Sum of $v_i$ such that $R_i$ is in (0.44,0.485].	
SMR_VSA6	Sum of $v_i$ such that $R_i$ is in (0.485,0.56].	
SMR_VSA7	Sum of $v_i$ such that $R_i > 0.56$ .	

II. Atom Counts and Bond Counts and Kier&Hall Connectivity and Kappa Shape Indices

Code De	scription
a_nH Nu	mber of hydrogen atoms (including implicit hydrogens). This is calculated as the sum of h <sub>i</sub> over
all nor	-trivial atoms i plus the number of non-trivial hydrogen atoms.
zagreb Zag	greb index: the sum of d <sub>i</sub> <sup>2</sup> over all heavy atoms i.
KierFlex Kie	er molecular flexibility index: (KierA1) (KierA2) /n [18].

III. Adjacency	and Distance Matrix Descriptors
Code	Description
weinerPath	Wiener path number: half the sum of all the distance matrix entries as defined in ref. [19] and [25].
weinerPol	Wiener polarity number: half the sum of all the distance matrix entries with a value of 3 as defined
j	in ref. [19].
IV. Pharmacop	hore Feature Descriptors
0.1	

III. Adjacency and Distance Matrix Descriptors

 Code
 Description

 vsa\_don
 Approximation to the sum of VDW surface areas of pure hydrogen bond donors (not counting basic atoms and atoms that are both hydrogen bond donors and acceptors such as -OH).

 vsa\_other
 Approximation to the sum of VDW surface areas of atoms typed as "other".

V. Partial Charge Descriptors (Let  $q_i$  denote the partial charge of atom *i* as defined above. Let  $v_i$  be the van der Waals surface area of atom *i*.)

Code	Description
Q_PC+	Total positive partial charge: the sum of the positive q <sub>i</sub> . Q_PC+ is identical to PC+ which
PEOE_PC+	has been retained for compatibility.
Q_PC-	Total negative partial charge: the sum of the negative q <sub>i</sub> . Q_PC- is identical to PC- which
PEOE_PC-	has been retained for compatibility.
Q_RPC+	Relative positive partial charge: the largest positive q <sub>i</sub> divided by the sum of the positive q <sub>i</sub> .
PEOE_RPC+	Q_RPC+ is identical to RPC+ which has been retained for compatibility.
Q_PRC-	Relative negative partial charge: the smallest negative q <sub>i</sub> divided by the sum of the negative
PEOE_RPC-	q <sub>i</sub> . Q_RPC- is identical to RPC- which has been retained for compatibility.
Q_VSA_POS	Total positive van der Waals surface area. This is the sum of the vi such that qi is non-
PEOE_VSA_POS	negative. The v <sub>i</sub> are calculated using a connection table approximation.
Q_VSA_NEG	Total negative van der Waals surface area. This is the sum of the v <sub>i</sub> such that q <sub>i</sub> is negative.
PEOE_VSA_NEG	The v <sub>i</sub> are calculated using a connection table approximation.
Q_VSA_PPOS	Total positive polar van der Waals surface area. This is the sum of the v <sub>i</sub> such that q <sub>i</sub> is
PEOE_VSA_PPOS	greater than 0.2. The v <sub>i</sub> are calculated using a connection table approximation.
Q_VSA_PNEG	Total negative polar van der Waals surface area. This is the sum of the v <sub>i</sub> such that q <sub>i</sub> is less
PEOE_VSA_PNEG	than $-0.2$ . The v <sub>i</sub> are calculated using a connection table approximation.
Q_VSA_HYD	Total hydrophobic van der Waals surface area. This is the sum of the $v_i$ such that $ q_i $ is less
PEOE_VSA_HYD	than or equal to 0.2. The $v_i$ are calculated using a connection table approximation.
Q_VSA_POL	Total polar van der Waals surface area. This is the sum of the $v_i$ such that $ q_i $ is greater than
PEOE_VSA_POL	0.2. The $v_i$ are calculated using a connection table approximation.
Q_VSA_FPOS	Fractional positive van der Waals surface area. This is the sum of the v <sub>i</sub> such that q <sub>i</sub> is non-
PEOE_VSA_FPOS	negative divided by the total surface area. The $v_i$ are calculated using a connection table
	approximation.
Q_VSA_FNEG	Fractional negative van der Waals surface area. This is the sum of the $v_i$ such that $q_i$ is
PEOE_VSA_FNEG	negative divided by the total surface area. The $v_i$ are calculated using a connection table
	approximation.
Q_VSA_FPPOS	Fractional positive polar van der Waals surface area. This is the sum of the $v_i$ such that $q_i$ is
PEOE_VSA_FPPOS	greater than 0.2 divided by the total surface area. The $v_i$ are calculated using a connection table
	approximation.
Q_VSA_FPNEG	Fractional negative polar van der Waals surface area. This is the sum of the $v_i$ such that $q_i$ is
PEOE_VSA_FPNEG	less than $-0.2$ divided by the total surface area. The v <sub>i</sub> are calculated using a connection table
	approximation.
Q_VSA_FHYD	Fractional hydrophobic van der Waals surface area. This is the sum of the $v_i$ such that $ q_i $ is
PEOE_VSA_FHYD	less than or equal to 0.2 divided by the total surface area. The $v_i$ are calculated using a
	connection table approximation.
Q_VSA_FPUL	Fractional polar van der waals surface area. This is the sum of the $v_i$ such that $ q_i $ is greater
PEOE_VSA_FPOL	than 0.2 divided by the total surface area. The $v_i$ are calculated using a connection table
DECE VEALA	approximation. Sum of $x$ where $x$ is greater than 0.2
FEUE_VSATO DEOE_VSAT5	Sum of v where q is in the range $[0.25, 0.30)$
$\frac{1}{2} EOE_VSAT3$ $\frac{1}{2} DEOE_VSAT3$	Sum of v where q is in the range $[0.23, 0.30]$ .
FEUE_VSAT4 DEOE_VSA+2	Sum of v where q is in the range $[0.20, 0.25]$ .
I LUE_V SATS	Sum of $v_1$ where $q_i$ is in the range [0.13,0.20].

V. (Con	tinued)
Code	Description
PEOE_V	$\sqrt{SA+2}$ Sum of v <sub>i</sub> where q <sub>i</sub> is in the range [0.10,0.15).
PEOE_V	$\sqrt{SA+1}$ Sum of v <sub>i</sub> where q <sub>i</sub> is in the range [0.05,0.10).
PEOE_V	$\sqrt{SA+0}$ Sum of v <sub>i</sub> where q <sub>i</sub> is in the range [0.00,0.05).
PEOE_V	$\sqrt{SA-0}$ Sum of v <sub>i</sub> where q <sub>i</sub> is in the range [-0.05,0.00).
PEOE_V	$\sqrt{SA-1}$ Sum of v <sub>i</sub> where q <sub>i</sub> is in the range [-0.10,-0.05).
PEOE_V	$\sqrt{SA-2}$ Sum of v <sub>i</sub> where q <sub>i</sub> is in the range [-0.15,-0.10).
PEOE_V	$\sqrt{SA-3}$ Sum of v <sub>i</sub> where q <sub>i</sub> is in the range [-0.20,-0.15).
PEOE_V	$\sqrt{SA-4}$ Sum of v <sub>i</sub> where q <sub>i</sub> is in the range [-0.25,-0.20).
PEOE_V	$\sqrt{SA-5}$ Sum of v <sub>i</sub> where q <sub>i</sub> is in the range [-0.30,-0.25).
PEOE_V	$\sqrt{SA-6}$ Sum of v <sub>i</sub> where q <sub>i</sub> is less than $-0.30$ .
VI. Pote	ential Energy Descriptors
Code	Description
E_ang	Angle bend potential energy. In the Potential Setup panel, the term enable flag is ignored, but the term
<b>F</b> 1	weight is applied.
E_sol	Solvation energy. In the Potential Setup panel, the term enable flag is ignored, but the term weight is
E «4h	applied.
E_SID	Bond stretch-bend cross-term potential energy. In the Potential Setup panel, the term enable hag is
E strain	I coal strain energy: the current energy minus the value of the energy at a near local minimum. The
E_strain	current energy is calculated as for the E descriptor. The local minimum energy is the value of the E
	descriptor after first performing an energy minimization. Current chirality is preserved and charges are left
	undisturbed during minimization. The structure in the database is not modified (results of the minimization
	are discarded)
E tor	Torsion (proper and improper) potential energy. In the Potential Setup panel, the term enable flag is
	ignored but the term weight is applied
E vdw	van der Waals component of the potential energy. In the Potential Setup panel, the term enable flag is
	ignored, but the term weight is applied.
VII. Sur	tace Area, Volume and Shape Descriptors
Code	
ASA	Water accessible surface area calculated using a radius of 1.4 A for the water molecule. A polyhedral
1	representation is used for each atom in calculating the surface area.
VOI	van der waals volume calculated using a grid approximation (spacing 0.75 A).
V SA	van der waals surface area. A polynedral representation is used for each atom in calculating the surface
VII. Cor	nformation Dependent Charge Descriptors
Code	Description
ASA+	Water accessible surface area of all atoms with positive partial charge (strictly greater than 0).
ASA_H	Water accessible surface area of all hydrophobic ( $ q_i  < 0.2$ ) atoms.
DASĀ	Absolute value of the difference between ASA+ and ASA <sup>-</sup> .
DCASA	Absolute value of the difference between CASA $+$ and CASA $-$ [26].

#### **5 REFERENCES**

- [1] R. G. G. Russell, M. J. Rogers, Bisphosphonates: From the Laboratory to the Clinic and Back Again, *Bone* **1999**, 25, 97–106.
- [2] T. H. Cromartie, K. J. Fisher, and J. N. Grossman, The Discovery of a Novel Site of Action for Herbicidal Bisphosphonates, *Pesticide Biochemistry and Physiology* **1999**, *63*, 114–126.
- [3] S. Oura, T. Sakurai, G. Yoshimura, T. Tamaki, and T. Umemura, Study on the Safty of Rapid Infusion and the Efficacy of Incadronate Against Bone Metastase of breast Cancer, *Gan to Kagaku Ryoho* **1999**, *26*, 1623–1628.
- [4] M. B. Martin, J. S. Grimley, J. C. Lewis, H. T. Heath, and B. N. Bailey, Bisphosphonates Inhibit the Growth of Trypanosoma brucei, Trypanosoma Cruzi, Leishmania donovani, Toxoplasma gondii, and Plasmodium

falciparum: A Potential Route to Chemotherapy, J. Med. Chem. 2001, 44, 909-916.

- [5] M. B. Martin, J. M. Sanders, H. Kendrick, K. d. Luca–Fradley, J. C. Lewis, J. S. Grimley, E. M. Van Brussel, J. R. Olsen, G. A. Meints, A. Burzynska, P. Kafarski, S. L. Croft, and E. Oldfield, Activity of Bisphosphonates against Trypanosoma brucei rhodesiense, *J. Med. Chem.* 2002, 45, 2904–2914.
- [6] D. Amin, S. A. Cornell, S. K. Gustafson, S. J. Needle, J. W. Ullrich, G. E. Bilder, and M. H. Perrone, Bisphosphonates used for the Treatment of Bone Disorders Inhibit Squalene Synthase and Cholesterol Biosynthesis, *J. Lipid. Res.* **1992**, *33*, 1657–1663.
- [7] D. Amin, S. A. Cornell, M. H. Perrone, and G. E. Bilder, 1-hydroxy-3-(methylpentylamino)-propylidene-1,1-Bisphosphonic Acid as a Potent Inhibitor of Squalene Synthase, *Drug Res.* **1996**, *46*, 759-762.
- [8] T. H. Cromartie, and K. J. Fisher, Method of Controlling Plants by Inhibition of Farnesyl Pyrophosphate Synthase, Zeneca Limited, London, England **1998**, 5, 756, 423 (US Patent).
- [9] J. E. Grove, R. J. Brown, and D. J. Watts, The Intracellular Target for the Antiresorptive Aminobisphosphonate Drugs in *Dictyostelium discoideum* is the Enzyme Farnesyl Diphosphate Synthase, *J. Bone Miner. Res.* **2000**, *15*, 971–981.
- [10] M. Sato, W. Grasser, N. Endo, R. Akins, H. Simmons, D. D. Thompson, E. Golub, and G. A. Rodan, Bisphosphonate action: Alendronate Localization in Rat Bone and Effects on Osteoclast Ultrastructure, J. Clin. Invest. 1991, 88, 2095–2105.
- [11] J. D. Bergstrom, R. G. Bostedor, P. J. Masarachia, A. A. Reszka, and G. Rodan, Alendronate Is a Specific, Nanomolar Inhibitor of Farnesyl Diphosphate Synthase, *Biochemistry and Biophysics* 2000, 373, 231–241.
- [12] C. M. Szabo, M. B. Martin, and E. Oldfield, An Investigation of Bone Resorption and Dictyostelium discoideum Growth Inhibition by Bisphosphonate Drugs, J. Med. Chem. 2002, 45, 2894–2903.
- [13] C. M. Szabo, Y. Matsumura, S. Fukura, M. B. Martin, J. M. Sanders, S. Sengupta, J. A. Cieslak, T. C. Loftus, C. R. Lea, H. J. Lee, A. Koohang, R. M. Coates, H. Sagami, and E. Oldfield, Inhibition of Geranylgeranyl Diphosphate Synthase by Bisphosphonates and Diphosphates: A Potential Route to New Bone Antiresorption and Antiparasitic Agents, *J. Med. Chem.* 2002, 45, 2185–2196.
- [14] L. Widler, K. A. Jaeggi, M. Glatt, K. Muller, R. Bachmann, M. Bisping, A. R. Born, R. Cortesi, G. Guiglia, H. Jeker, R. Klein, U. Ramseier, J. Schmid, G. Schreiber, Y. Seltenmeyer, and J. R. Green, Highly Potent Geminal Bisphos phonates. From Pamidronate Disodium (Aredia) to Zoledronic Acid (Zometa), *J. Med. Chem.* 2002, 45, 3721–3738.
- [15] MOE 2001.01, Chemical Computing Group Inc., Montreal, Canada, E-mail info@chemcomp.com, www http://www.chemcomp.com.
- [16] JMP 4.5, SAS Institute Inc., Cary, NC, USA, www http://www.jmp.com.
- [17] S. A. Wildman, G. M. Crippen, Prediction of Physicochemical Parameters by Atomic Contributions, J. Chem. Inf. Comput. Sci. 1999, 39, 868–873.
- [18] L H. Hall and L. B. Kier, The Molecular Connectivity Chi Indices and Kappa Shape Indices in Structure–Property Modeling; in: *Reviews in Computational Chemistry*, Eds, K. B. Lipkowitz and D. B. Boyd, VCH Publishers, New York, **1991**, Vol. 2, pp. 367–422.
- [19] A. T. Balaban, Five New Topological Indices for the Branching of Tree–like Graphs, *Theor. Chim. Acta.* 1979, *53*, 355–375.
- [20] A. T. Balaban, Highly Discriminating Distance-based Topological Index, Chem. Phys. Lett. 1982, 89, 399-404.
- [21] M. Petitjean, Applications of the Radius–Diameter Diagram to the Classification of Topological and Geometrical Shapes of Chemical Compounds, J. Chem. Inf. Comput. Sci. 1992, 32, 331–337.
- [22] J. Gasteiger, M. Marsili, Iterative Partial Equalization of Obital Electronegativity A Rapid Access to Atomic Charges, *Terahedron* **1980**, *36*, 3219–3228.
- [23] F. E. Grubbs, Procedures for Detecting Outlying Observations in Samples, Technometrics 1969, 11, 1–21.
- [24] D. R. Lide, CRC Handbook of Chemistry and Physics, CRC Press, 2000 NW Corporate Blvd Boca Raton, FL 33431, USA, 1994
- [25] H. Wiener, Structural Determination of Paraffin Boiling Points, J. Am. Chem. Soc. 1947, 69, 17-20.
- [26] D. T. Stanton and P. C. Jurs, Development and Use of Charged Partial Surface Area Structural Descriptors in Computer Assissted Quantitative Structure Property Relationship Studies, *Anal. Chem.* 1990, 62, 2323–2329.